

1 **Hollow SiC@MnO₂ nanospheres with tunable core size and shell**
2 **thickness for excellent electromagnetic wave absorption**

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20 **Abstract**

21 The hollow microstructure can permit more incident waves to enter the absorber
22 and increase the attenuation ability by multiple reflections and diffraction. In addition,
23 different substances can introduce the heterogeneous interface and further attenuate
24 electromagnetic waves through interfacial polarisation relaxation. Here, hollow
25 SiC@MnO₂ nanospheres are synthesised with tunable hollow SiC core size and
26 flower-like layered MnO₂ shell thickness. By adjusting the concentration of KMnO₄,
27 different shell thicknesses of hollow SiC@MnO₂ nanospheres can be prepared,
28 ranging from 40 nm to 110 nm. Furthermore, hollow SiC@MnO₂ composites with
29 different inner diameters between 300 nm to 470 nm can be obtained by tailoring the
30 amount of tetraethyl orthosilicate. The results indicate that the effective absorbing
31 bandwidth of hollow SiC@MnO₂ can reach 5.71 GHz with a core size of 360 nm and
32 a shell thickness of 90 nm at only 1.8 mm. This work provides a valuable core-shell
33 strategy of hollow SiC@MnO₂ towards excellent electromagnetic wave-absorbing
34 properties.

35 **Keywords:** H-SiC@MnO₂ composite, core-shell, hollow, absorber, electromagnetic
36 wave

37 **1 Introduction**

38 As stepping into a new era of fifth-generation technology, web-connected devices
39 and wireless equipment are everywhere. Along with those advances, a consequence is
40 that a new electromagnetic wave (EMW) pollution is exposing living organisms to
41 potential health risks. To decrease the radiation and minimise harmful effects,
42 researchers have increasingly engaged in finding various EMW absorption materials,
43 including carbon-based materials [1-4], magnetic components [5-7], conductive
44 polymers [8-11], and ferrite [12, 13]. However, challenges remain, and the practical
45 applications are not straightforward. Silicon carbide (SiC) has been widely used in the
46 field of EMW absorption due to its high-temperature oxidation resistance, high
47 melting point, high strength, low thermal expansion, low density, and good chemical
48 inertness [14]. The electromagnetic wave absorption performance of pure SiC is poor
49 [15-17]. Hence, hollow structures [18, 19] and compounding with other materials [20-
50 22] have been studied to improve its electromagnetic absorption performance. More
51 attractively, the hollow structure can further lower the weight of the products, which
52 can achieve low mass density and a thinner absorber.

53 The wide attention to core-shell structure mainly lies in the diverse composition
54 and heterogeneous microstructure. For instance, Zhong et al. [23] synthesised bulk
55 core-shell SiC@Ti₃SiC₂ through molten salt and the effective absorption bandwidth
56 reached 5.6 GHz at 2.3 mm. Yan et al. [24] employed one-step chemical vapour
57 infiltration to grow SiC nanowires on the surface of carbon fibres and improved the
58 EMW absorbing performance of carbon fibres from -5.58 dB to -45.98 dB. Very
59 recently, some reports have been detailing the use of magnetic materials as the core to
60 further enhance the EMW absorption ability. For example, Li et al. [25] prepared
61 flower-like core-shell Fe/Fe₄N@SiO₂ composites with the widest effective absorption

62 bandwidth of 6.1 GHz at 1.16 mm. It is widely accepted that there are many choices
63 of materials and preparation methods for different core-shell structures and
64 morphology to improve the EMW properties of the composites. Commonly, the
65 EMW absorption ability with fixed components can be adjusted through the shell
66 thickness and core size. And compared with other irregular morphologies, the hollow
67 spherical core-shell structure is easy to obtain the consistency of property because of
68 the isotropic. However, the uniform monodisperse hollow spherical core-shell
69 structure with a tunable core diameter and shell thickness is rarely studied.

70 MnO_2 is one of the most abundant materials on earth. It has many outstanding
71 advantages, including significant dielectric losses, low cost, good chemical durability
72 and thermal stability, low toxicity and simple preparation methods [26-28]. These
73 unique properties make it a promising EMW absorber [29-31]. Recent reports have
74 also shown that MnO_2 compounding with other materials can further improve EMW
75 absorbing performance [32]. Li et al. [33] successfully prepared MnO_2 -covered cotton
76 cloths, and the effective bandwidth is 5.84 GHz at 2.0 mm. Zhang et al. [34]
77 synthesised CoFe@C@MnO_2 hierarchical nanocubes through a MOF-driven process,
78 and the minimum reflection loss is -64 dB at 1.3 mm. Until now, no report has been
79 synthesising hollow SiC@MnO_2 with tunable core size and shell thickness and
80 studying their EMW absorbing performance.

81 Here in this work, SiO_2 spheres are used as sacrificial templates to prepare hollow
82 SiC . Then MnO_2 is covered with hollow SiC through a hydrothermal strategy to
83 prepare a uniform core-shell hollow SiC@MnO_2 structure. Adjusting the shell
84 thickness and hollow core diameter can obtain a series of hollow SiC@MnO_2 with
85 customised EMW properties. When the shell thickness is 90 nm and the core diameter
86 is 360 nm, the product has the optimal EMW absorbing performance. The

87 corresponding frequency region with reflection loss lower than -10 dB can reach 5.71
88 GHz at only 1.8 mm. Our results highlight that the hollow spherical structure
89 combined with shell evolution is a valuable strategy to boost EMW absorbing
90 capacity.

91 **2 Experimental sections**

92 **2.1 Materials**

93 Tetraethyl orthosilicate (TEOS, Macklin, China), ammonium hydroxide solution
94 ($\text{NH}_3\cdot\text{H}_2\text{O}$, 25-28 wt%, Nanjing Reagent, China), ethanol absolute (EtOH, Shanghai
95 LingFeng Chemical Reagent Co., Ltd, China), resorcinol (Macklin, China),
96 formaldehyde (37-40 wt%, Macklin, China), potassium permanganate (KMnO_4 ,
97 Sinopharm Chemical Reagent Co., Ltd, China), hydrofluoric acid (HF, 40 wt%,
98 Macklin, China). All materials, including distilled water (DI) were used without
99 further purification.

100 **2.2 Preparation of uniform hollow SiC (H-SiC)**

101 H-SiC was synthesised through the sacrificial templating method. Firstly, 18 ml
102 TEOS was added to the solution of 400 ml EtOH, 56 ml DI and 24 ml $\text{NH}_3\cdot\text{H}_2\text{O}$ and
103 magnetically stirred for 1 hour. Secondly, 3.2 g resorcinol and 4.5 ml formaldehyde
104 were added and stirred for another 24 hours. Then the products were put in a tube
105 furnace to react at 1350 °C for 4 hours under argon protection. Finally, H-SiC was
106 obtained after etching in 20 wt% HF for 12 hours and unreacted carbon removal at
107 650 °C for 3 hours under an air atmosphere.

108 **2.3 Preparation of tunable shell thickness of H-SiC@MnO₂**

109 H-SiC@MnO₂ were prepared through the hydrothermal method. First, 40 mg H-
110 SiC was added to 10 mM KMnO_4 solution and ultrasonic for 30 minutes. Then the
111 mixture was transferred to a 100 ml Teflon and heated at 140 °C for 2 hours. The H-

112 SiC@MnO₂ can be obtained after centrifugation, washed with DI three times and
113 dried at 80 °C for 4 hours. In order to adjust the shell thickness of MnO₂, different
114 concentrations of KMnO₄ solution were prepared. The resultant core-shell H-
115 SiC@MnO₂ for 5 mM, 10 mM, 20 mM, 40 mM and 60 mM KMnO₄ solution was
116 named H-SiC@MnO₂-5, H-SiC@MnO₂-10, H-SiC@MnO₂-20, H-SiC@MnO₂-40
117 and H-SiC@MnO₂-60, respectively.

118 **2.4 Preparation of controllable size of H-SiC@MnO₂**

119 The size of the sacrificial template determines the size of the hollow core.
120 Different sizes of templates can form by changing the volume of TEOS from 18 ml to
121 8 ml and 28 ml. After the same reaction and treatment, the different hollow diameters
122 of H-Si@MnO₂ at the same concentration of 10 mM KMnO₄ solution can be obtained
123 and labelled as H-SiC@MnO₂-8, H-SiC@MnO₂-18 and H-SiC@MnO₂-28.

