

# 1 Plasma-assisted CO<sub>2</sub> conversion in a gliding arc discharge: Improving 2 performance by optimizing the reactor design

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## 8 **Abstract:**

9 In this paper, a gliding arc discharge (GAD) is investigated for CO<sub>2</sub> decomposition,  
10 with particular efforts directed toward improving the performance by optimizing the  
11 reactor design. The effects of various parameters, e.g., gas flow rate, configuration of  
12 the injector nozzle, and structure of the quartz cover, on the CO<sub>2</sub> conversion and energy  
13 efficiency are investigated. The results indicate that the variation profiles of CO<sub>2</sub>  
14 conversion upon rising flow rate can be clearly be divided into two patterns: Pattern A  
15 with outlet gas temperature > 440 °C and Pattern B with outlet gas temperature < 440  
16 °C. The CO<sub>2</sub> conversion rises in Pattern A but decreases in Pattern B with the increase  
17 in flow rate. The relatively high temperature in Pattern A is negatively correlated with  
18 CO<sub>2</sub> conversion because it can stimulate the recombination of CO and O, which leads  
19 to the increase in CO<sub>2</sub> conversion with increasing flow rate (and decreasing gas  
20 temperature). A smaller injector nozzle diameter (1.0 mm) exhibits a better performance  
21 in terms of both CO<sub>2</sub> conversion and energy efficiency under most of the conditions  
22 studied, and a longer distance between the injector nozzle and electrodes is beneficial  
23 to the CO<sub>2</sub> conversion at relatively high flow rates ( $\geq 4$  L/min). A quadrangular  
24 reactor cover was proved to have a better space utilization and a higher fraction of gas  
25 treatment by plasma, which can ensure a more adequate contact between the injected  
26 gas and plasma, and thus a better performance. The optimum conditions are: flow rate  
27 = 3 L/min, nozzle diameter = 1.0 mm, distance between injector nozzle and electrodes  
28 = 5.0 mm, and a quadrangular cover, under which CO<sub>2</sub> conversion and energy efficiency  
29 up to 11.1% and 20.7% can be achieved. Compared to other typical non-thermal  
30 plasmas, such as dielectric barrier discharge and corona discharge, GAD shows a

1 significantly higher energy efficiency along with a flow rate that is an order of  
2 magnitude higher.

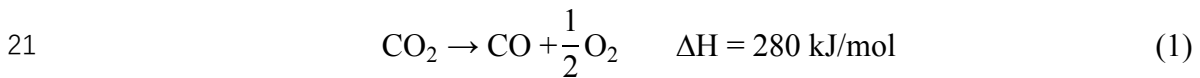
3 **Keywords:** CO<sub>2</sub> decomposition, gliding arc discharge, flow rate, reactor design, gas  
4 flow field

5

## 6 **1 Introduction**

7 Each year, human activities result in carbon emission of up to 7 Gt into the earth's  
8 atmosphere, a large portion of which is in the form of gaseous CO<sub>2</sub> [1]. The  
9 concentration of atmospheric CO<sub>2</sub> has reached 400 ppm, the highest level that has ever  
10 been recorded in the past 800,000 years [2],[3]. It is clear that increasing anthropogenic  
11 emission of greenhouse gases is leading to severe global climate change, thus  
12 motivating efforts to develop effective strategies for mitigation or valorization of CO<sub>2</sub>.  
13 In this regard, three major strategies have been proposed: boosting the use of clean and  
14 renewable energy, improving the energy utilization efficiency, and carbon capture,  
15 utilization, and storage (CCUS) [4].

16 As one of the CCUS routes, direct decomposition of CO<sub>2</sub> has attracted particular  
17 interest, because it can convert the greenhouse gas into value-added CO, a product that  
18 can serve not only as a fuel but also as a widely used chemical feedstock [5][6].  
19 However, CO<sub>2</sub> is a highly stable molecule and its dissociation requires a large amount  
20 of activation energy in traditional thermal processes (Eq. (1)) [7].



22 Thermodynamic equilibrium calculation of CO<sub>2</sub> conversion shows that CO<sub>2</sub> starts  
23 to decompose only near 2000K with a very low conversion rate (< 1%). A reasonable  
24 conversion of CO<sub>2</sub> (~60%) can be obtained only at an extraordinarily high temperature  
25 of 3000–3500K [6], which leads to a high energy cost thus rendering it undesirable in  
26 practical applications.

27 Non-thermal plasma (NTP) is increasingly considered as one of the promising  
28 approaches for CO<sub>2</sub> utilization [8]-[10], because it enables the thermodynamically  
29 unfavorable CO<sub>2</sub> activation to occur at a reduced energy cost under mild conditions,

1 i.e., lower temperature and atmospheric pressure. In non-thermal plasmas, the average  
2 electron temperature, i.e., the kinetic energy of free electrons typically in electron volts  
3 (eV), can be as high as 1–10 eV, giving rise to the generation of a variety of reactive  
4 species (e.g., excited species, radicals, ions, and photons) that are responsible for the  
5 efficient initiation and propagation of reactions [11]-[12]. Meanwhile, the gas  
6 temperature of non-thermal plasmas can be very low (e.g., 200–500 °C or even room  
7 temperature), which ensures a low energy cost due to the reduced heat loss [13]. Because  
8 of the unique merits of non-thermal plasmas, they have been widely investigated and  
9 applied to a variety of fields, such as material treatment [14]-[15], waste water disposal  
10 [16], fuel reforming [10][17], and plasma-aided combustion [18][19], etc.

11 Among various non-thermal plasmas used for CO<sub>2</sub> activation, gliding arc  
12 discharge (GAD) is attracting particular attention because it features the merits of both  
13 non-thermal and thermal plasmas [20], enabling it to simultaneously provide high  
14 power, good selectivity to chemical processes and high energy efficiency. Traditionally,  
15 a GAD reactor consists of two diverging electrodes. The arc is generated in the  
16 narrowest electrode gap, and then moves along the electrodes with increasing length  
17 due to the pushing force of gas flow. In this way, a discharge zone between the two  
18 electrodes can be generated for the plasma chemical reactions [21]. The GAD shows a  
19 relatively high electron density of up to 10<sup>20</sup> m<sup>-3</sup> [22]. The electron temperature is in the  
20 range of 0.9-2 eV (10000 K-23000 K) (typically 1 eV), which is about 5-10 times higher  
21 than the gas temperature [22][23]. It is already known that a mean electron temperature  
22 of 1 eV is most suitable for the efficient vibrational excitation of CO<sub>2</sub> [24]. In plasmas,  
23 the excitation of the asymmetric vibrational mode of CO<sub>2</sub> has been proven to be the  
24 most efficient pathway for CO<sub>2</sub> activation [25][26]. The proper electron temperature  
25 and relatively high electron density of the GAD plasma can give somewhat more  
26 vibrational excitation, indicating that a larger overpopulation of the vibrational states  
27 can be reached, and an efficient conversion of CO<sub>2</sub> is achievable in GAD [22][25].

