Journal of Food Composition and Analysis Insights into the phytochemical composition of selected genotypes of organic kale (Brassica oleracea, L. var. acephala) --Manuscript Draft--

Manuscript Number:	JFCA-D-23-01396R2
Article Type:	Research Paper
Section/Category:	
Keywords:	Brassicaceae; biodiversity; organic cropping; phytochemical quality; sulphur compounds; carotenoids; phenolic compounds; prebiotic oligosaccharides
Corresponding Author:	Roberto Lo Scalzo Council for Research in Agriculture and Agricultural Economy Analysis Milano, ITALY
First Author:	Giulia Bianchi
Order of Authors:	Giulia Bianchi
	Valentina Picchi
	Aldo Tava
	Filippo Doria
	Peter Glen Walley
	Louisa Dever
	Maria Concetta Di Bella
	Donata Arena
	Hajer Ben Ammar
	Roberto Lo Scalzo
	Ferdinando Branca
Abstract:	Seven genotypes of kale (B. oleracea L. var. acephala), selected from a collection set up in the framework of the BRESOV H2020 Project, aimed at breeding under organic conditions, were analysed for the content in the characteristic phytochemicals of Brassica spp. The presence of prebiotic oligosaccharides was an important characteristic of this crop, with values ranging from 4.2 to 10.8 g/100 g d.w. The other detected soluble sugars were sucrose, glucose, and fructose, with glucose predominating (3.9-9.0 g/100 g d.w.).
	Sulphur compounds, such as sulphoxides, were detected in the form of pyruvic acid, their catabolic product (36-154 mg/100 g d.w.). In addition, the levels of main breakdown products of glucosinolates, such as sulforaphane and indole-3-carbinol, were found to be in average contents of 6.7 and 3.9 mol/g d.w., respectively. Finally, the concentrations of major phytochemicals in kale, such as polyphenols and carotenoids, reported considerable concentrations (around 1400 mg/100 g d.w. and 19 mg/100 g d.w, respectively), typical for this Brassica crop.
Suggested Reviewers:	Qiaomei Wang Zhejiang University College of Environmental and Resource Sciences qmwang@zju.edu.cn
	Dunja Samec Ruđer Bošković Institute dsamec@irb.hr
	Nicole L. Waterland West Virginia University nicole.waterland@mail.wvu.edu

Response to Reviewers:	
------------------------	--

Insights into the phytochemical composition of selected genotypes of organic kale (*Brassica* oleracea L. var. acephala)

Giulia Bianchi¹, Valentina Picchi¹, Aldo Tava², Filippo Doria³, Peter Glen Walley⁴, Louisa Dever⁴, Maria Concetta di Bella⁵, Donata Arena⁵, Hajer Ben Ammar⁵, Roberto Lo Scalzo^{1*}, Ferdinando Branca⁵

¹CREA Research Centre for Engineering and Agro-Food Processing, via G. Venezian 26, 20133 Milano, Italy;

²CREA Research Centre for Animal Production and Aquaculture, viale Piacenza 29, 26900 Lodi, Italy;

³Department of Chemistry, University of Pavia, viale Taramelli 10, 27100 Pavia, Italy;

⁴Department of Biochemistry, Cell, and Systems Biology, Institute of Systems, Molecular & Integrative Biology, University of Liverpool, Liverpool, L69 7ZB, United Kingdom;

⁵Department of Agriculture, Food and Environment (Di3A), University of Catania, via Valdisavoia 5, 95123 Catania, Italy.

*Corresponding Author; E-mail: roberto.loscalzo@crea.gov.it; full postal address: via G. Venezian 26, 20133 Milano, Italy; Phone +3902239557205

Type of paper: Original Research Paper

Funding information:

The present work was supported by the project BRESOV (Breeding for Resilient. Efficient and Sustainable Organic Vegetable production) funded by EU H2020 Program SFS-07-2017. Grant Agreement n. 774244.

Declarations of interest: none

Authors email addresses:

Giulia Bianchi: giulia.bianchi@crea.gov.it

Valentina Picchi: valentina.picchi@crea.gov.it

Aldo Tava: aldo.tava@crea.gov.it

Filippo Doria: filippo.doria@unipv.it Peter Glen Walley: Peter.Walley@liverpool.ac.uk Louisa Dever: L.Dever@liverpool.ac.uk Maria Concetta di Bella: maria.dibella@unict.it Donata Arena: donata.arena@phd.unict.it Hajer Ben Ammar: hejer.biologie@gmail.com Ferdinando Branca: fbranca@unict.it Dear Editor,

I and all co-Authors of the proposed work completely agree with your final comment. We understand the reasons on which the JFCA policy regarding the definition of antioxidant in food and its consequent potential bioactivity is based.

Consequently, we carefully changed the manuscript according to your indications, changing the term "antioxidant" to the concept of "compositional traits", better agreeing with the Policy of the Journal. This remembering, in agreement with the JFCA policy, that our concept of "antioxidant", when expressed, was related to the potential antioxidant content of the food matter, which obviously should be validated by further *in vivo* assays targeted to the antioxidant action in living cells or bodies when counteracting the oxidative stress by ROS and RNS.

Hoping that the changes made will meet the Editor's recommendations, we send to you the revised manuscript.

Roberto Lo Scalzo Corresponding Author on behalf of all co-Authors

Reply to the Reviewers

We are very grateful for the Comments of the Reviewers, which surely improve the scientific quality of the proposed work. To our knowledge, we have not further replies to the comments, so the reply has been included in the Cover Letter, being the comments made by the Editor.