124 **2.5 Characterisation**

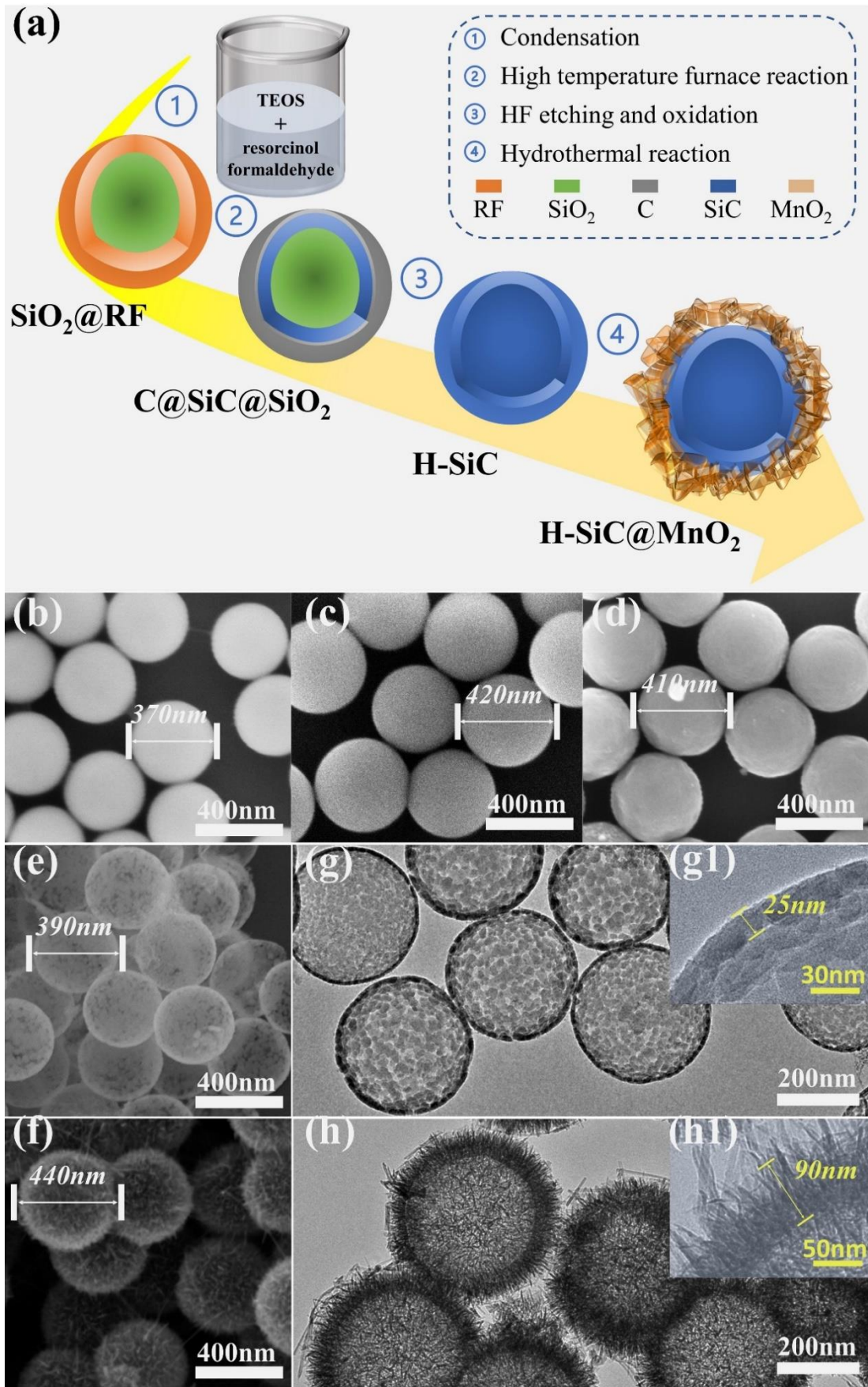
125 The products' crystalline phases and phase purity were tested by X-ray diffraction
126 (XRD), operated at 40 kV and 40 mA and the 2-theta from 10 ° to 80 °. The valence
127 of H-SiC@MnO₂ was analysed by an X-ray photoelectron spectrometer (XPS). The
128 microstructure of the products was observed through a field emission scanning
129 electron microscope (FESEM). The hollow morphology was tested on a transmission
130 electron microscope (TEM). The specific surface area and pore size of H-SiC@MnO₂
131 were tested by a Nitrogen adsorption-desorption analyser (BET). The electromagnetic
132 wave absorption performance was tested by a vector network analyser between 2 GHz
133 and 18 GHz using a circular ring obtained by mixing with paraffin with a filling
134 weight ratio of 35 %.

135 **3 Results and discussion**

136 Figure 1 (a) depicts the preparation process of H-SiC@MnO₂ core-shell

137 composites. Firstly, SiO₂ sphere templates were prepared through the classic Stober
138 method [35]. Figure 1 (b) shows the templates have good uniformity and
139 monodispersity, and the diameter of SiO₂ spheres is around 370 nm. Then, resorcinol
140 and formaldehyde were added to the solution to form SiO₂@RF. The amount of
141 resorcinol and formaldehyde is crucial to the monodispersity of SiO₂@RF. As seen in
142 Figure 1 (c), the condensation reaction of RF resin is coated on the surface of the
143 template, and the spherical size is increased to around 420 nm. However, if the
144 concentration of resorcinol and formaldehyde is high, the microstructure of SiO₂@RF
145 is sea-island, as shown in Figure.S1. Thirdly, SiC can be obtained in the interface
146 between SiO₂ and carbonised RF resin due to the diffusion effect of SiO₂ at 1350 °C,
147 which follows the vapour-solid mechanism. Figure 1 (d) shows that the spherical
148 structure was maintained after the carbon thermal reduction, but the size shrunk
149 slightly to about 410 nm. Subsequently, H-SiC can be obtained after etching and
150 oxidation treatment. The monodisperse hollow spherical structure is clearly seen in
151 Figure 1 (e, g). The diameter and shell thickness of H-SiC is approximately 390 nm
152 and 25 nm, respectively. Finally, the H-SiC@MnO₂ can be prepared after a
153 hydrothermal reaction. The growth of MnO₂ nanosheets on the surface of hollow SiC
154 is mainly generated by the decomposition and deposition of KMnO₄. The produced
155 nanocrystal MnO₂ precipitated on the surface of hollow SiC and covered the whole
156 sphere, acting as a nucleation site to further promote the decomposition of KMnO₄.
157 Finally, the nanosheets can be formed after further depositing along the surface of
158 MnO₂. Figure 1 (f, h) show that the surface character significantly changes due to the
159 MnO₂ nanolayer coating, which maintains the hollow structure. The diameter and
160 shell thickness of H- SiC@MnO₂ increase to around 440 nm and 90 nm, respectively.
161 Furthermore, different shell thickness samples were synthesised by adjusting the