28 There are still some drawbacks in GAD reactors which limit their energy efficiency.  
29 Although the GAD shows a somewhat 3D structure, as demonstrated by Kusano et al.

1 and Zhu et al. [14][27], the volume of plasma is limited due to the nearly 2D structure  
2 of the reactor. In addition, the gas flow speed needed is very high and the volume of the  
3 reactant gas flow is significantly larger than the plasma volume, indicating that a large  
4 part of the reactant gas cannot pass through the plasma area or pass through it only with  
5 a limited residence time. This property leads to a low gas conversion and limited  
6 efficiency [11][28]. In short, the small treated fraction is still the major limiting factor  
7 for the conversion. In this regard, an improvement in the CO<sub>2</sub> conversion performance  
8 of GAD can be clearly expected by optimizing the design of the plasma source, e.g.,  
9 shape of the cover and diameter of the injector nozzle together with the distance  
10 between the nozzle and electrodes. In this way, the gas flow field and the distribution  
11 of reactive species (e.g., excited species, radicals, ions, and photons) can be optimized  
12 to achieve a more adequate treatment of the reactant gas.

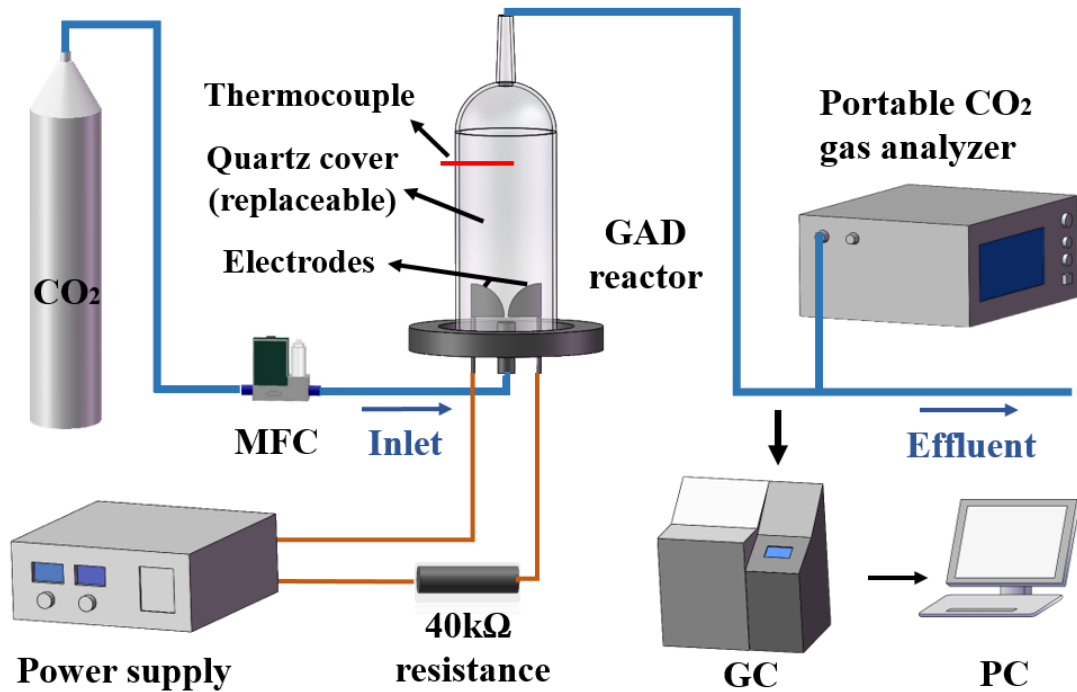
13 Therefore, this work devotes particular effort to obtaining insights toward  
14 improving the CO<sub>2</sub> activation performance in GAD by optimizing the source design.  
15 The effects of the diameter of the injector nozzle, the distance between the injector and  
16 electrodes as well as the structure of the quartz cover (cylindrical or quadrangular) on  
17 the reaction performance of CO<sub>2</sub> decomposition in GAD were investigated. Unlike the  
18 cylindrical cover that is commonly used to form the plasma chamber in the literature  
19 [29]-[31], a quadrangular cover was designed in this study in order to improve the  
20 treatment of the reactant gas by plasma. In addition, the influence of the feed flow rate  
21 has also been emphasized, as it was identified as a key parameter affecting plasma  
22 chemical reactions, especially in GAD reactors.

## 24 **2 Experimental setup**

25  
26 The schematic diagram of the experimental setup is shown in Fig. 1. The GAD  
27 reactor consists of two knife-shaped electrodes (with a length of 17 mm and base width  
28 of 2 mm), a gas nozzle and a quartz cover. The plasma was powered by a customized  
29 10 kV DC power supply (TLP2040, Teslaman). A 40 k $\Omega$  resistance was connected in  
30 series in the circuit to limit and stabilize the current. The CO<sub>2</sub> reactant gas (purity, 99.9%)

1 was injected into the reactor via the gas nozzle. A thermocouple thermometer (TM-  
 2 902C+ WRNK-81530) was placed 80 mm downstream of the electrode tip to measure  
 3 the temperature of the outlet gas. The concentration of CO<sub>2</sub> was measured in real time  
 4 with a portable CO<sub>2</sub> analyzer (GXH-3010E1, Huayun Instrument Co.). The other  
 5 gaseous products were analyzed by a gas chromatograph (GC, GC9790A, Fuli  
 6 Analytical Instrument Co.) equipped with a thermal conductivity detector (TCD) for  
 7 the measurement of O<sub>2</sub> and a flame ionization detector (FID) with a catalytic  
 8 methanation unit for detecting CO.

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Fig. 1. Schematic diagram of the experimental setup.

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Figure 2 shows the configuration of the GAD reactor with different designs for the quartz cover. The cylindrical quartz cover (Fig. 2(a)) is 24 mm in diameter and 120 mm in length. The quadrangular quartz cover (Fig. 2(b1), Fig. 2(b2)) has dimensions of 45×11×120 mm. Other variable parameters in the experiments are shown in Fig. 3, i.e., the distance between the injector nozzle and the electrodes ( $D$ ) and the diameter of the injector nozzle ( $\Phi$ ).