Highlights

- The breeding hystory of germplasm strongly determines the kale biochemical profile
- Sulphur defence compounds appeared equally distributed in the studied kale genotypes
- Sucrose and total tocopherols were positively correlated with total carotenoids
- Organic kale crops showed adequate to elevated levels of bioactive phytochemicals

1 Insights into the phytochemical composition of selected genotypes of organic kale

2 (Brassica oleracea L. var. acephala)

3

4 Abstract

Seven genotypes of kale (B. oleracea L. var. acephala), selected from a collection set up in the 5 framework of the BRESOV H2020 Project, aimed at breeding under organic conditions, were 6 analysed for the content in the characteristic phytochemicals of Brassica spp. The presence of 7 8 prebiotic oligosaccharides was an important characteristic of this crop, with values ranging from 4.2 to 10.8 g/100 g d.w. The other detected soluble sugars were sucrose, glucose, and fructose, with 9 glucose predominating (3.9-9.0 g/100 g d.w.). Sulphur compounds, such as sulphoxides, were 10 11 detected in the form of pyruvic acid, their catabolic product (36-154 mg/100 g d.w.). In addition, the levels of main breakdown products of glucosinolates, such as sulforaphane and indole-3-carbinol, 12 were found to be in average contents of 6.7 and 3.9 µmol/g d.w., respectively. Finally, the 13 concentrations of major phytochemicals in kale, such as polyphenols and carotenoids, reported 14 considerable concentrations (around 1400 mg/100 g d.w. and 19 mg/100 g d.w, respectively), typical 15 for this Brassica crop. 16

Keywords: *Brassicaceae*, biodiversity, organic cropping, phytochemical quality, sulphur compounds,
 carotenoids, phenolic compounds, prebiotic oligosaccharides

Abbreviations: HAD, hydroxycinnamoyl derivatives; QD, quercetin derivatives; KD, kaempferol derivatives; HCG,
hydroxycinnamoyl gentiobiosides; GSH, glutathione; SRA, sulforaphane; I3C, indole-3-carbinol

21

The Brassicaceae family has a complex taxonomy and systematics: the number of genera and species 23 are often reviewed, due to the discovery of new species or to the application of novel genetic 24 techniques, which allow the distinction of different groups (Šamec et al., 2019; Maggioni et al., 2015; 25 Treccarichi et al., 2023). More specifically, the term kale refers to a group of plants belonging to the 26 species Brassica oleracea var. acephala, which comprises plants whose leaves do not form a head 27 (Arena et al., 2022). The acephala group includes the following morphotypes: kale (B. oleracea L. 28 var. acephala DC.), scotch kale (B. oleracea L. convar. acephala DC. Alef. var. sabellica L.), collard 29 (B. oleracea L. var. viridis L.), palm kale (B. oleracea L. convar. acephala DC. Alef. var. palmifolia 30 L.), marrow stem kale (B. oleracea L. convar. acephala DC. Alef. var. medullosa L.), thousand-head 31 kale (B. oleracea L. var. ramosa DC.) and Portuguese Tronchuda cabbage (B. oleracea L. var. costata 32 DC.) (Diederichsen 2001). In some cases, the acephala varieties include landraces with distinct 33 properties, such as the Laciniato kale, known as "Cavolo Nero Toscano", an appreciated Italian palm 34 kale. 35

Kale has a wide variety of culinary uses; it can be eaten raw in salads or cooked in soups or side 36 dishes. Many recipes belong to local traditions and others arise from the renewed attention which is 37 paid to this Brassica species. In recent years, indeed, there has been a growing interest in the 38 nutritional properties of kale plants, due to their complex profile of bioactive compounds (Di Bella et 39 al., 2021; Tribulato et al., 2020). Previous studies report the correlation between Brassicaceae 40 consumption and the protection against cancer and cardiovascular diseases (Sarıkamış et al, 2008; 41 Manchali et al., 2012). In fact, these vegetables are a rich source of ascorbic acid, carotenoids, sulphur 42 organic compounds and phenolics. Kale contains high level of ascorbic acid, in the range between 43 680 and 1120 mg/100 g dry weight (Korus, 2011; Vargas et al., 2022). In addition to ascorbic acid, 44 thiol-type compounds constitute a class of organic sulphur derivatives with sulfhydryl functional 45 groups (-SH), which play a crucial role as ROS (Reactive Oxygen Species) scavengers. Among them, 46

glutathione (GSH) is the prevalent metabolite, which is involved in the regeneration of reduced 47 ascorbic acid via the ascorbate-glutathione cycle. Being a strong ROS scavenger, glutathione 48 represents a phytochemical that is worth considering both for plant and human health. Compared to 49 other phytochemicals, glutathione and total thiols are often less studied and quantified in plants, and 50 little literature is available regarding their content in kale. In the work by Łata (2014), concerning the 51 variability in bioactive compounds in red and green-leafy kale in relation to soil type and N-level, the 52 author showed that GSH was the main thiol compound in kale (around 6 mg/100 g fresh weight), and 53 the share of GSH in the global GSH pool amounted to approximately 80%. In addition to GSH, L-54 cysteine was the other more abundant thiol compound (around 0.6 mg/100 fresh weight), followed 55 by γ -glutamylcysteine. 56

In Brassica oleracea, sulphur content is high, ranging from 0.6 to 1.0% of dry matter (Rosa and 57 Heaney, 1996). Sulphur is present as inorganic sulphate and involved in the formation of proteins, S-58 methyl-L-cysteine sulphoxide (SMCO) and glucosinolates (GSLs), a group of phytochemicals 59 characteristics of cruciferous plants. As a response to abiotic stress or biotic attacks, myrosinase 60 enzyme catalyzes the hydrolysis of GLSs, producing glucose and other breakdown products, 61 including isothiocyanates, thiocyanates and nitriles. These specialized metabolites are known for their 62 anti-inflammatory, anti-bacterial, and phase II detoxification effects (Steinbrecher et al., 2010). 63 Specifically, two of these metabolites, such as the sulforaphane (SRA), which derives from the 64 hydrolysis of glucoraphanin, and the indole-3-carbinol (I3C), which derives from the hydrolysis of 65 glucobrassicin, are widely recognized to have relevant healthy properties (Singh et al., 2021; 66 Sivakumar et al., 2007). 67

Among other leafy vegetables, kale stands out for the high content of carotenoids, such as lutein,
 violaxanthin and β-carotene.

Phenolic compounds are important specialized metabolites which contribute to kale health-promoting
 properties. They include glycosides of quercetin and kaempferol and derivatives of *p*-coumaric,

ferulic, sinapic and caffeic acid (Olsen et al., 2009), with caffeic and ferulic acid as predominant
 compounds (Ayaz et al., 2008).