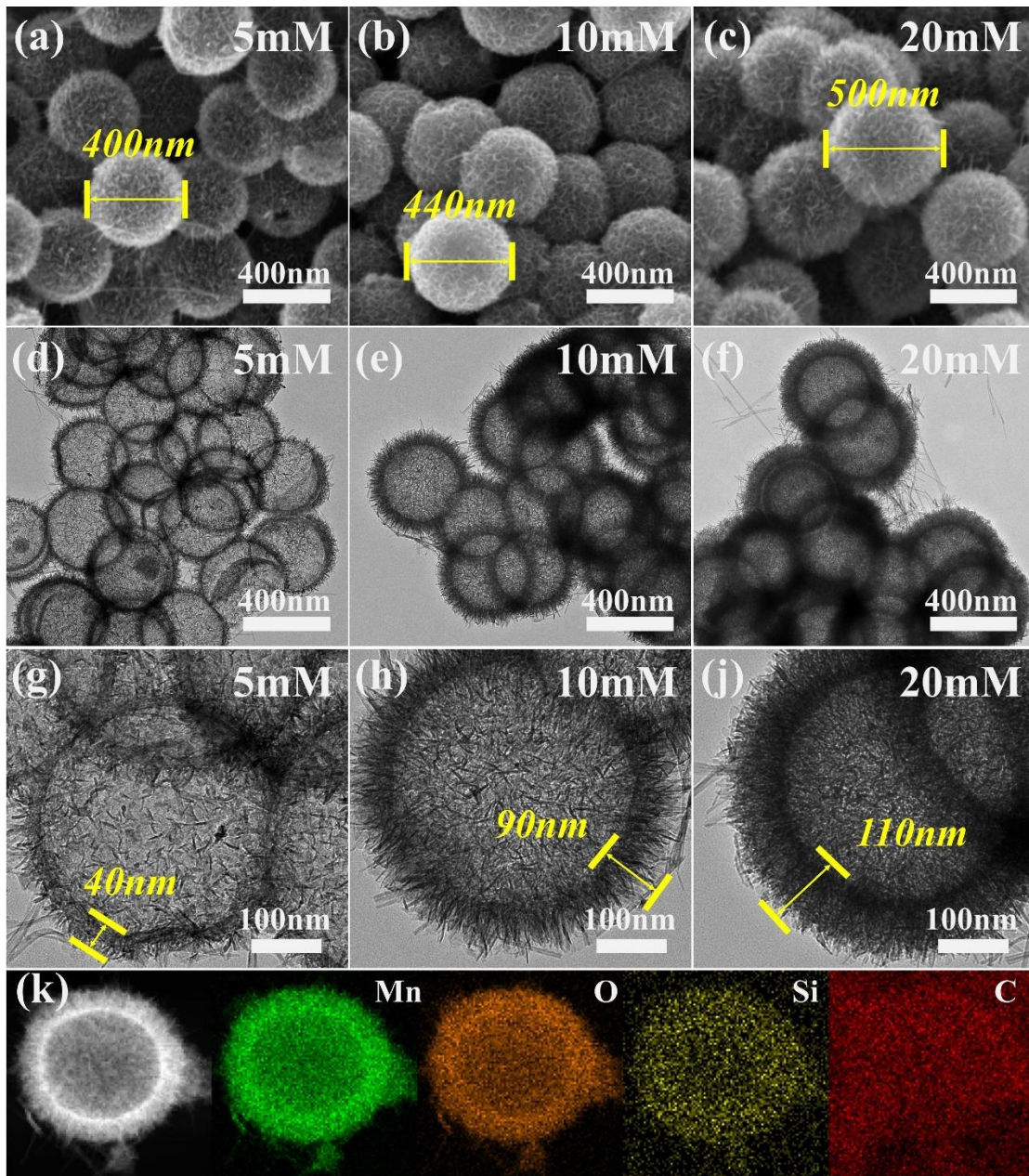
162 concentrations of KMnO_4 solution, ranging from 5 mM to 60 mM, and the
163 morphologies are displayed in Figure 2.



164

165 Figure 1 (a) Schematic diagram of the synthesis process of H-SiC@MnO₂ core-shell composites. (b-f) FESEM

166 images of products in different stages. (b) SiO₂, (c) SiO₂@RF, (d) C@SiC@SiO₂, (e) H-SiC, (f) H-SiC@MnO₂.
167 (g, h) TEM images of H-SiC and H-SiC@ MnO₂.

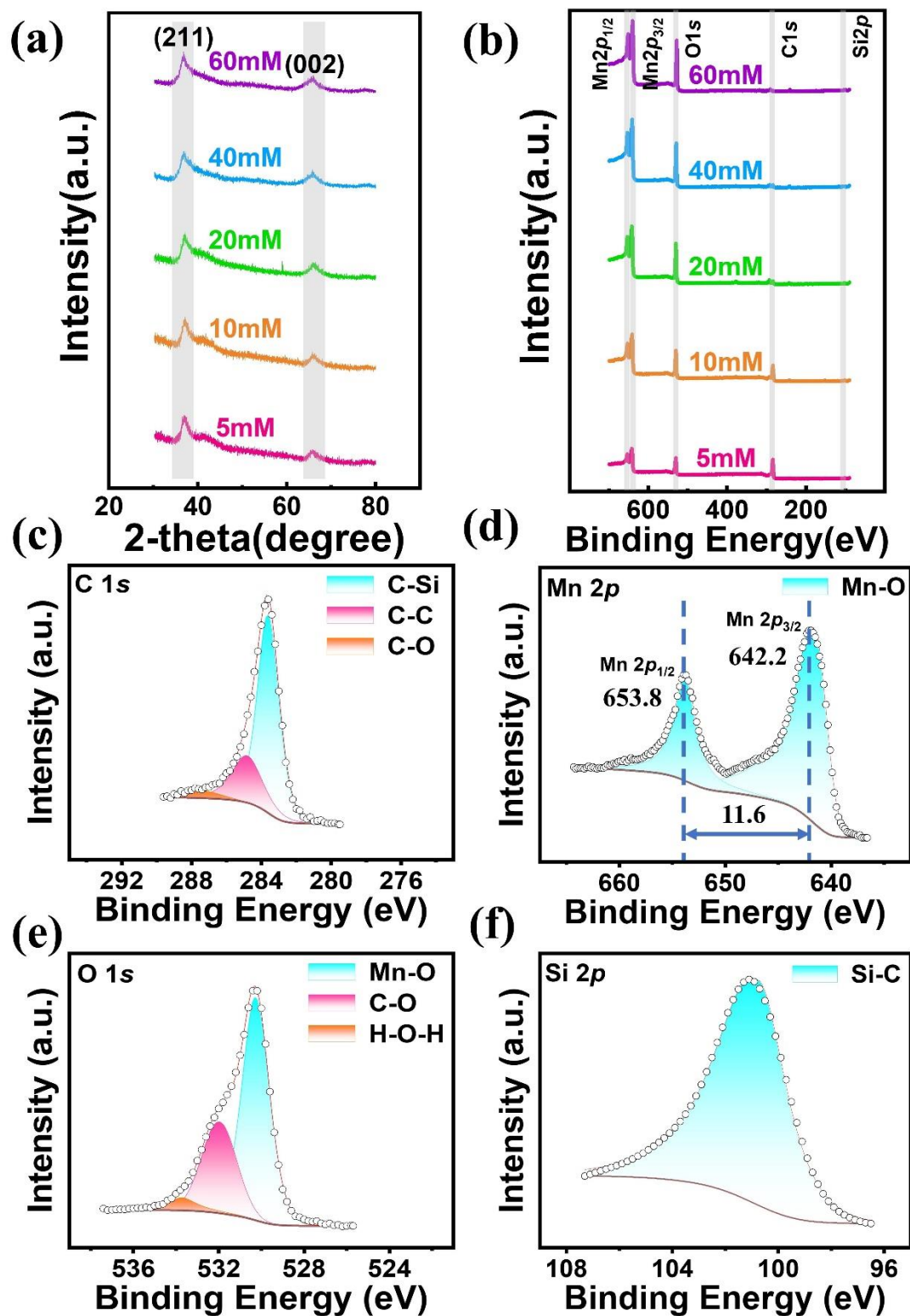


168
169 Figure 2 (a-c) FESEM images of different shell thicknesses of H-SiC@MnO₂ composites. (d-j) TEM images of
170 varying shell thickness of H-SiC@MnO₂ composites. (k) elemental mapping distribution of H-SiC@MnO₂-10.

171 As shown in Figure 2 (a-c), with the concentration of KMnO₄ solution increasing
172 from 5 mM to 20 mM, the diameter of the monodisperse spheres presents a slight
173 upward. The particle size of H-SiC@MnO₂-5, 10 and 20 is about 400, 440 and 500
174 nm, respectively. The FESEM images of H-SiC@MnO₂-40 and H-SiC@MnO₂-60 are
175 shown in Figure.S2. Due to the high concentration of KMnO₄, the spheres adhere to

176 each other, which illustrates poor dispersibility. As shown in Figure 2 (d-j), the core-
177 shell hollow structure can be distinctly seen, and as the concentration of KMnO_4
178 solution climbs, the loading thickness of the MnO_2 layer goes up, which is consistent
179 with the diameter in Figure 2 (a-c). The detailed shell structure is shown in Figure 2
180 (g-j), and the shell thickness of H-SiC@ MnO_2 -5, 10 and 20 is approximately 40, 90
181 and 110 nm, respectively. The elements mapping scan shows that Mn, O, Si and C are
182 uniformly distributed in the shell edge (Figure 2 k). The Si and C are not obvious
183 because they are inside the MnO_2 shell.

184 As shown in Figure 3 (a), the two diffraction peaks at 37° and 66° correspond to
185 the (111) and (020) planes of MnO_2 (JCPDS card no. 44-0141) [36]. SiC did not
186 appear in the XRD spectrum, probably because the MnO_2 coating is relatively thick,
187 and the detection depth of XRD cannot penetrate. For pure H-SiC, the XRD pattern is
188 shown in Figure.S3. XPS (Figure 3 b) determined the elemental composition of the
189 prepared H-SiC@ MnO_2 and confirmed the presence of Mn, C, O and Si in the H-
190 SiC@ MnO_2 composites. The detail scan spectra observed the more detailed functional
191 groups on each element (Figure 3 c-f). From Figure 3 (c), the $\text{C}1s$ display three fitting
192 peaks at 283.8, 284.8 and 286.8 eV corresponding to the C-Si, C-C and C-O,
193 respectively [37]. The Mn $2p$ spectrum in Figure 3 (d) shows two prominent peaks at
194 642.2 and 653.8 eV, coinciding with Mn $2p_{3/2}$ and Mn $2p_{1/2}$, respectively [38]. In
195 addition, the valence state of Mn is +4 due to the spin energy separation of Mn $2p$
196 being 11.6 eV [33]. The fitting peaks at 530.2, 532.2 and 533.3 eV correspond to Mn-
197 O, C-O and H-O-H groups (Figure 3 e) [39]. At the same time, there is only one
198 central peak in Si $2p$ at 101.1 eV, which belongs to Si-C bonding (Figure 3 f) [20].