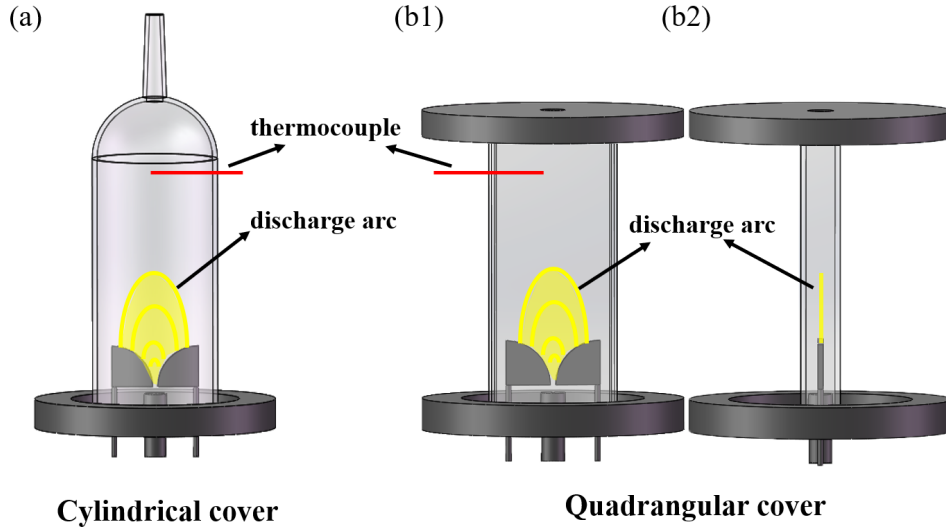


Fig. 2. GAD reactor with different quartz covers.

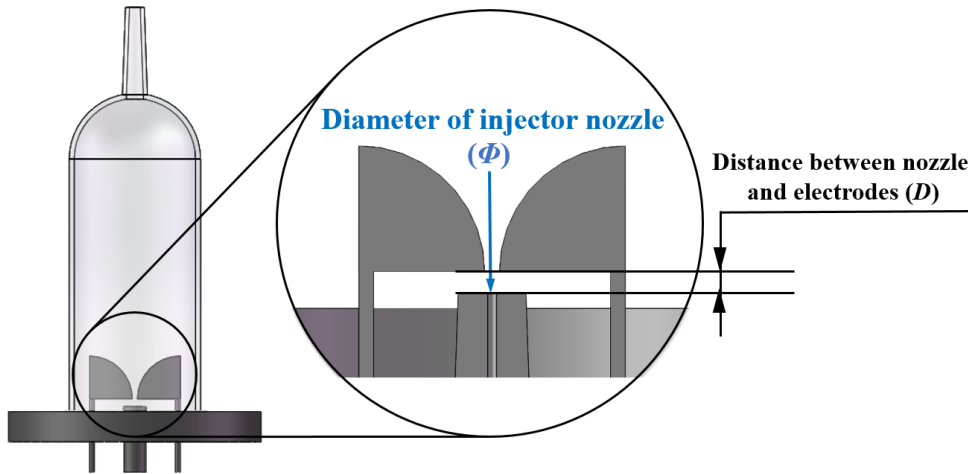


Fig. 3. Configuration of the injector nozzle and electrodes.

The performance of the GAD assisted  $\text{CO}_2$  decomposition process is presented mainly in terms of  $\text{CO}_2$  conversion  $C_{\text{CO}_2}$ , carbon balance  $B_{\text{Carbon}}$  and energy efficiency  $\eta$ , which were defined as:

$$C_{\text{CO}_2}(\%) = \frac{\text{CO}_2 \text{ converted (mol/s)}}{\text{CO}_2 \text{ introduced (mol/s)}} \times 100 \quad (2)$$

$$B_{\text{Carbon}}(\%) = \frac{\text{CO}_2 \text{ output (mol/s)} + \text{CO produced (mol/s)}}{\text{CO}_2 \text{ introduced (mol/s)}} \times 100\% \quad (3)$$

1

2

$$\eta (\%) = \frac{\text{CO}_2 \text{ converted (mol/s)} \cdot \Delta H \text{ (J/mol)}}{\text{Power (W)}} \times 100\% \quad (4)$$

3

4 Wherein  $\Delta H$  is the standard reaction enthalpy and here it is 280 kJ/mol.

5 Specific energy input (*SEI*) was used to evaluate the energy density, as defined in

6 Eq. (5).

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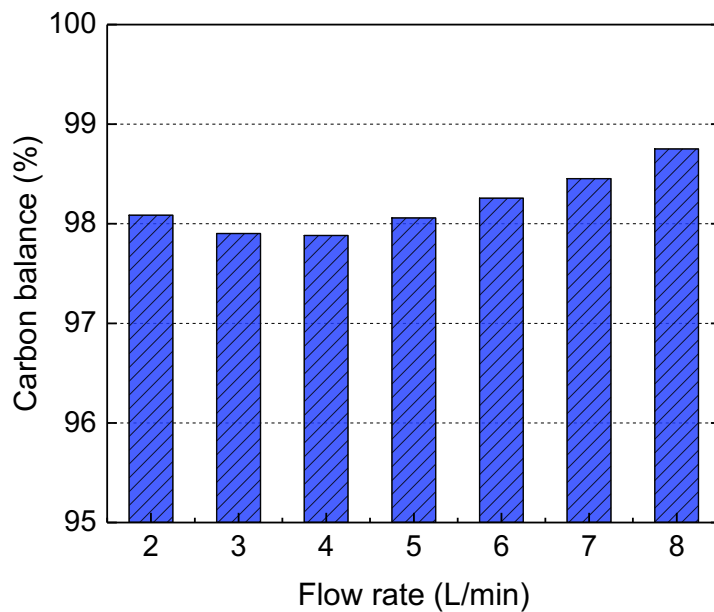
$$SEI \text{ (kJ/L)} = \frac{\text{Discharge power (W)}}{\text{CO}_2 \text{ flow rate (mL/s)}} \quad (5)$$

9

### 10 3 Results and discussion

#### 11 3.1 Effect of feed flow rate

12 In GAD, feed flow rate is an important parameter influencing the dynamic  
 13 behavior of the discharge, as reported by Zhu et al. [32], therefore its effect is  
 14 investigated in this part of the study. The feed flow rate was adjusted from 2 to 8 L/min,  
 15 with a  $\Phi$  of 1.5 mm and a  $D$  of 5.0 mm, using the cylindrical quartz cover.