The increasing concern about environmental issues and the demand for sustainable products with 74 75 high nutritional and sensory quality has led to the diffusion of organic cropping systems: specifically, previous research on *Brassica* crops indicate the positive influence of organic fertilizers on the 76 flavonoids and carotenoids content (Dos Reis et al., 2015). Moreover, some experimental results 77 indicate that the effect of the cropping system is cultivar dependent (Picchi et al., 2012; Naguib et al., 78 2012). The choice of plant material specifically developed for organic agriculture, which can 79 guarantee a good yield and high quality in diverse climate and stress conditions, is therefore a key 80 requirement to meet the demands of expanding organic farming systems. 81

In the framework of the BRESOV H2020 project "Shaping the future of organic breeding and 82 farming" (https://bresov.eu) a breeding program was established with the aim of developing new 83 high-quality Brassica spp. genotypes suitable for organic farming, through the exploitation of genetic 84 biodiversity. The present study has been focused on 7 selected kale genotypes belonging to the 85 collections of the University of Liverpool (UL) and the University of Catania (UNICT) that were 86 grown together under organic conditions in a controlled environment. The selection was based on the 87 genotypes showing the most relevant content in phytochemicals. Until now, few studies have reported 88 the concentration of other kale micronutrients, such as sulforaphane (SRA) and indole-3-carbinol 89 (I3C), organic volatiles and tocopherols (Sasaki et al., 2012; Wibowo et al., 2019; Korus 2020). 90

Resuming, the aim of the present study was a thorough characterization of phytochemical composition of these selected genotypes of kale specifically adapted for organic cultivation. The levels of soluble solids content, single sugars content, organic acids, ascorbic acid, thiols, glucosinolate breakdown products (I3C and SRA), volatile compounds, polyphenols, chlorophylls, tocopherols, and total carotenoids, were determined. Moreover, variability within each genotype was analysed, as well as the correlation among the different measured phytochemicals. 97

98 2. Materials and methods

99 2.1 Plant material

The assayed plant materials were selected among a germplasm collection grown in the experimental 100 field crops of University of Catania Di3A during season 2018-2019 in the framework of the BRESOV 101 H2020 Project (www.bresov.eu). The BRESOV collection is represented by hundreds of Brassica 102 103 genotypes, provided by several genebanks of different countries, such as Italy, United Kingdom, Portugal, Czech Republic, People's Republic of China. The organic growing system was carried out 104 105 according to the EC Rule 834/2007 and the Italian Ministerial Decree 220/1995. The experiment focused on seven Brassica oleracea var. acephala genotypes, selected with a screening approach 106 based on the relevant presence of phytochemicals by the total reducing power measured through the 107 Folin-Ciocalteu reaction and the DPPH scavenging capacity, carried out on the entire available 108 BRESOV collection of Brassica genotypes. These measurements, even if aspecific, were considered 109 a useful tool to distinguish, among a broad number of genotypes, the ones with higher potential 110 phytochemical content. In fact, a number of studies indicate high positive correlations between DPPH 111 scavenging capacity and ascorbic acid (Picchi et al., 2012 and 2020; Heimler et al., 2006) or phenolic 112 content (Jin et al., 2012). The genotypes (Table 1) were provided by the University of Liverpool (UK) 113 and the University of Catania, Di3A (Italy) and the seeds were sown in August 2018 cellular trays 114 using organic substrate (Terri® Bio) placed in a cold greenhouse in Catania. The plants were 115 transplanted in September 2018 in a potted experiment field at the University of Catania (37°31'10'' 116 N 15°04'18''E) under natural light (4.6 to 9.2 MJ m⁻² d⁻¹), and the mean temperature registered 117 during the growing year was $19.1^{\circ}C \pm 8.6^{\circ}C$. The plants were growing according to organic 118 agriculture methods utilizing amino acids and microorganisms for the nutritional support of the 119 plants, and Bacillus thuringiensis and pyrethrin based products for pest plant protection. Plant samples 120 were collected after the first year of growth from three different plants for each genotype. Single plant 121

samples were quickly flash-frozen at -50°C, freeze-dried, homogenized (22-74 mesh powder) and
stored at -20°C until analyses. The analyses have been performed in triplicate for each sample.

124 2.2 Soluble solids content and sugars composition

The soluble solid content (SSC) was measured following the rationale by Lamb (1972), with few 125 modifications, on an aqueous extract obtained by adding 0.5 g of freeze-dried kale powder to 5 mL 126 of distilled water, after 1,000g centrifugation. The SSC was expressed as °Bx on a dry weight (d.w.) 127 basis, using a digital bench refractometer (Bellingham-Stanley, mod. RFM 91). The same extract, 128 after centrifuging at 10,000g, was used for the determination of single soluble sugars, such as high 129 DP sugars (retention time 6.1 to 7.5 min), sucrose (8.5 min), glucose (10.4 min), and fructose (12.4 130 min). Soluble sugars have been analyzed following the validated guidelines by Li et al (2002), slightly 131 modified: they were analysed by HPLC with a refractive index as detector, using a Benson BP-100 132 300×7.8 carbohydrate column and quantification was made by comparison with external standards at 133 known concentrations. The mobile phase was H₂O at 80 °C, at a flow rate of 0.6 mL/min, and the 134 injection volume was 30 µL. The calibration was made by solutions of raffinose for high DP sugars, 135 sucrose, glucose, and fructose ranging from 0.05 to 0.33 mg/mL concentrations. The resulting average 136 conversion factors (concentration vs peak area) were 0.0021, 0.0023, 0.0025 and 0.0024 for raffinose, 137 sucrose, glucose, and fructose, respectively. 138

139 2.3 Analysis of organic acids

For organic acids determination, an HPLC method was developed following the validated guidelines by Jayaprakasha et al. (2003), with some modifications. Briefly, 3 mL of distilled water was added to 100 mg of freeze-dried powder. The mixture was vortexed at 1000 rpm for 5 minutes at room temperature, centrifuged (10000*g*, 5 min at 4 °C) and the supernatant was filtrated on a 0.45 μ m nylon filter.