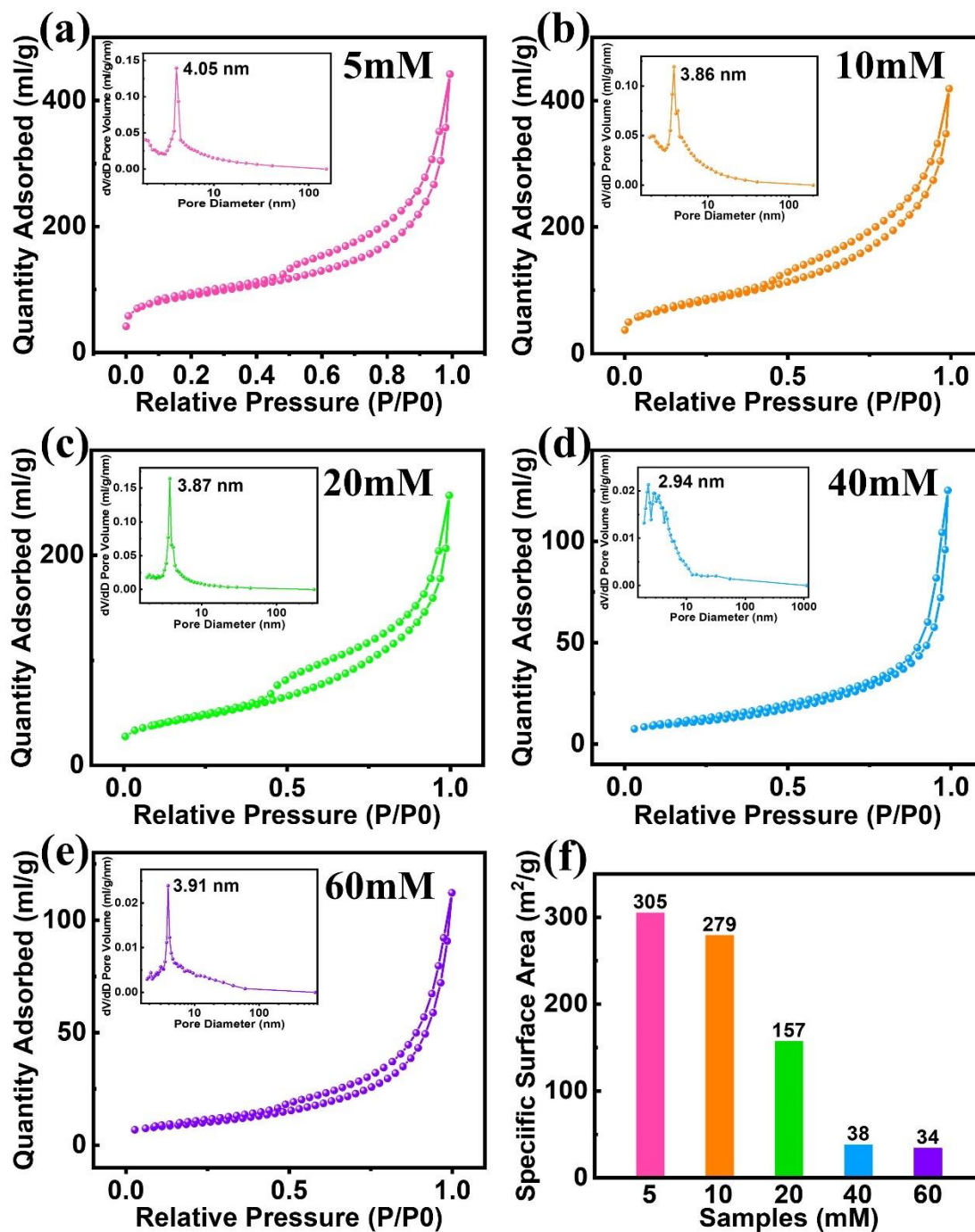


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Figure 3 (a) The XRD patterns of H-SiC@MnO₂ composites. (b) XPS survey scan of H-SiC@MnO₂ composites. (c-f) XPS detail scan of H-SiC@MnO₂-10 composite. (c) C 1s, (d) Mn 2p, (e) O 1s, (f) Si 2p.



202

203 Figure 4 Nitrogen adsorption-desorption curves of H-SiC@MnO₂ composites. (a) H-SiC@MnO₂-5, (b) H-
 204 SiC@MnO₂-10, (c) H-SiC@MnO₂-20, (d) H-SiC@MnO₂-40, (e) H-SiC@MnO₂-60. Insets in (a-e) are the
 205 corresponding pore diameter distribution curves. (f) The specific surface area of each composite is in the bar chart.

206 The specific surface area and porosity of H-SiC@MnO₂ composites were analysed
 207 using Nitrogen adsorption-desorption BET (Figure 4). As seen in Figure 4 (a-e), when
 208 the relative pressure P/P_0 is greater than 0.5, the absorbed quantity soars, showing a
 209 typical IV type adsorption isotherm curve and H3 type hysteresis ring, which indicates

210 capillary condensation and containing mesoporous in the composite [40]. The BET
 211 method calculated the specific surface area of each composite and is shown in Figure
 212 4 (f). With the increasing MnO₂ shell thickness, the specific surface area value
 213 decreased, changing from 305 m²·g⁻¹ for H-SiC@MnO₂-5 to 34 m²·g⁻¹ for H-
 214 SiC@MnO₂-60. The drop in specific surface area value was primarily due to the
 215 adhesion of spheres and the high density of MnO₂. In theory, the larger the specific
 216 surface area of the material, the greater the number of atoms on the surface, which are
 217 more likely to polarise and lead to polarisation loss [41]. The specific absorption
 218 performance of the composite is discussed as follows.

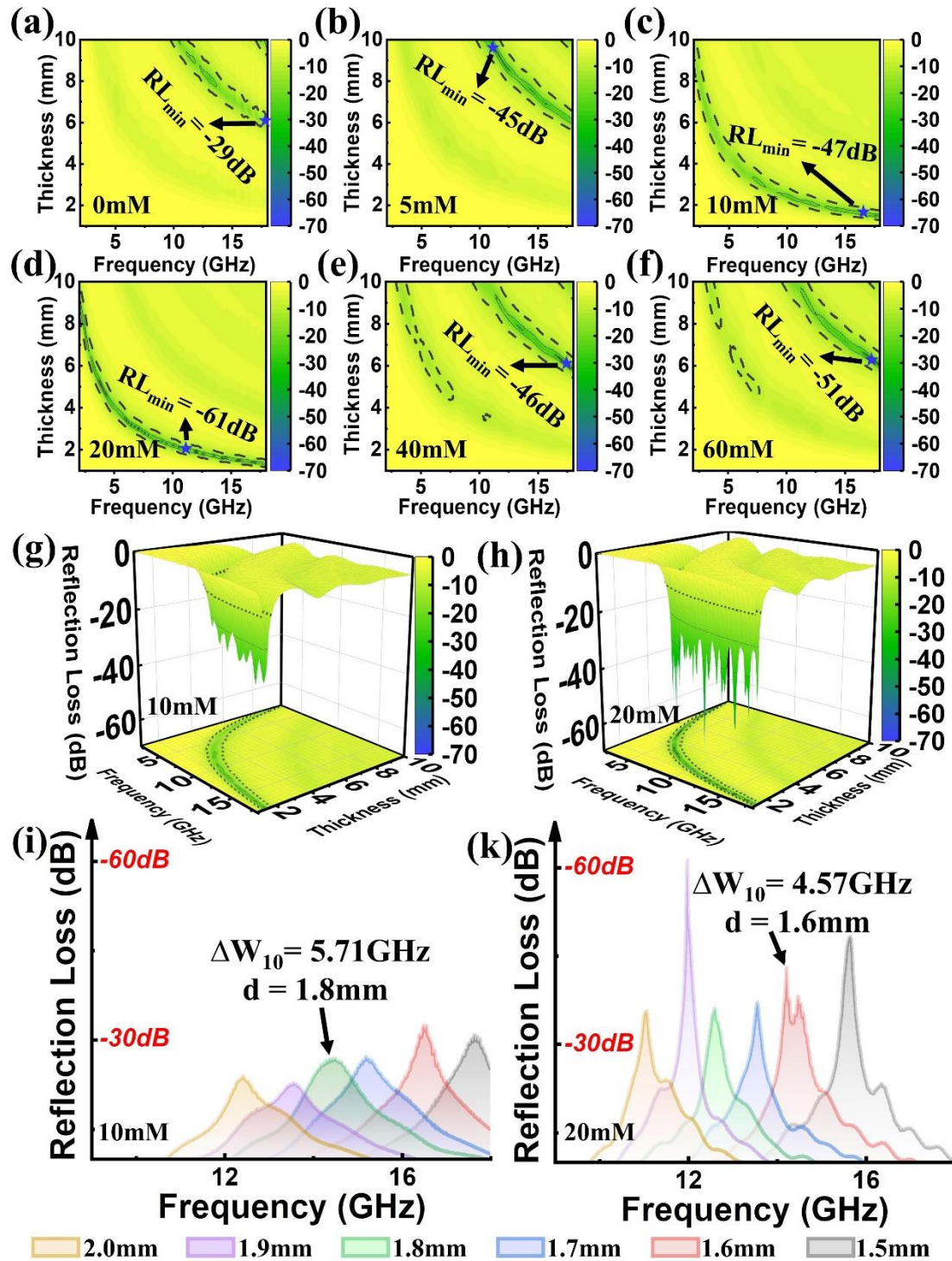
219 The influence of different shell thicknesses of H-SiC@MnO₂ composites on the
 220 EMW properties is determined based on the transmission-line theory. For single layer
 221 absorber, the reflection loss (RL) can be calculated as follows [42, 43]:

$$222 \quad RL = 20 \log_{10} |(Z_{in} - Z_0) / (Z_{in} + Z_0)| \quad (1)$$

$$223 \quad Z_{in} = Z_0 \sqrt{\mu_r / \varepsilon_r} \tanh[j(2\pi f d / c) \sqrt{\mu_r \varepsilon_r}] \quad (2)$$

224 where $Z_0 = (\mu_0 / \varepsilon_0)^{1/2} = 120\pi = 377\Omega$, Z_0 means the free space impedance. Z_{in}
 225 represents the input impedance of the absorber. ε_r ($\varepsilon_r = \varepsilon' - i\varepsilon''$) and μ_r ($\mu_r = \mu' -$
 226 $i\mu''$) are complex permittivity and complex permeability, respectively. f is the
 227 frequency. d is the thickness of the absorber, and c is the propagation velocity of
 228 EMW in free space. The frequency bandwidth in which the RL is less than -10 dB is
 229 expressed by ΔW_{10} , corresponding to 90 % of the EMW energy absorbed. Thus, ΔW_{10}
 230 is used to measure the effective absorption bandwidth. Based on Equations (1) and
 231 (2), the RL of different concentrations of H-SiC@MnO₂ composites were calculated,
 232 and the results can be seen in Figure 5. For 0 mM (pure H-SiC) and 5 mM H-
 233 SiC@MnO₂ composites, the highest filling ratio of the tested ring is only 20 wt% due
 234 to the soft trait. For other H-SiC@MnO₂ composites (10 mM, 20 mM, 40 mM and 60

235 mM), the filling ratio is 35 wt%. The 3D colourmap surface plot with the projection
236 of the reflection loss of each sample is shown in Figure.S4. As seen in Figure 5 (a-f),
237 adding the MnO₂ shell can improve the absorbing strength and widen the absorption
238 bandwidth of the composites. The minimum RL (RL_{min}) of H-SiC@MnO₂-0, 5, 10,
239 20, 40 and 60 is -29, -45, -47, -61, -46, and -51 dB, respectively. The region within the
240 dotted line represents the effective absorption bandwidth of each composite in the
241 range of 1-10 mm. Among them, H-SiC@MnO₂-10 and H-SiC@MnO₂-20 can
242 achieve EMW absorbing properties in the entire frequency range, and the smaller the
243 frequency, the greater the thickness. The reason is that the wavelength increases as the
244 frequency decreases, leading to the longer distance the EMW travels in the same
245 period, which needs a higher thickness [44]. Figure 5 (g-k) clearly shows the better
246 EMW performance of H-SiC@MnO₂-10 and H-SiC@MnO₂-20 composites. The
247 ΔW_{10} of H-SiC@MnO₂-10 and H-SiC@MnO₂-20 were measured to be 5.71 GHz and
248 4.57 GHz at only 1.8 mm and 1.6 mm, respectively, which improved by 225 % and
249 180 % compared to the pure H-SiC (ΔW_{10} =2.54 GHz at 10 mm).



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Figure 5 (a-f) The 2D colour-filled contour plots of reflection loss of pure H-SiC (0mM) and different

252

concentrations of H-SiC@MnO₂ composites. (g-h) The 3D colourmap surface plot with the projection of reflection

253

loss of H-SiC@MnO₂-10 and H-SiC@MnO₂-20 composites. (i-k) The reflection loss curves of H-SiC@MnO₂-10

254

and H-SiC@MnO₂-20 composites at different thicknesses.

255

As thin interference-absorbing material, the absorption intensity and bandwidth

256

are controlled by electromagnetic parameters. SiC and MnO₂ are non-magnetic

257 materials; therefore, complex permeability and magnetic loss are not affected. The
258 complex permittivity was studied to explain the improved EMW performance of H-
259 SiC@MnO₂ composites and the curves are shown in Figure 6. With the increase of the
260 concentration of KMnO₄, the dielectric constant of H-SiC@MnO₂ composites
261 increased at first and then decreased, which is due to the changes in the scattering
262 effect of the hollow spheres in paraffin. If the concentration of KMnO₄ was above 40
263 mM, agglomeration would happen, leading to decreased scattering effect and lower
264 dielectric constant. When the concentration falls below 40 mM, the hollow spherical
265 composites have a good dispersity, and the scattering effect increases with the larger
266 spheres, resulting in a higher dielectric constant. The complex permittivity (ϵ' and ϵ'')
267 of each sample tends to increase with the decrease in frequency, which can result in a
268 widened frequency bandwidth. Furthermore, the faster the increase, the more
269 pronounced the broad bandwidth is [45, 46]. The results in Figure 6 (a) present the ϵ'
270 value increase from 5.5 at 0 mM to 17.1 at 20 mM; however, it then decreases to 6.2
271 at 60 mM with the rising thickness of the MnO₂ shell. H-SiC@MnO₂-20 shows the
272 fastest variation trend from 10.5 to 17.1, while H-SiC (0 mM) present the lowest
273 increase trend from 4.3 to 5.5. In addition, the ϵ' value and growth trend of H-
274 SiC@MnO₂-10 are very close to the H-SiC@MnO₂-20 product, which rises from 8.3
275 to 15.2. The ϵ'' value and dielectric loss tangent ($\tan \delta_\epsilon$) present the similar tendency
276 except for the composite at the concentration of 10 mM, which has the highest ϵ''
277 value of 7.2 and the biggest $\tan \delta_\epsilon$ value at 0.54. Meanwhile, as observed in Figure 6
278 (b), there are three peaks exist in ϵ'' curves, which represent multiple relaxation
279 processes due to the interface polarisation and dipole polarisation [47, 48]. The
280 numbers and positions are consistent well with the Cole-Cole curves in Figure.S5, and
281 each semicircle presents the Debye relaxation process. There are two semicircles for

282 pure hollow SiC (0 mM), which may be due to dipole polarisation coming from
 283 defects. While for H-SiC@MnO₂ composites, three semicircles can be observed,
 284 demonstrating the enhancements of the Debye relaxation process, which are put down
 285 to the interfacial polarisation between SiC and MnO₂ [49, 50]. Furthermore, a higher
 286 ϵ'' and $\tan \delta_\epsilon$ value equates to a more significant dielectric loss capability. Thus, H-
 287 SiC@MnO₂-10 possesses the broadest absorbing bandwidth among the composites,
 288 which aligns well with the result in Figure 5.