16

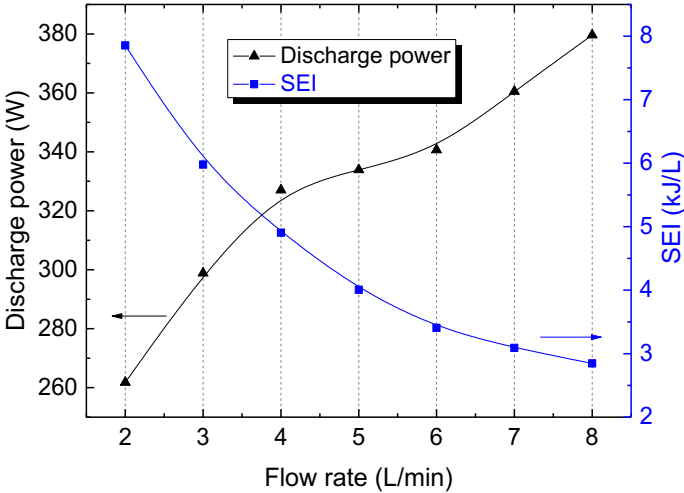
17 Fig. 4 Carbon balance of the reaction as a function of CO<sub>2</sub> feed flow rate.

18

(cylindrical cover)

1 The carbon balance of the CO<sub>2</sub> dissociation reaction in the GAD at different feed  
 2 flow rates is illustrated in Fig. 4. Clearly, the carbon balance reaches almost 100%  
 3 (ranging from 97.9 to 98.8%) under all the conditions studied. In addition, carbon  
 4 deposition was not found in the reactor. This indicates that CO is the main product and  
 5 the major reaction in the GAD assisted CO<sub>2</sub> decomposition is  $\text{CO}_2 \rightarrow \text{CO} + 1/2 \text{O}_2$ .  
 6 Therefore, in the following sections, the CO<sub>2</sub> conversion and energy efficiency will be  
 7 considered as the main indicators for evaluating the performance of the reaction,  
 8 without focusing on the selectivity and yield of products.

9 Figure 5 shows the variation in discharge power and SEI with the increasing flow  
 10 rate. As expected, a continuous decrease of SEI is observed due to the increasing  
 11 amount of feed gas. A higher breakdown power was needed to form an arc discharge at  
 12 the higher gas flow rate, leading to a higher discharge power with the increasing flow  
 13 rate [33]. In addition, the increased heat loss with rising flow rate results in the need for  
 14 a higher discharge power. Figure 6 shows the effect of flow rate on the CO<sub>2</sub> conversion  
 15 and energy efficiency. With the increase in flow rate, CO<sub>2</sub> conversion first rises to a  
 16 maximum value of 7.2% at a flow rate of 4 L/min, and then decreases. The energy  
 17 efficiency continuously rises and tends to reach a limitation after the flow rate is over  
 18 6 L/min due to the drop of the CO<sub>2</sub> conversion. The maximum energy efficiency can  
 19 reach 25.8% at a flow rate of 8 L/min.



20  
 21 Fig. 5. Effect of gas flow rate on the discharge power and SEI. (cylindrical cover)



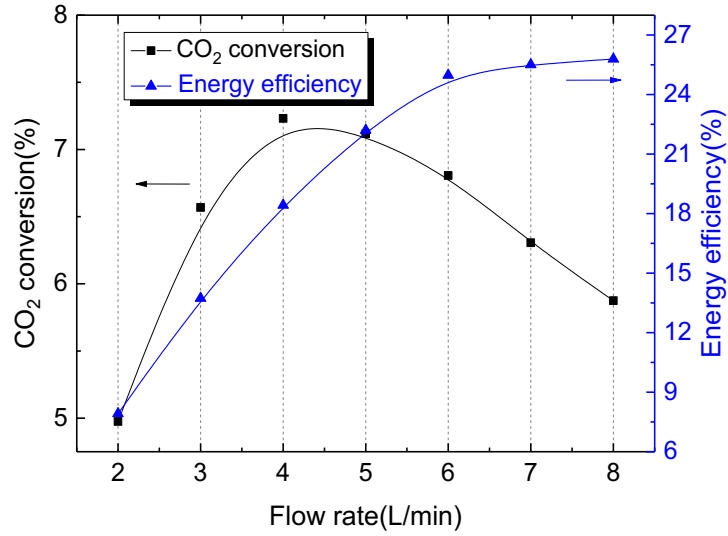


Fig. 6. Effect of gas flow rate on the CO<sub>2</sub> conversion and energy efficiency.  
(cylindrical cover)

Similar variation tendencies in energy efficiency with feed flow rate can be found in other studies [29][34]. However, the tendency toward CO<sub>2</sub> conversion is surprisingly not in line with other research on the of plasma CO<sub>2</sub> dissociation process, including GAD [29][35], dielectric barrier discharge (DBD) [36]-[38], microhollow cathode discharge [34][39], nanosecond-pulsed discharge [40] and radio-frequency discharge [41]. In these studies, increasing flow rate has a negative effect on the CO<sub>2</sub> conversion, which should be ascribed to the decreased retention time of gas in the plasma and lowered *SEI*.

The somewhat strange phenomenon in CO<sub>2</sub> conversion in this work is probably related to the reaction temperature. The rate coefficient (*k*) of the reverse reaction of CO<sub>2</sub> dissociation (see Eq. (6), Arrhenius equation) is positively correlated with the gas temperature. The gas temperature in the central area of the GAD plasma can reach 1000 K [22], which is thus high enough to enable the presence of a reverse reaction [42]. To illustrate the role of gas temperature in the GAD CO<sub>2</sub> dissociation process, the time-dependent gas temperature of the outlet gas was detected and plotted in Fig. 7.