The main identified organic acids (citric, malic, pyruvic and fumaric) were analysed by HPLC using 145 a Repromer H+ column, 300×8 mm, 9 µm (Dr. Maisch, Ammerbuch-Entringen, Germany) at 50 °C, 146 33 mM H₂SO₄ as mobile phase at a flow of 0.5 mL/min, with UV detection at 214 nm. Solutions of 147 commercial standards at known concentrations were used for calibration in concentration ranges of 148 0.05-1.50 mg/mL for citric, malic, and pyruvic and a range of 0.002-0.02 mg/mL for fumaric acid. 149 The resulting response factors were of 2277, 2452, 4118 and 156461, respectively. In these conditions 150 the retention times were: 10.9 min citric, 12.8 min malic, 14.6 min pyruvic, and 20.6 min fumaric. 151 Data were given as mg/100 g d.w.152

153 2.4 Ascorbic acid (AsA)

For this analysis, a previously validated method on Brassica samples was followed (Picchi et al., 154 2012): 50 mg of lyophilized powder was homogenized in the dark with 1.5 mL of 6% metaphosphoric 155 acid and centrifuged at 10000g for 5 min at 4 °C. The supernatant was filtered on 0.45 mm nylon 156 filters and immediately analysed. L-ascorbic acid (AsA) content was determined by HPLC using a 157 Supelco Ascentis Phenyl column (250 × 4 mm i.d.) at 44 °C, and 0.02 M orthophosphoric acid as 158 mobile phase. The flow rate was 0.6 mL/min and 20 µL samples were injected and monitored at 254 159 nm, and the retention time of AsA was observed at 5.9 minutes. The concentration was calculated by 160 interpolation of samples area data with a calibration curve made using solutions at known 161 concentrations of an AsA standard (purity HPLC \geq 98%, Sigma-Aldrich) (y = 36447x, r²= 0.9967, 162 range 2-100 μ g/ml), and was expressed as milligrams per 100 g/d.w. 163

164 *2.5 Total thiols*

The analysis of free soluble –SH groups was performed on 50 mg of lyophilised powder extracted with 1 mL of a 6% metaphosphoric acid solution, according to the already validated method used by Riener et al (2002), with few modifications. Briefly, an aliquot of 0.25 mL was treated with 0.5 mL of a solution composed of 160 mM 5-sulphosalicylic acid, 15 mM Na ascorbate and 2 mM Na₂– EDTA. Then, 1 mL of 0.8 M, pH 8.0 mono- and di-Na phosphate buffer and 0.25 mL of 10 mM DTNB (dithionitrobenzoic acid) were added. The absorbance of the resulting solution was read at 415 nm after 30 seconds of reaction in a 1 cm-path cuvette. The level of total thiols was calculated from a standard curve made with glutathione (GSH) (y = 11.681x, $r^2 = 0.9998$, range 10-50 µg/ml), and data were given as mg/100 g of glutathione (GSH) equivalents per 100 g of d.w.

174 2.6 HPLC analysis of sulforaphane (SRA) and indole-3-carbinol (I3C) contents

These two catabolites of glucoraphanin and glucobrassicin, respectively, were separated, identified, 175 and quantified by an SPE followed by an HPLC-DAD method, following the procedures proposed 176 by Pilipczuk et al., 2015 and 2017, with slight modifications: a kale powder aliquot of 50 mg was 177 incubated with 1 mL distilled water at room temperature for 30 minutes. Then, 1 mL of HPLC grade 178 EtOH 80% was added to the mixture, vortexed and centrifuged at 10000g. The clear supernatant was 179 placed on a C_{18} SPE cartridge (Alltech, MaxiClean, 300 mg), previously conditioned with 1×2 mL of 180 EtOH and 1×2 mL of water. After washing with 1×2 mL of water, the elution was performed with 181 3×0.5 mL of EtOH 96%, the eluate was filtered on 0.45 mm nylon filters and divided into two aliquots 182 for subsequent analyses of I3C and SRA, following previously validated methods: I3C has been 183 detected by fluorescence (Pilipczuk et al., 2015) and SRA was derivatized with N-acetyl cysteine in 184 sodium bicarbonate and detected at 272 nm (Pilipczuk et al., 2017). The HPLC separation was carried 185 out with a Pinnacle C₁₈ column (250 mm \times 4.6 mm i.d., 5 µm) under gradient elution of solvent A 186 (HCOOH 0.5%, MeOH 10% in water) and solvent B (CH₃CN/MeOH 9:1 HCOOH 0.5%) as follows: 187 from 90% (5 min isocratic condition) to 50% of solvent A in 20 min, then to 50% of solvent A in 25 188 min and to 90% of solvent A in 10 min. Compounds were eluted at 0.60 mL/min at 43°C and UV 189 detected at 272 for SRA and 280/360 nm ex./em. for I3C, with retention times at 8.3 and 11.4 minutes, 190 respectively. Peak identification was based on the comparison of retention times with those of pure 191 reference compounds treated the same as the samples. Quantitation of the detected compounds was 192 performed by an external standard method using available commercial standards (Sigma, purity 193

HPLC \geq 98%) of SRA with a concentration range of 0.1-0.8 µmol/mL and of I3C with a concentration range of 0.2-1.8 µmol/mL. The given concentration units were µmol/g d.w.

196 2.7 Volatile compounds profile

The volatile analysis from kale samples was performed by using a headspace solid-phase micro extraction (HS-SPME) coupled with GC-MS and GC-FID analyses. Each sample was composed by 200 mg freeze-dried *Brassica* kale sample added with 10 mL deionized water in a 20 mL glass vial, kept for 5 h at 60°C. Each vial was then closed with an aluminium/silicone-PTFE septum after the addition of 2 g of NaCl. Volatiles were identified using GC-MS and quantified using GC-FID.

Samples extraction was performed using a DVB-CAR-PDMS fiber, at 80°C for 20 min. Volatiles
desorption was obtained exposing the fiber in the GC injector at 200°C for 5 min.