289 The absorption intensity of single-layer dielectric material is achieved by
 290 improving the impedance matching degree, and the complete matching means the
 291 total reflection is 0 ($Z_{in} = Z_0$). For pure dielectric material ($\mu'=1, \mu''=0$), the
 292 relationship between ϵ' , ϵ'' and d/λ_0 can be expressed as follows [44]:

$$293 \quad \tan h[j(2\pi d / \lambda_0)\sqrt{\epsilon' - \epsilon''}] = \sqrt{\epsilon' - \epsilon''} \quad (3)$$

294 where λ_0 ($\lambda_0 = c / f$) is the wavelength of EMW in free space. Through Equation
 295 (3), the matching value of ϵ' , ϵ'' and d/λ_0 can be calculated. In theory, if the $\epsilon' - \epsilon''$
 296 (Cole-Cole) curves of an absorber intersect with the calculated matching curve, the
 297 electromagnetic parameters at the intersection are entirely matched, and the EMW
 298 absorbing strength is the most significant [51]. If it is outside the matching curve, the
 299 electromagnetic parameters do not match, and a minor interference-absorbing peak
 300 can be generated. To put it simply, the closer the Cole-Cole plot is to the matching
 301 curve, the better the matching condition is and, thus, the stronger the absorption peak.
 302 The matching line and Cole-Cole curves of each composite are shown in Figure 6 (d).
 303 Curves of H-SiC@MnO₂-10 and H-SiC@MnO₂-20 are close to the matching line,
 304 especially the curve of H-SiC@MnO₂-20; as can be seen, it generated a significant
 305 number of intersection points with the matching curve, which represents a lot of
 306 stronger EMW absorbing peaks. The curve of pure H-SiC (0 mM) is farthest from the

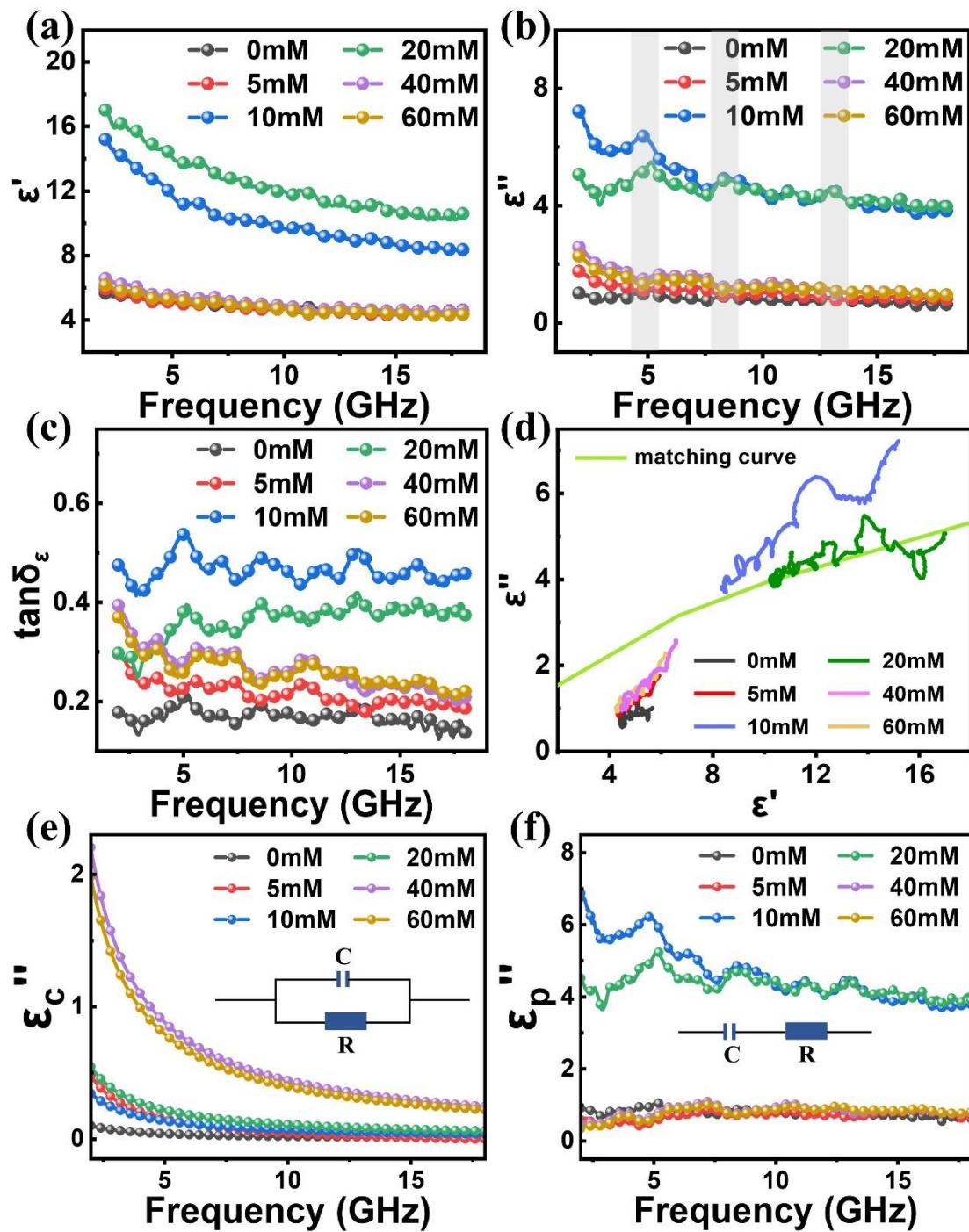
307 matching curve, and the EMW absorbing strength is weak, which is caused by the
308 lower ε'' value. Furthermore, when the real part of permittivity is low, the range of
309 the matching imaginary part is narrow. When the real part of permittivity is high, the
310 scope of the matching imaginary part is broad [44]. Therefore, the larger the complex
311 permittivity value, the easier it is to achieve matching conditions, resulting in better
312 EMW absorption. The improved EMW absorbing intensity of each composite can be
313 clearly seen in the 3D colour-mapping graphs (Figure.S4), which can be attributed to
314 the enhanced matching condition of ε' and ε'' value due to the introduction of the
315 MnO₂ shell. The particular hollow core-shell structure with different electronegativity
316 can induce interfacial polarisation and boost ε' and ε'' value, which favours the
317 impedance matching and broad absorption bandwidth of the composite absorber. In
318 addition, the hollow cavity not only facilitates the multireflection and multi-scattering
319 in the core but also brings down the density, which endows the composite with an
320 immense amount of filling at the same weight ratio.

321 According to Debye's theory, Equation (4) is used to quantitatively analyse the
322 proportion of polarisation and conductive loss.

$$323 \quad \varepsilon'' = \varepsilon_p'' + \varepsilon_c'' = (\varepsilon_s - \varepsilon_\infty)[(\omega\tau)/(1 + \omega^2\tau^2)] + \sigma/\omega\varepsilon_0 \quad (4)$$

324 where ε_s is the static dielectric constant, ε_∞ is the optical dielectric constant, τ is the
325 relaxation time, $\varepsilon_0 = 10^7/4\pi c^2 \approx 8.854 \times 10^{-12} F/m$, ε_0 means the static dielectric
326 constant of free space, σ is conductivity, ε_p'' and ε_c'' represent polarisation loss and
327 conductive loss, respectively. An equivalent circuit can describe the interaction
328 between EMW and absorber, and the ε_c'' can be equivalent to a parallel circuit with a
329 shunt effect, the ε_p'' can be equivalent to a series circuit with a divider effect. The
330 fitted results are displayed in Figure 6 (e-f). The ε_c'' increases with the increase
331 contents of MnO₂, which is mainly due to the micro-current generated by nanoflake

332 MnO₂ (Figure 6 e). For ϵ_p'' curves, when the concentration is lower than 40 mM, the
333 ϵ_p'' value raises with the increase of MnO₂ content which is due to the increase in
334 interface polarisation between MnO₂ and SiC. When the content of MnO₂ further
335 increases, the shell is thicker, and the spheres adhere to each other; thus, the value
336 decreases obviously. Furthermore, when the concentration is lower than 40 mM, the
337 dielectric loss of the H-SiC@MnO₂ composites is dominated by polarisation loss. In
338 contrast, the conductive loss is higher than the polarisation loss when the
339 concentration further increases.