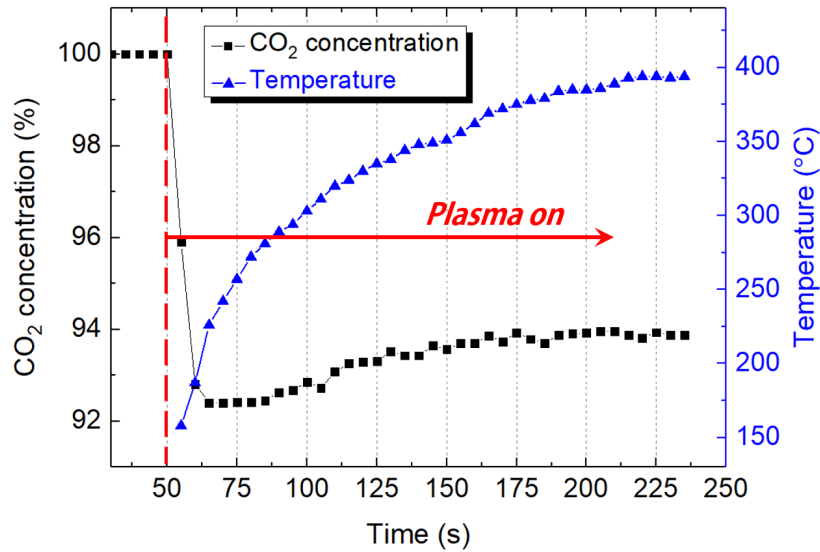


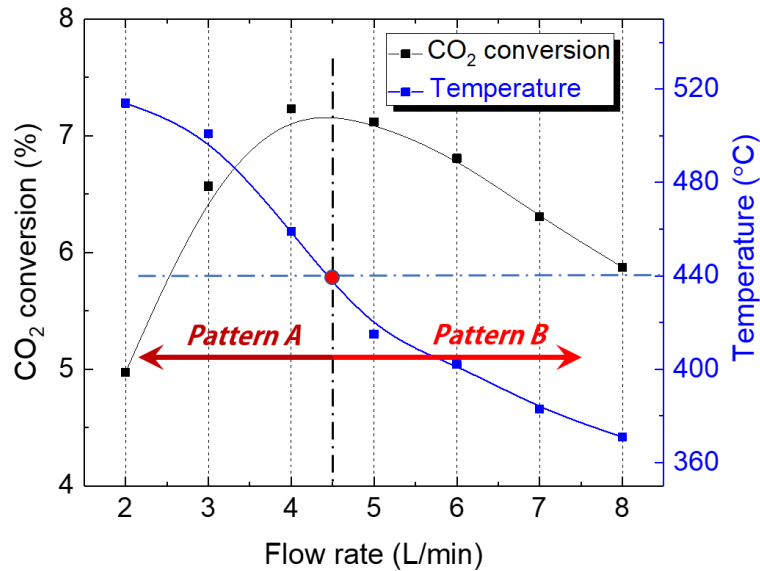
Fig. 7 Time-dependent CO<sub>2</sub> concentration and outlet gas temperature (cylindrical cover, flow rate = 5 L/min,  $D = 10.0$  mm,  $\Phi = 1.5$  mm).

Clearly, the CO<sub>2</sub> concentration has a variation trend similar to the gas temperature, i.e., it increases after the plasma is on (at 50s) and then is stabilized after 3 minutes of operation. It should also be noted that with the increase in feed flow rate (and decrease in gas temperature), the time-dependent variation of online CO<sub>2</sub> concentration was significantly weakened. These phenomena partly manifest the negative effect of high temperature on CO<sub>2</sub> decomposition in the GAD plasma, especially at low feed flow rate. That is similar to the study by Faisal et al., who explained that the conversion of toluene slightly decreases when increasing the temperature as a result of the increasing rate of the recombination reaction of the CO and O radicals [43].

In order to further understand the role of gas temperature under different conditions, the outlet gas temperature (3 mins after plasma is on) and CO<sub>2</sub> conversion are plotted in Fig. 8 with increasing flow rates. Interestingly, a maximum value at a flow rate of around 4.3 L/min in the CO<sub>2</sub> conversion curve is clearly observed, which can divide the variation profile into two patterns, as shown in Fig. 8. The CO<sub>2</sub> conversion increases with increasing gas flow rate in Pattern A; Whereas, after the maximum value, a completely opposite tendency in CO<sub>2</sub> conversion presents in Pattern B. The variation profile of CO<sub>2</sub> conversion with increasing feed flow rate could be the

1 result of several factors, e.g., decreasing *SEI*, decreasing gas temperature as well as  
 2 decreasing retention time of CO<sub>2</sub> in plasma. Considering the negative effect of high  
 3 temperature on CO<sub>2</sub> conversion, the drop in gas temperature and the decrease in the  
 4 retention time are both probably responsible for the enhancement of CO<sub>2</sub> conversion in  
 5 Pattern A. Meanwhile, from our previous study, an increasing gas flow rate can enhance  
 6 the vibrational kinetics energy of a rotating gliding arc plasma [44], which could  
 7 probably vibrationally excite CO<sub>2</sub> to a more efficient pathway and improve the  
 8 efficiency of CO<sub>2</sub> conversion to some extent. In Pattern B, the negative effect of gas  
 9 temperature is weakened with decreasing temperature, and thus the CO<sub>2</sub> conversion  
 10 starts to drop with decreasing retention time and *SEI*.

11 As indicated in Fig. 8, the outlet gas temperature at the turning point is around 440  
 12 °C. When the results of other typical conditions (see Table 1) in this work are plotted in  
 13 one figure (see Fig. 9), it is interesting to find that almost the same turning point in the  
 14 outlet gas temperature (440 °C) for CO<sub>2</sub> conversion can be observed. These phenomena  
 15 allow us to make a plausible conclusion that, with increasing gas temperature to a  
 16 certain value, the negative effect of high temperature starts to dominate the conversion  
 17 of CO<sub>2</sub>.



18

19 Fig. 8 CO<sub>2</sub> conversion and outlet gas temperature at different flow rates.

20

(Cylindrical cover)