GC-MS analyses were performed using a previously validated method (Picchi et al., 2012), with 204 modifications; they were carried out with an Agilent 6890 N GC connected to an Agilent 5973 mass 205 spectrometer and equipped with a DB-1 column (60 m \times 0.25 mm I.D., film thickness 0.25 µm), in 206 207 splitless mode, using He as carrier gas (flow 1 mL/min). The column temperature program was: 40 °C for 5 min, 3°C/min to 180 °C, 8 °C/min to 220 °C for 5 min (duration: 61,7 min). Injector and 208 detector temperatures were 200 and 230 °C, respectively; interconnecting line temperature, 200 °C. 209 The MS settings were as follows: filament voltage, 70 eV; scan range, 39-450 amu; scan speed, 1.4 210 scan/s. The compounds were identified comparing their mass spectra with those stored in Wiley 7n 211 library and analysing authentic standards. 212

The quantification GC-FID system comprised extraction, injection, and desorption from SPME fibers performed using a HT2800T autosampler (HTA S.r.l., Brescia, Italy) connected to the GC-FID system. The GC-FID analysis followed a procedure previously validated on other vegetables (Lo Scalzo et al., 2020) and was performed on 16 selected compounds using the same column and chromatographic conditions as GC-MS. Quantification was performed by the interpolation of calibration curves made with known concentrations of hexanal (slope equation y=0.0052 x, r^2 =0.90, LOQ 0.4 ng/g, LOD 0.1 ng/g, average ranges 1-100 ng/g, used for aldehydes and ketones) and allyl isothiocyanate (slope equation y=0.0055x, r^2 =0.98, LOQ 0.3 ng/g, LOD 0.1 ng/g, average ranges 0.5-100 ng/g, used for other compounds) and the values were expressed as ng/g d.w.

222 2.8 Chlorophylls and total carotenoids

For the analysis of chlorophylls and total carotenoids, a spectrophotometric method was used, following the instructions and equations previously performed and validated by Lichtenthaler and Wellburn (1983). An aliquot of around 50 mg sample exactly weighted was extracted with 1.5 mL of an EtOH/acetone 1:1 solution added with 0.1% BHT, to avoid autoxidation. After vortexing, sonication and centrifugation, the clear supernatant was diluted with acetone and analysed by spectrophotometer, set at the wavelengths of 470, 646 and 663 nm, requested by the equations from Lichtenthaler and Wellburn (1983). Results were given in mg/100 g d.w.

230 2.9 Tocopherols

The evaluation of tocopherols was performed by HPLC, with a fluorescence detection, following the rationale by Meckelmann et al. (2015).

The chromatographic system used for the separation, identification and quantification of kale tocopherols comprised a column C₃₀ YMC Europe S-3 μ m, 4.6 mm × 250 mm, flow 0.4 mL min⁻¹ at 40 °C, eluent A CH₃CN/MeOH 9:1, eluent B MTBE (methyl *tert*-butyl ether), with a gradient elution from 95% to 85% solvent A in ten minutes, then 85% solvent A for 10 minutes and return to the starting conditions in ten minutes, for a total run length of 30 minutes. The detection was spectrofluorimetric, setting the detector at 295 nm excitation and 325 nm emission.

In these conditions, the retention times were 13.4 min for δ -tocopherol, 14.7 min for γ -tocopherol and 16.4 min for α -tocopherol. The calibration was performed by standard solutions of δ -, γ - and α -tocopherols at known concentrations in an average range from 10 to 100 mg/mL, with resulting equations for sample data interpolation having slopes of 2×10^7 , 2×10^7 , 4×10^7 , respectively, and an average experimental point fitting of $r^2 = 0.999$.

244 2.10 Phenols analysis

Phenols were extracted starting from powdered plant material (20 mg), treated with 1 mL of EtOH/0.06 N HCl (1:1) and shaken at room temperature overnight. The samples were then centrifuged at 25,000g for 20 minutes, the supernatant was filtered on 0.2 mm nylon filters and stored at -80 °C until analyses.

The phenolic composition was evaluated by a double chromatographic approach, including the extract
 analysis by UPLC-ESI-MS and the subsequent compound quantification in an HPLC-DAD system.

For ESI/MS-MS, a Jasco UPLC system equipped with a binary pump system, photo diode array 251 detector (Jasco Chemstation ChromNAV) and coupled to a Thermo LTQ (linear ion trap mass 252 detector) with an electrospray ionization (ESI) source was used. All data were acquired and processed 253 using Thermo Xcalibur Qual Browser software. Chromatographic runs were carried out with an 254 Acquity UPLC BEH C₁₈ column (50 mm \times 2.1 mm i.d., 1.7 µm particles, 13 nm pore size) (Waters) 255 under a linear gradient of solvent A (H₂O/0.1% HCOOH) and solvent B (CH₃CN/0.1% HCOOH) as 256 follows: 0.0-5.0 min (5% B), 30.0 min (25% B), 40.0 min (70% B), 50.0 min (100% B). The flow 257 rate was 0.3 mL/min, the column temperature was set at 30 °C, injection volume 5.0 µL. The eluates 258 were spectrophotometrically checked at 9 different wavelengths (from 250 nm to 340 nm). For MS 259 detection, negative ESI was used as ionization mode. Capillary voltage, 3100 V; sheath gas (He), aux 260 gas (He), sweep gas (He) heated at 275 °C and introduced with a source heater temperature of 80 °C. 261 Full scan spectra were acquired over the range of 100-2000 m/z. Automated MS/MS was performed 262 by isolating the base peaks (molecular ions) using an isolation width of 2.0 m/z, normalized collision 263 energy of 25 V, threshold set at 500 and ion charge control on, with max acquire time set at 300 ms. 264 All the compounds were tentatively identified by a comparison of their retention times and mass 265 spectral data obtained in positive and negative ion mode with those of standard compounds or with 266

11

compounds previously reported in literature for *Brassica* spp. (Francisco et al., 2009; Lin et al., 2011;
Velasco et al., 2011; Yang et al., 2015; Picchi et al., 2020).