340

341 Figure 6 Complex permittivity of different concentrations of H-SiC@MnO₂ composites. (a) real permittivity,

342 (b) imaginary permittivity, (c) dielectric loss tangent, (d) Cole-Cole curves, (g) contribution of conductive loss, (h)

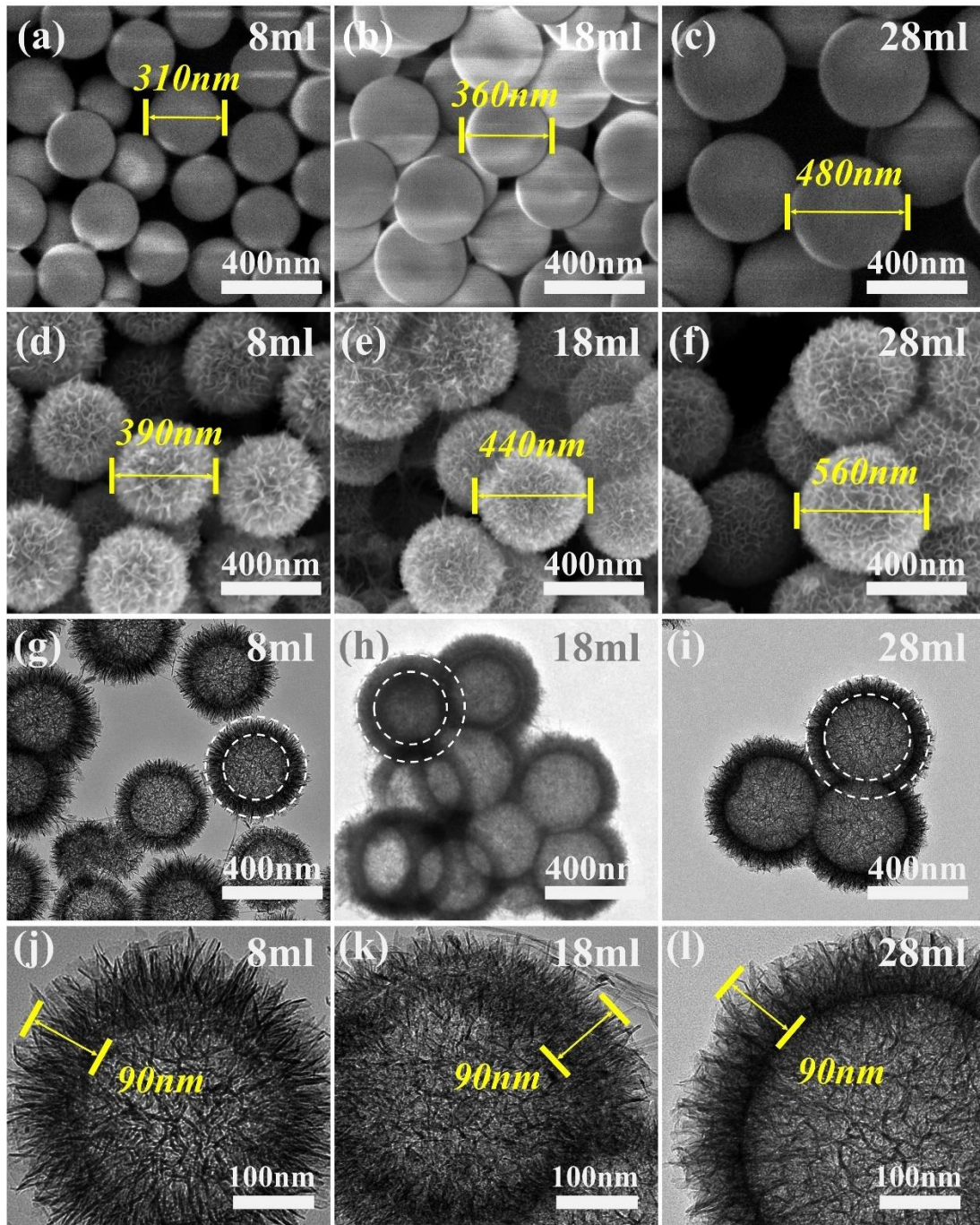
343 contribution of polarisation loss.

344 To further study the contribution of hollow core size on EMW absorption

345 capabilities, a string of H-SiC@MnO₂ composites with a concentration of 10 mM

346 KMnO₄ were prepared with three different volumes (8 ml, 18 ml, and 28 ml) of

347 TEOS. Figure 7 (a-c) shows that the diameter of SiO₂ templates is 310, 360, and 480
348 nm when the volume of TEOS changes from 8 ml to 18 ml and further to 28 ml. The
349 silicon sources increase with the high volume of TEOS. Sufficient raw material
350 continuously supplies the condensation nuclei and grows up to larger SiO₂ templates.
351 Then, after a series of reactions and treatment, H-SiC@MnO₂ composites with 390,
352 440 and 560 nm were synthesised under the different sizes of templates (Figure 7 d-f).
353 The diameters of the templates and final composites are aligned well with the
354 statistical size distribution as shown in Figure.S6, where the average diameter of SiO₂
355 templates is 309.1, 359.7 and 481.4 nm, the average particle size of H-SiC@MnO₂
356 composites are 390.1, 440.9 and 561.2 nm, respectively. As seen in Figure 7 (g-l), all
357 the products show a hollow structure and the same shell thickness of 90 nm.



358

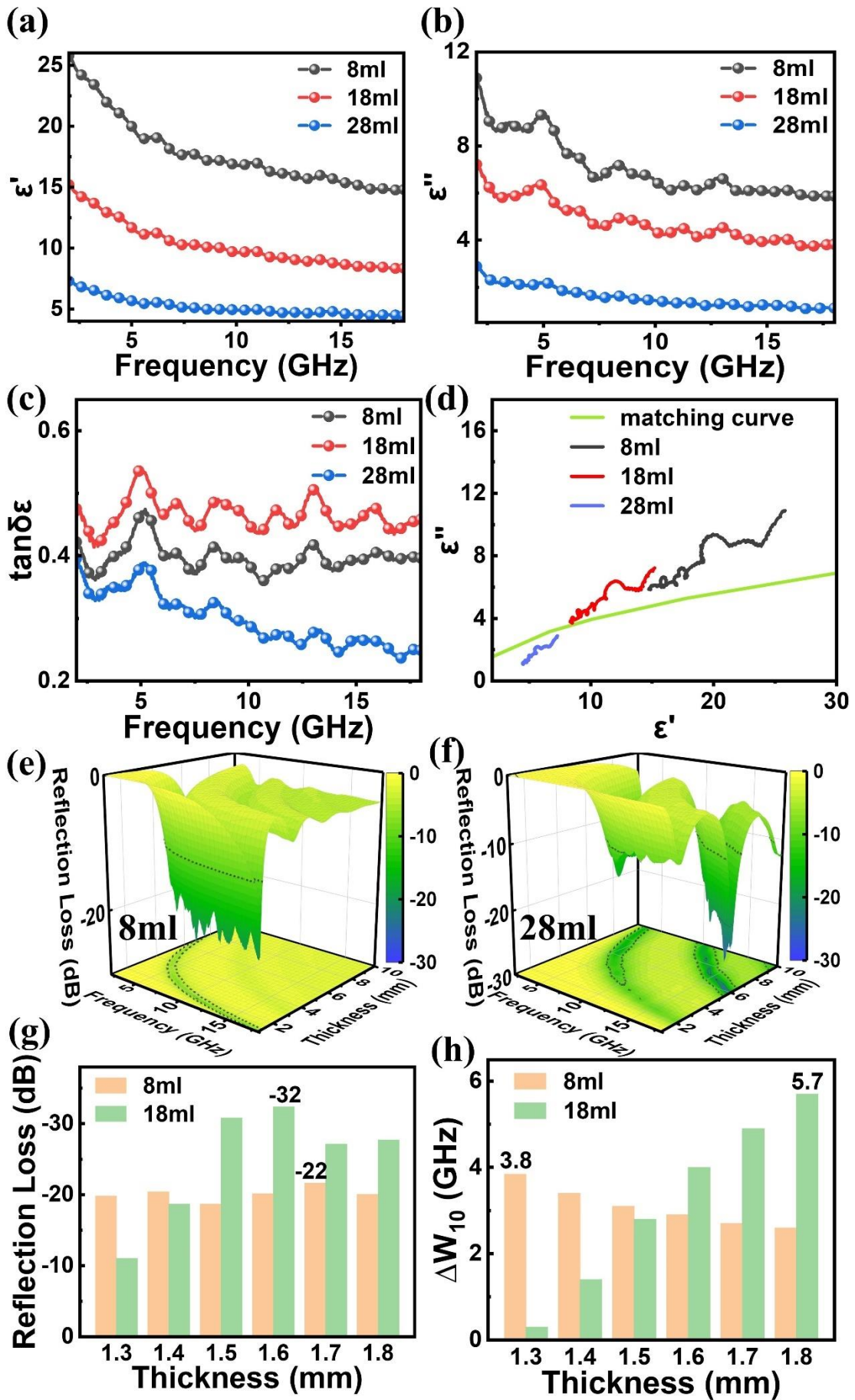
359 Figure 7 (a-c) FESEM images of different sizes of SiO₂ templates. (d-f) FESEM images of varying core sizes
 360 of H-SiC@MnO₂ composites. (g-l) TEM images of different core sizes of H-SiC@MnO₂ composites.