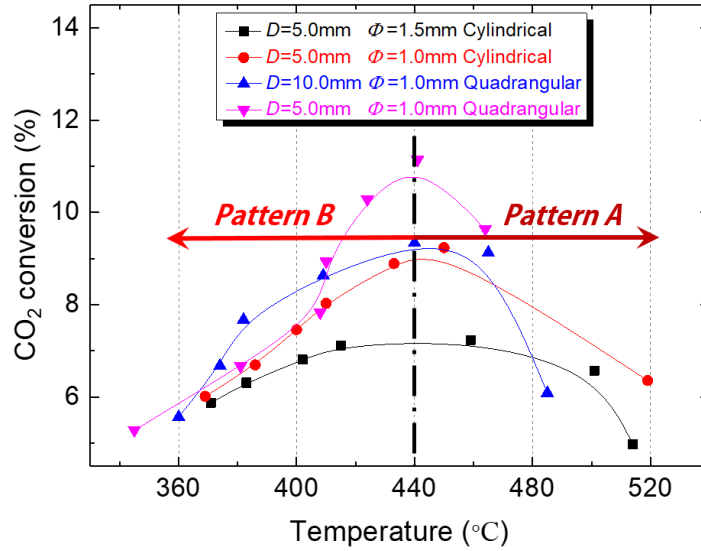


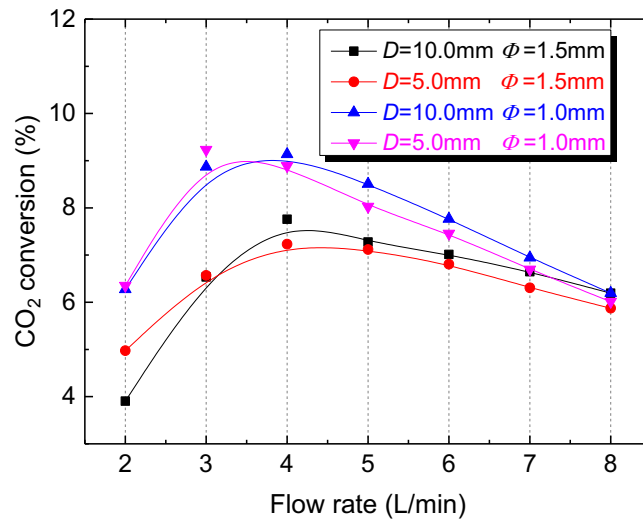
Fig. 9 Relation of the outlet gas temperature and CO<sub>2</sub> conversion under different conditions (see Table 1).

Table 1. Experimental conditions

Experiment	$D$ (mm)	$\Phi$ (mm)	Flow rate (L/min)	Cover shape
(1)	5.0	1.5	2-8	Cylindrical
(2)	5.0	1.0	2-8	Cylindrical
(3)	10.0	1.0	2-8	Quadrangular
(4)	5.0	1.0	2-8	Quadrangular

### 3.2 Effect of injector nozzle configuration

The configuration of the injector nozzle is an important parameter that can influence the gas flow field and plasma characteristics, and thus the conversion of CO<sub>2</sub>. In order to determine the optimized configuration for the injector nozzle, the injector nozzle diameter ( $\Phi$ ) and the distance between injector and electrodes ( $D$ ) were adjusted while using the cylindrical quartz cover. The conditions studied in this part are listed in Table 2. CO<sub>2</sub> conversions under different conditions are schematically shown in Fig. 10.



1

2 Fig. 10 Effects of  $\Phi$  and  $D$  on the CO<sub>2</sub> conversion with increasing feed flow rate.

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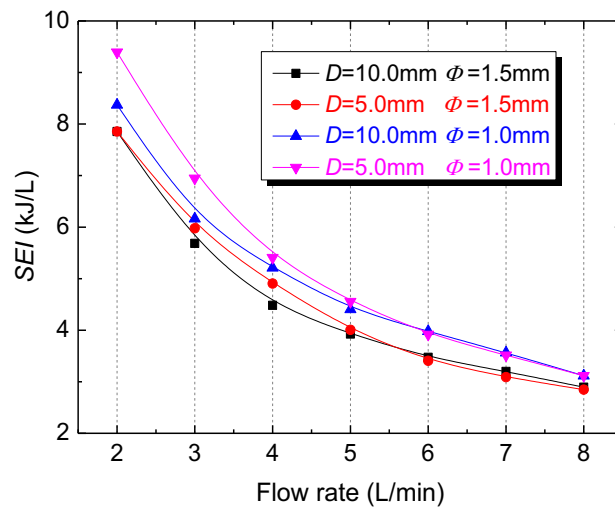
(Cylindrical cover)

4

5

Table 2. Experimental conditions

Experiment	$D$ (mm)	$\Phi$ (mm)	Flow rate (L/min)	Cover shape
(5)	10.0	1.5	2-8	Cylindrical
(6)	5.0	1.5	2-8	Cylindrical
(7)	10.0	1.0	2-8	Cylindrical
(8)	5.0	1.0	2-8	Cylindrical



6

7 Fig. 11 Effects of  $\Phi$  and  $D$  on the SEI with increasing feed flow rate.

8

(Cylindrical cover)

1

2 As seen from Fig. 10, the four curves with different nozzle configurations show  
3 the same trend in variation with increasing feed flow rate. The CO<sub>2</sub> conversion rises  
4 when the flow rate is lower than 4 L/min and then decreases when further increasing  
5 the flow rate. However, it is obvious that the four curves can be divided into two major  
6 categories depending on the  $\Phi$ . The efficiency of CO<sub>2</sub> conversion with  $\Phi = 1.0$  mm is  
7 clearly higher in comparison to  $\Phi = 1.5$  mm, especially at a gas flow rate  $\leq 7$  L/min.  
8 Young et al. also reported that a smaller nozzle diameter has a better performance when  
9 using a GAD reactor to decompose benzene [45]. A reasonable explanation for this  
10 phenomenon is given below. At a fixed flow rate, a smaller nozzle diameter gives rise  
11 to a higher gas injection velocity, which ensures a shorter retention time of gas in the  
12 plasma. As mentioned earlier, the retention time may play a negative role in the  
13 conversion of CO<sub>2</sub>, especially at a high gas temperature, due to the recombination  
14 reaction of CO and O<sub>2</sub>. In this regard, a decreased retention time with smaller  $\Phi$  is  
15 possibly responsible for the higher CO<sub>2</sub> conversion. A phenomenon that can partially  
16 manifest this explanation is, decreasing flow rate (and thus increasing gas temperature)  
17 results in a larger difference of CO<sub>2</sub> conversion between the results of  $\Phi = 1.0$  mm and  
18 1.5 mm, as clearly shown in Fig. 10. Meanwhile, a higher gas injection velocity needs  
19 a higher breakdown power to produce a discharge arc [33], as shown from the *SEI*  
20 plotted in Fig. 11.

21 The distance between the electrodes and nozzle does not exhibit any significant  
22 impact on the CO<sub>2</sub> conversion, as shown in Fig. 10. Generally, a longer distance leads  
23 to a better conversion performance, especially at a high flow rate  $> 4$  L/min.

24 The optimized results are CO<sub>2</sub> conversion = 9.1% with energy efficiency of 21.9%  
25 and flow rate of 4 L/min in experiment (7) and CO<sub>2</sub> conversion = 9.2% with energy  
26 efficiency of 16.6% and flow rate of 3 L/min in experiment (8). The former has better  
27 energy efficiency due to the higher flow rate.