The phenolic composition of all extracts was also evaluated by HPLC-DAD using a PerkinElmer 269 270 (Norwalk, CT, USA) chromatograph equipped with a LC250 binary pump and DAD 235 detector. Chromatographic runs were carried out with a Discovery® HS C18 column (250 mm × 4.6 mm i.d., 271 5 µm) (Supelco) under gradient elution of solvent A (CH₃CN/0.05% CF₃COOH) and solvent B 272 (H₂O/1%MeOH/0.05%CF₃COOH) as follows: from 2% (10 min isocratic condition) to 15% of 273 solvent A in 30 min, then to 50% of solvent A in 50 min and to 90% of solvent A in 10 min. Twenty 274 µL of all extracts were injected. Compounds were eluted at 1.0 mL/min and detected by UV 275 monitoring at 330 nm, as illustrated in the example chromatogram (Figure 1). Peak identification was 276 based on LC-MS data and UV spectra were also recorded. Quantitation of the detected compounds 277 was performed at 330 nm by an external standard method using available commercial standards 278 (Sigma, purity HPLC \geq 98%) of chlorogenic acid (y = 63581x - 358152, r² = 0.9982) for compound 279 1; kaempferol-3-O-glucoside (y = 10701x + 78553, $r^2 = 0.9946$) and guercetin-3-O-glucoside (y = 10701x + 78553, $r^2 = 0.9946$) 280 42804 x + 5375, $r^2 = 0.9995$) for flavonol derivatives (peaks 2-36) and sinapoyl-4-O-glucoside (y = 281 43694x - 156367, $r^2 = 0.9935$) for hydroxycinnamoyl gentiobiosides (peaks 37-42). For calibration 282 curves, all standards were injected in triplicate in the range from 6 to 350 μ g/ml and a linear response 283 284 was observed. The given units for phenol quantification were mg/100 g d.w. As for the data analysis, the phenol composition was grouped into four classes of compounds, comprising the phenols with 285 similar structures, such as hydroxycinnamic esters, quercetin derivatives, kaempferol derivatives and 286 hydroxycinnamoyl gentiobiosides. 287

288 2.11 Data analysis

The present results are the mean \pm standard deviation (SD) of three plants per genotype. Each plant was analysed in triplicate. The percentage coefficient of variation (CV) among plants of the same genotype was evaluated as (SD/mean) *100. For each parameter, mean values for each genotype were compared using Tukey's test and significant differences were accepted at $P \le 0.05$. One-way ANOVA was carried out using Statgraphics v. 5.1 (Manugistics, US). Correlations among variables and principal component analysis (PCA) were computed with PAST v. 4.03 (Hammer et al., 2001).

296 **3. Results and Discussion**

3.1 SSC and single sugars content

298 The SSC is related to total soluble sugars content, and therefore represents an important sensory quality index. SSC significantly differed among genotypes and ranged from 26.2 ± 10.3 °Bx/d.w. (in 299 UNICT 4959) to 41.1 ± 3.2 °Bx/d.w. (in UL 5010) (Table 2). Previous studies have measured SSC 300 on fresh Galega kale plants, obtaining results of 11.2 and 13.2 °Bx referred to fresh weight (Armesto 301 et al., 2015; Martinez et al., 2010). Moreover, Akdas and Bakkalbasi (2017) indicate a value of 9.4 302 °Bx for kale grown in Turkey. In the work by Lefsrud et al (2008), the dry weight of kale widely 303 ranged from 15.0 to 21.4 % with high and low nitrogen input, respectively. Therefore, the here 304 presented data are within the ranges given by existing literature. 305

The presence of high degree of polymerization (DP) sugars (Table 2) followed that of SSC, with the 306 higher amount observed in UL genotypes. In fact, SSC was strongly correlated with high DP sugars 307 308 (r= 0.77, P= 0.000), and with fructose (r=0.49, P= 0.023). In the analytical conditions used in the present work, this fraction includes oligosaccharides composed by three or more sugar units. The 309 presence of these high potential prebiotic sugars was confirmed in previous studies (Jovanovic-310 Malinovska et al., 2014). In more recent studies on kale (Thavarajah et al., 2021) the characterization 311 of sugar profile highlighted the presence of the prebiotic oligosaccharides stachyose, raffinose, 312 verbascose, nystose and kestose in commercial kale genotypes adapted for organic cropping, ranging 313 from 5.7 to 8.7 g/100g of total prebiotic carbohydrates, in a similar range to the levels reported in the 314 present study (Table 2). 315

The sucrose content was generally lower compared to glucose and fructose, with slightly higher 316 amounts in UNICT genotypes, namely 375, 4946 and 4959 (0.5 g/100 g d.w.). Among single 317 monomeric sugars (Table 2), the prevalent compound was glucose, which ranged from 3.8 (in UL 318 5010) to 9.0 g/100 g d.w. (in UL 2066), without significant differences among the genotypes. The 319 highest (4.5 \pm 0.6 g/100 g d.w.) and the lowest (1.8 \pm 0.4 g/100 g d.w.) fructose levels were observed 320 in UL 2066 and UNICT 3332, respectively. Fructose and glucose amounts are overall higher than 321 those observed by Ayaz et al. (2006), which indicates fructose as the predominant sugar (2 g/100 g 322 d.w.) followed by glucose (1 g/100 g d.w.) and sucrose (0.9 g/100 g d.w.). By contrast, the present 323 kale genotypes seem to have slightly lower sugar content than those reported by Megiaz-Pérez et al. 324 325 (2020), and Jovanovic-Malinovska et al. (2014), with the predominance of glucose, in accordance with here presented data. 326

Plant development stage and environmental conditions affect both sugar profile and amount. Megias-Pérez et al. (2020) studied the relation of sugar composition to cold acclimation. Using HILIC-ESI-MS and GC-MS techniques, they observed that sucrose decreased significantly during kale plant development; however, at low temperature, fructose increased in kales grown at 2 °C, evidently increasing the potential sweetness.