361 The complex permittivity closely relates to the size of the nanoparticles due to
 362 the mutual effect of free electrons. Since the theoretical formula for the size effect on
 363 the dielectric medium has not been established, the Euler equation, which describes
 364 the size effect on metal particles, is used to analyse the relationship between the
 365 complex permittivity and particle size. When the effective average free path of

366 electrons in the particles equals the particle size, the size effect on the complex
367 permittivity can be calculated as follow [35]:

$$368 \quad \varepsilon(\omega, r) = \varepsilon'(\omega) + i[\varepsilon''(\omega)] + (\omega_p^2 v_f) / (\omega^3 r) \quad (5)$$

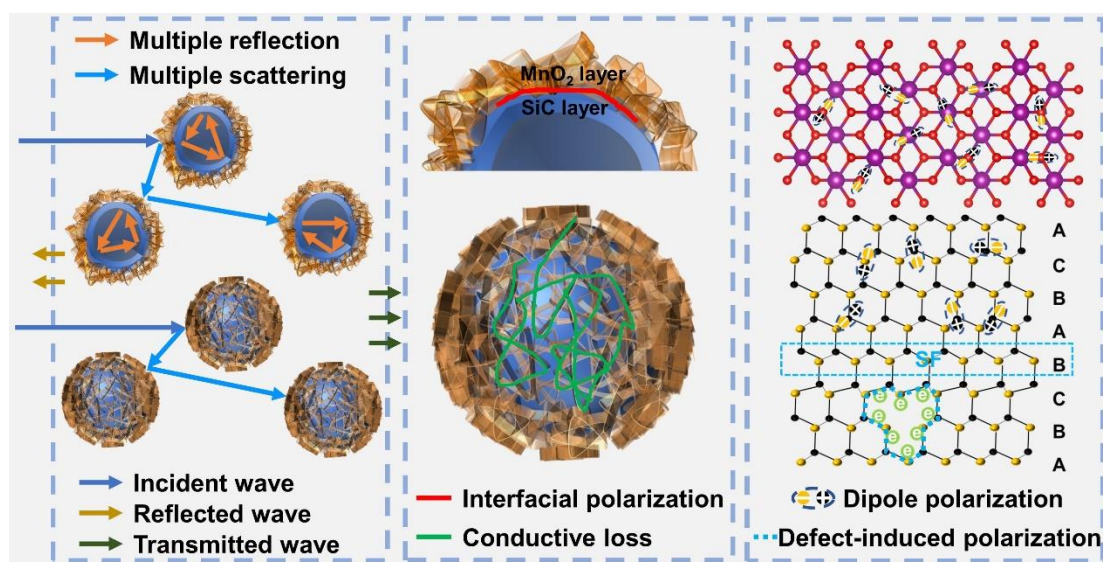
369 where ω_p is the plasma frequency of the bulk medium, ω is the electromagnetic wave
370 frequency, v_f is Femi velocity, and r is the radius of the particle. Equation (5) shows
371 that the imaginary permittivity is inversely proportional to the radius of the particles.
372 Therefore, the imaginary permittivity increases with the decrease in particle size. As
373 observed in Figure 8 (a-c), the particle size has a noticeable effect on the dielectric
374 property of H-SiC@MnO₂ composites. The ε' increase from 7.3 at 28 ml to 15.2 at 18
375 ml, then further rises to 25.7 at 8 ml as the core size decrease. The ε'' value increases
376 from 2.8 at 28 ml to 10.8 at 8 ml as the diameter falls, indicating an increase in the
377 loss capacity. The growth of complex permittivity is because the smaller the particle
378 size, the higher the ratio of atoms on the surface, which are more likely to polarise and
379 thus lead to the increase of permittivity. Even though the smallest H-SiC@MnO₂
380 composite at 8 ml possesses the highest complex permittivity, its EMW ability is not
381 the best (Figure 8 e-h) due to the poor impedance matching condition. As observed in
382 Figure 8 (d), the Cole-Cole curve of the H-SiC@MnO₂ composite at 18 ml is the
383 closest to the matching curve, which means the best matching condition between ε'
384 and ε'' . It is commonly accepted that a well-matched impedance matching condition is
385 a primary condition for designing high-performance EMW absorbers [43, 52].
386 Especially at a thickness of 1.3-1.8 mm, the ΔW_{10} and RL_{\min} of H-SiC@MnO₂
387 composite at 18 ml can reach 5.7 GHz and -32 dB. In comparison, the two value of H-
388 SiC@MnO₂ composite at 8 ml is only 3.2 GHz and -22 dB, respectively. The results
389 indicate that the H-SiC@MnO₂ composite at 18 ml has the best EMW absorbing
390 performance due to the suitable size.



392 Figure 8 (a-d) Complex permittivity of the different core sizes of H-SiC@MnO₂ composites. (a) real
393 permittivity, (b) imaginary permittivity, (c) dielectric loss tangent, (d) Cole-Cole curves. (e-f) The 3D colourmap
394 surface plot with the projection of reflection loss of H-SiC@MnO₂-10 at 8 ml and 28 ml. (g-h) The reflection loss
395 and effective absorption bandwidth (ΔW_{10}) of H-SiC@MnO₂-10 composites at different thicknesses at 8 ml and 18
396 ml.

397 The EMW absorbing mechanism of the H-SiC@MnO₂ composite is illustrated in
398 Figure 9. Firstly, when the EMW contacts the monodisperse core-shell spheres, the
399 particle can become a source to emit waves. Thus, the incident wave can be divided
400 into three parts: absorbed, propagating, and scattered. Notably, multiple reflections
401 occur when the EMW enters the sphere due to the hollow structure. The multiple
402 scattering, reflection and absorption of the composites significantly attenuate the
403 EMW energy to realise the high-performance absorber. Furthermore, the scattering
404 strength of the particle is closely related to the particle size. If the particle size is
405 appropriately large, the scattering is enhanced, and the attenuation of EMW energy is
406 improved. However, the complex permittivity falls with the growth of particle size.
407 Thus, the particle size to EMW absorbing ability is complex and contradictory. In this
408 work, the medium particle size absorbs the widest EMW. Also, the hollow structure
409 and nanoflower morphology improve the impedance matching condition, which
410 enables more EMW to enter the absorber to be consumed. Secondly, the MnO₂ shell
411 is uniformly coated on the spherical H-SiC, resulting in heterogeneous mass interfaces
412 where free electrons accumulate and generate strong interfacial polarisation to induce
413 dielectric loss. Meanwhile, according to equivalent circuit theory, there are three
414 kinds of conduction loss: (1) the nanosheet MnO₂ shell can form a sizeable
415 conductive network, promoting conductive loss. (2) the electron migration and
416 hopping between the MnO₂ shell and hollow SiC core can induce conductive loss. (3)
417 the conductive path between adjacent hollow spheres. In addition, SiC inside the
418 MnO₂ cover has a low bandgap, which can increase electron transport and further

419 boost the conductance loss of the composite. Finally, stacking faults (SF), defects and
 420 vacancies in SiC and MnO₂ change the dielectric constant, generate more dipole
 421 polarisation, and lead to dipole relaxation loss, making it easier to attenuate the EMW
 422 energy. Based on the discussion above, the medium-size H-SiC@MnO₂-10 with the
 423 concentration of TEOS at 18 ml has the optimum EMW absorbing performance.



424
 425 Figure 9 Schematic diagrams showing the EMW absorbing mechanisms.

426 4 Conclusions

427 In summary, a string of H-SiC@MnO₂ composites with size-tunable hollow
 428 spherical SiC core and thickness-tunable MnO₂ shell were successfully synthesised
 429 through carbon thermal reduction and hydrothermal reactions. The results show that
 430 the complex permittivity of the composites can be customised by adjusting the shell
 431 thickness and the core size. As expected, the hollow core-shell morphology of H-
 432 SiC@MnO₂ composites can improve the impedance matching condition and bring in
 433 multiple reflections of EMW. Moreover, the fully coated core-shell structure
 434 accelerates abundant heterogeneous interfaces, generating intense interfacial
 435 polarisation relaxation loss. Furthermore, the monodisperse and nanoscale characters
 436 of the composite enhance multiple scattering and reduce the ratio of reflection waves.
 437 When the concentration of KMnO₄ is 10 mM, and the volume of TEOS is 18 ml, the

438 prepared H-SiC@MnO₂ composite with 90 nm shell and 360 nm core has the highest
439 EMW absorbing ability, and the ΔW_{10} can reach 5.71 GHz at only 1.8 mm. This work
440 confirms that the H-SiC@MnO₂ composite is a very novel and potential EMW
441 absorbing material.

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448 **Declaration of competing interest**

449 The authors declare no conflict of interest.

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