28 In summary, the configuration of the nozzle (especially the nozzle diameter) has a  
29 significant impact on CO<sub>2</sub> conversion. The optimum conditions for flow rate  $\leq 3$  L/min

1 are  $D = 5.0$  mm and  $\Phi = 1.0$  mm, while for a flow rate of 4 to 8 L/min,  $D = 10.0$  mm  
2 and  $\Phi = 1.0$  mm show a better performance.

### 3.3 Effect of quartz cover structure

5 As mentioned above, the structure of the quartz cover is associated with the gas  
6 flow field and distribution of reactive species in the plasma, and is thus important for  
7 the efficiency of CO<sub>2</sub> conversion. Two types of covers, i.e., cylindrical and  
8 quadrangular ones, were investigated when the  $D$  and  $\Phi$  were fixed at 5.0 mm and 1.5  
9 mm, respectively. Figure 12 and 13 show the variation in CO<sub>2</sub> conversion and energy  
10 efficiency with increasing flow rate with different reactor covers, respectively. When  
11 using the cylindrical cover, CO<sub>2</sub> conversion rises from 6.4% to a maximum of 9.2%  
12 with the flow rate increasing from 2 to 3 L/min, and then drops nearly linearly to 6.0%  
13 when further increasing the flow rate from 3 to 8 L/min. The energy efficiency  
14 continuously grows from 8.5 to 24.1% but with a gradually slower growth rate with  
15 increasing flow rate. For the quadrangular cover, the variation profiles of both the CO<sub>2</sub>  
16 conversion and energy efficiency with increasing flow rate are similar to that of the  
17 cylindrical cover. The CO<sub>2</sub> conversion increases from 9.6 to 11.1% first and then  
18 gradually drops to 5.3%. The energy efficiency increases from 16.1 to 25.2% with flow  
19 rate increasing from 2 to 5 L/min, then slightly fluctuates between 23.4 and 25.9%.  
20 Interestingly, if the energy efficiency is defined as (lower heating value in the product  
21 CO – J/mol)  $\times$  (CO flow rate – mol/s) / (discharge power in W), the energy efficiency  
22 under the conditions of the best CO<sub>2</sub> conversion (flow rate = 3 L/min,  $\Phi = 1.0$  mm,  $D$   
23 = 5.0 mm, and a quadrangular cover) should be 21.4%.

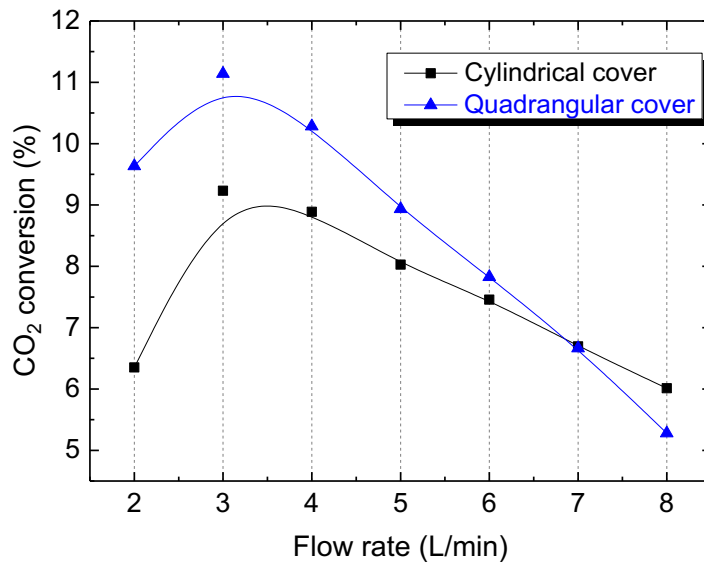
24 Clearly, the quadrangular cover exhibits a better performance than the cylindrical  
25 cover in terms of both CO<sub>2</sub> conversion and energy efficiency. In the cylindrical cover,  
26 a large amount of unreacted gas bypasses the plasma area, causing an inefficient  
27 conversion of CO<sub>2</sub>. The quadrangular cover can gather the gas forcibly to the central  
28 reactive area of the plasma, which enables a better space utilization and thus a higher  
29 possibility of the reaction between CO<sub>2</sub> gas and plasma. In addition, the cross section

1 area of the quadrangular cover ( $4.95 \text{ cm}^2$ ) is smaller than that of the cylindrical ( $18.10$   
2  $\text{cm}^2$ ) one, resulting in a higher gas velocity (and lower retention time of gas in the  
3 plasma) in the quadrangular cover at the same flow rate. As mentioned above, a lower  
4 retention time at a relatively high gas temperature could be helpful toward a higher  $\text{CO}_2$   
5 conversion, which is probably also responsible for the better performance of the  
6 quadrangular cover. At a lower flow rate, a larger difference in both the  $\text{CO}_2$  conversion  
7 and energy efficiency between the two covers can be clearly seen, partly manifesting  
8 the role of retention time in  $\text{CO}_2$  conversion, because the gas temperature decreases  
9 with increasing flow rate.

10 To better understand the effect of the shape of the cover on the gas flow field, the  
11 distributions of the gas flow field and gas velocity inside the GAD have been simulated  
12 and calculated using COMSOL Multiphysics 5.2 software (three-dimensional laminar  
13 flow module), as shown in Fig. 14. Clearly, the gas speed is higher and more injected  
14 gases are concentrated in the reaction area when using the quadrangular cover because  
15 of the controlled space. Meanwhile, the downstream configuration of the quadrangular  
16 cover ensures the formation of a recirculation region (see Fig. 14) that can remarkably  
17 increase the fraction of gas treatment by plasma because a large amount of unreacted  
18 gas can return to the reaction area for further reaction.

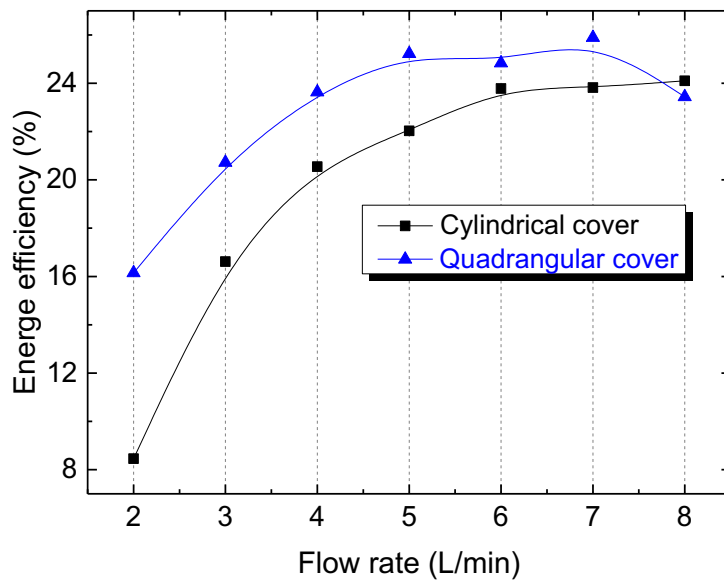
19 In short, the quadrangular cover shows a better processing efficiency in  
20 comparison to the cylindrical cover, especially at a lower flow rate, due to a better space  
21 utilization and a higher fraction of gas treatment by the plasma. Therefore, this work  
22 provides valuable insights into optimizing a GAD reactor by optimizing the cover  
23 structure.