332 *3.2 Organic acids*

Citric and malic acids were the two main organic acids detected in the seven kale genotypes (Table 333 3). They were found in amounts that are overall in accordance with previous studies (Ayaz et al., 334 2006; Nemzer at al., 2021). Citric acid concentrations (Table 3) ranged from 1962 to 3743 mg/100 g 335 d.w. in UL 2066 and UL 2075 respectively, without significant differences. Malic acid, instead, varied 336 from 454 mg/100 g d.w. in UNICT 4946 to 3175 mg/100 g d.w. in UL 5010 (Table 3). Interestingly, 337 the UL genotypes showed almost equal contents of these acids, while in UNICT genotypes citric acid 338 was largely prevalent, reflecting the acid pattern observed by Ayaz et al. (2006) in kales grown in 339 Turkey. Pyruvic acid content (Table 3) was similarly higher in UL genotypes and in UNICT 3332 340

(range 148-154 mg/100 g d.w.) than in the other UNICT genotypes (range 36-81 mg/100 g d.w.). 341 Pyruvic acid, together with ammonia and methanesulphenic acid, is a catabolic product of S-342 methylcysteine sulfoxide (MCSO), produced by cysteine sulphoxide lyases released after cellular 343 disruption (Edmands et al., 2013). In the past, the main use of kale was as forage, however breeding 344 activity has led to the decrease of this compound in modern kale varieties, due to the antinutrient 345 effects of MCSO, which is normally present in relevant concentrations (1-2 % on d.w.) and it is the 346 cause of hemolytic anemia of cattle (Edmands et al., 2013). This is particularly evident in UNICT 347 genotypes (Table 3), where pyruvic acid has an average content of 82 mg/100 g d.w., with respect to 348 UL genotypes (150 mg/100 g d.w.). The content in MCSO of these kale genotypes (average 13.2 349 µmol/g d.w.) is similar to that found by Marks et al. (1992) using other *Brassica* species, and lower 350 351 than that found in broccoli (around 50 µmol/g d.w.) (Traka et al., 2013). Interestingly, the average MCSO concentration is similar to the sum of the breakdown forms of glucosinolates, since SRA and 352 I3C averaged 6.7 and 3.9 µmol/g d.w., respectively, and it was higher than the levels of reducing 353 thiols, averaging at 3.3 µmol/g d.w; this may indicate that, in the studied genotypes, S-containing 354 defence compounds are almost equally distributed. 355

356 *3.3 Ascorbic acid and total thiols*

Ascorbic acid concentrations (Table 3) ranged from 209.4 (in UNICT 4959) to 243.8 mg/100 g d.w. 357 (in UL 2075). The values were overall slightly higher in UL genotypes, even though differences 358 among means were not significant. These amounts are in accordance with those reported by Hagen 359 et al. (2009) and Korus (2011) for dried kale samples. Total thiols content represents an important 360 part of plant sulphur metabolism, including total glutathione and non-protein –SH groups (Hawrylak 361 and Szymanka, 2004). In the present samples, thiol content differed significantly among genotypes 362 (Table 3). UNICT 3332 showed the highest level (119.9 mg/100 g d.w.) while the lowest (77.7 363 mg/100 g d.w.) was found in UL 2075. In her work, Łata (2014) found an average of 6 mg/100g fresh 364 weight of GSH in kale, while much higher concentrations of GSH, 6.9 µmol/g d.w. (211 mg/100 g 365

d.w.), were found in broccoli samples by Traka et al. (2013), and the present have shown intermediate 366 levels between the referenced ones. Ascorbic acid and glutathione are two bioactive compounds that 367 scavenge the ROS (mainly hydrogen peroxide) produced during the physiological processes of the 368 cells or in response to stress conditions. AsA and GSH cooperate in the AsA/GSH cycle to detoxify 369 the hydrogen peroxide (Noctor and Foyer, 1998). The total thiols were at some extent higher in 370 UNICT genotypes. The AsA/thiols ratio (Table 3) differed among genotypes, being significantly 371 higher in UL 2075 than other genotypes. AsA and thiols are both important parameters in determining 372 plant resistance to oxidative stress, such as environmental stress (Picchi et al., 2017). High GSH 373 regeneration is essential for AsA formation and AsA to thiols ratio is positively related to oxidative 374 375 stress tolerance (Fatima et al., 2019). In the present study, plants were not exposed to oxidative stress, 376 but the positive values of AsA to thiols ratio indicated a constitutive plant capacity to cope with possible oxidative stress. 377

378 *3.4 Sulforaphane and indole-3-carbinol*

The content of glucosinolates and their breakdown products in *Brassica* spp. is dependent on several environmental and developmental factors (Velasco et al., 2007). The I3C and SRA HPLC data (Figure 2) indicate that the kale genotypes used in this work are a potential good source of glucoraphanin, representing the native glucosinolate precursor of the sulphoraphane, well noted for its health promoting properties (Matusheski et al., 2004).

The average content of the breakdown product of glucoraphanin, such as sulforaphane (6.7 μ mol/g d.w.) almost double than the average content of indole-3-carbinol (I3C), the glucobrassicin catabolite (3.9 μ mol/g d.w.). These data are in a full accordance with the findings of Sasaki et al. (2012), where the concentration of glucosinolates precursors was considered. Moreover, similar amounts of glucobrassicins were found in kale by Son et al. (2021), while West et al. (2004) found relevant concentrations of glucoraphanin and glucobrassicin in kale seeds. Sulforaphane content of the kale genotypes used in this study ranged from 3.4 μ mol/g d.w.in UL 5010 to 12.4 μ mol/g d.w. in UNICT 391 375 (Figure 2). Overall, UL genotypes had lower sulforaphane levels and a higher variability among 392 the plants of the same genotype. In UL 5010 the CV was 78.9%. Unfortunately, relatively few 393 published data with respect to raw glucosinolates is available to compare the presence of sulforaphane 394 in kale samples. An amount ranging from 2 to 8 μ mol/g d.w was reported in previous studies (Li et 395 al., 2016; Sivakumar et al., 2007). Interestingly, kale can be a good source of sulforaphane, since 396 some samples have concentrations at around 12 μ mol/g d.w, similar to the commercialized 397 "Benefortè TM" broccoli (Traka et al., 2013).