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3

Fig. 12. Effect of quartz cover structure on the CO<sub>2</sub> conversion.



4  
5

Fig. 13. Effect of quartz cover structure on the energy efficiency.



1

## different non-thermal plasmas

Plasma type	Flow rate (mL/min)	CO <sub>2</sub> conversion (%)	Energy efficiency (%)	Ref.
GAD	3000	11.1	20.7	This work
GAD	14000	9.0	18.0	[5]
GAD	850	18.0	-	[29]
GAD	1000	8.6	30.0	[35]
GAD	6500	10.0	20.0	[46]
MW	5000	30.0	40.0	[47]
MW	2200	90.0	10.0	[48]
ns-pulse	120	7.3	11.5	[40]
Corona	90	10.9	1.7	[49]
DBD		24.5	2.4	[37]
DBD	10	35.0	2.0	[50]
DBD	40	12.5	3.5	[51]
DBD	25	27.2	2.8	[52]
DBD	100	25.2	5.3	[53]
Thermal splitting	20	22.0	-	[54]

2

3 As shown in Table 3, the optimum CO<sub>2</sub> conversion and energy efficiency were  
4 reported using a MW plasma reaching up to 30% and 40%, or 90% and 10%,  
5 respectively, at a flow rate of 5000 or 2200 mL/min [47][48]. However, these results  
6 were observed only at a low-pressure condition of around 150 mbar, leading to  
7 significantly extra energy needed for vacuum supply. This is undoubtedly industrially  
8 undesirable. Clearly, DBD has a remarkable CO<sub>2</sub> conversion but with a fairly low  
9 energy efficiency of normally < 5%. Both corona discharge and nanosecond-pulsed  
10 discharge present a limited performance in CO<sub>2</sub> conversion and energy efficiency. In  
11 addition, DBD, nanosecond-pulsed discharge and corona discharge have extremely low  
12 flow rates which indicate a poor processing capacity and thus are industrially  
13 unfavorable. In general, GAD has a good performance in terms of both CO<sub>2</sub> conversion  
14 and energy efficiency. In our work, the CO<sub>2</sub> conversion reached 11.1% and the energy  
15 efficiency reached 20.7% while the flow rate reached 3000 mL/min. Not only a  
16 favorable efficiency but also a remarkably high processing capacity. It should be noted  
17 that in comparison to the conventional means of CO<sub>2</sub> decomposition, like solid-oxide  
18 electrolytic cell, where a current efficiency of up to 69% can be obtained [55], the GAD

1 plasma technology shows a relatively low energy efficiency but a remarkably high  
2 processing capacity. In addition, the unique properties of non-thermal plasma, e.g.,  
3 instant on/off, high specific production, low investment and operational cost, operation  
4 under atmospheric pressure and low temperature and potential to directly use the  
5 electricity produced from intermittent renewable energies (e.g., solar and wind), make  
6 it promising for CO<sub>2</sub> decomposition.

7 Considering the negative effect of high gas temperature (especially at low flow  
8 rates) in this GAD plasma, an enhancement in the CO<sub>2</sub> dissociation performance can  
9 be expected by cooling the plasma reaction area to inhibit the reverse reaction of CO<sub>2</sub>  
10 decomposition. In addition, as previously reported [56][57], the shape and length of the  
11 electrodes can also influence the reaction performance in GAD plasma, which is  
12 possibly another fruitful direction to improve the CO<sub>2</sub> dissociation performance.

#### 14 **4. Conclusion**

15 A gliding arc discharge (GAD) was investigated in this work for CO<sub>2</sub>  
16 decomposition. The influences of flow rate, distance between the injector nozzle and  
17 electrodes, diameter of injector nozzle and structure of the quartz cover on the direct  
18 dissociation of CO<sub>2</sub> were emphasized, providing insights on the route for improving the  
19 performance of a GAD assisted CO<sub>2</sub> activation process by optimizing the source design.

20 The trend in CO<sub>2</sub> conversion with increasing gas flow rate shows two patterns:  
21 Pattern A with flow rate < 4 L/min (outlet gas temperature > 440 °C) and Pattern B with  
22 flow rate > 4 L/min (outlet gas temperature < 440 °C). The CO<sub>2</sub> conversion increases  
23 in Pattern A but decreases in Pattern B. With increasing plasma gas temperature to a  
24 certain value (at outlet gas temperature of 440 °C), the negative effect of high  
25 temperature starts to dominate the conversion of CO<sub>2</sub> by stimulating the recombination  
26 reaction of CO and O<sub>2</sub>. In this case, a longer retention time of gas in the plasma may  
27 also provide a negative effect on the CO<sub>2</sub> conversion. The rising CO<sub>2</sub> conversion in  
28 Pattern A with increasing flow rate probably results from the decreased gas temperature  
29 and retention time.

1 A smaller injector nozzle diameter of 1.0 mm is more beneficial for CO<sub>2</sub>  
2 conversion compared with that of 1.5 mm under most of the conditions studied, because  
3 of the lower retention time in the plasma area. The 10 mm distance between the injector  
4 nozzle and electrodes shows a better performance in comparison to the 5 mm at  
5 relatively high flow rate (flow rate  $\geq$  4 L/min).

6 A quadrangular cover clearly shows a better performance than the traditional  
7 cylindrical cover, because of the better space utilization and enhanced fraction of gas  
8 treatment by plasma under the optimized gas flow field.

9 A smaller nozzle diameter of 1.0 mm, a distance between the electrodes and nozzle  
10 of 5.0 mm and a flow rate of 3 L/min with a quadrangular cover are suggested to  
11 simultaneously obtain a relatively high CO<sub>2</sub> conversion (11.1%) and energy efficiency  
12 (20.7%) in the GAD reactor.

13

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