The distribution of I3C concentration did not show clear differences between UL and UNICT genotypes (Figure 2). The higher average levels, found in UL 2075 and UNICT 4959 were 5.35 and 5.34 µmol/ g d.w., respectively, and the variability among plants was highest in UNICT 375, with a CV of 115.1%. Due to this high variability, no significant differences in the I3C content were found. Overall, the I3C concentrations were in full accordance with published data on commercial kale samples (Yu et al., 2018), while much lower concentrations were found in other works (Kapusta-Duch et al., 2016; Pilipczuk et al., 2015).

405 *3.5 Volatile compounds profile*

The 16 selected compounds identified and quantified in the SPME headspace of all genotypes are 406 listed in Table 4, with the data regarding the tentative identification listed in Table S2. Among them, 407 six compounds were aldehydes, with hexanal and (E)-2 hexenal as the prevalent compounds. These 408 aldehydes are the main products of the lipoxygenases (LOX) catabolism of linoleic and linolenic 409 acids, respectively (Cao et al., 2014). Kale is a good source of these essential fatty acids (Ayaz et al., 410 411 2006). (E)-2-hexenal concentration was higher in UL 5010 (100 ng/g d.w.), and in almost all other genotypes was around 60 ng/g d.w., with the exceptions of UNICT 3332 and UNICT 375 which had 412 lower levels (about 10 ng/g d.w.). The CV% among plants of the same genotype was higher in UNICT 413 3332 (34.7%). Hexanal content ranged from 1.0 in UNICT 3332 to 17.1 ng/g d.w. in UNICT 4946, 414 415 with a maximum CV related to the different plants in UNICT 375 (37.8%).

As regards sulphur volatile compounds, two main isothiocyanates were identified and quantified, 416 allyl- and 3-butenyl isothiocyanate (respectively a sinigrin and a gluconapin breakdown product). 417 Overall, higher concentrations of these compounds were found in UL compared to UNICT genotypes; 418 however, due to the large variability among plants, the differences among genotypes were not 419 significant. The 3-butenyl-isothiocyanate, characteristic of kale (Zeng et al, 2021; Oh and Cho, 2021) 420 was prevalent in HS-SPME, with the higher concentration in UL 2075 (539 ng/g d.w.) and lowest in 421 UNICT 3332 (0.4 ng/g d.w.). Allyl isothiocyanate was similarly higher in UL 2075 (141.3 ng/g d.w.) 422 and was present in concentrations around 1-2 ng/g in UNICT genotypes. For allyl isothiocyanate CV 423 ranged from 19.4% (BH 96) to 87.3% (UL 2075), while for 3-butenyl isothiocyanate it was not lower 424 425 than 52.6% (UL 2066), ranging in other genotypes from 55.8 to 80.3%.

Sulphides represent another class of sulphur compounds present in *Brassicaceae*, deriving from the MCSO catabolism (Edmands et al., 2013). The main compound of this class found in the studied kale accessions was dimethyl trisulfide, whose concentration varied from 23.9 ng/g d.w. in UL 5010 to 0.5 ng/g d.w. in UNICT 375, even though the CV% widely varied (0.21-62.8 in UNICT 375 and UL 2075, respectively).

Kales are naturally rich in carotenoids (Perry et al., 2009; Ashenafi et al., 2022). Apocarotenoids, 431 their respective catabolites, are another important class of volatile compounds, to which 6-methyl-5-432 hepten-2-one, geranyl acetone and β -ionone belong. 6-Methyl-5-hepten-2-one and geranyl acetone 433 are products of lycopene cleavage: the 6-methyl-5-hepten-2-one concentration varied from 0.6 in UL 434 2075 to 6.8 ng/g d.w. in UNICT 3332, and geranyl acetone amounts showed a similar distribution, 435 ranging from 0.1 ng/g d.w. in UNICT 3332 to 1.9 in UL 2075 (Table 4). UNICT 3332 and UNICT 436 375 had the lowest β -ionone amounts (0.3 ng/g d.w.), while UL 2075 had the highest (4.1 ng/g d.w.) 437 438 (Table 4).

Few literature data are available on kale volatile profiles, especially from a quantitative point of view.
Previous studies deal with changes in volatile profiles induced by processing (Wibowo et al., 2019;

Oh and Cho, 2021) or pest attack (Fernandes et al., 2010). Among the compounds found in this work,
(*E*)-2-hexenal, allyl isothiocyanate, dimethyl trisulphide and phenylacetaldehyde were identified by
Wibowo et al. (2019) and Oh and Cho (2021) as odour-active compounds in processed kale, while
some compounds found by Fernandes et al. (2010), such as (*E*)-2-hexenal, allyl isothiocyanate,
dimethyl trisulphide, and 6-methyl-5-hepten-2-one, were consistent with the volatile profile here
studied.

447 3.6 Chlorophylls (Chl) and total carotenoids

As reported in Table 5, the genotype UL 5010 distinguished by the highest concentration in 448 chlorophylls, both Chl a and Chl b at 199.8 and 71.1 mg/100 g d.w. respectively, followed by UL 449 2066 with 167.4 and 59.2 mg/100 g d.w. UNICT 3332 and UNICT 4946 showed lower amounts, 450 around 85 mg/100 g d.w. for Chl a and 25 mg/100 g d.w. for Chl b (Table 5). The differences in Chl 451 a and Chl b amounts observed among genotypes were significant. The detected levels are lower than 452 those reported in other studies (Kopsell et al., 2007; Akdaş and Bakkalbasi, 2017; Ljubej et al., 2021). 453 Previous research by Kopsell et al. (2007) indicate genotype as an important variation factor for 454 carotenoids concentration in kale; however, across the board, kale has been recognised as an 455 outstanding source for these secondary metabolites (Perry et al., 2009). In the analysed genotypes 456 some significant differences were found in total carotenoid content, which showed the same 457 distribution as chlorophylls, as evidenced by the positive correlation observed between both Chl a 458 (r=0.95, P=0.000) and Chl b (r=0.88, P=0.000). The carotenoid levels (Table 5), ranging from 11.9 459 in UNICT 3332 to 27.9 in UL 5010, were higher than those measured in other studies (Kopsell et al., 460 2007; Akdaş and Bakkalbasi, 2017). Ferioli et al. (2013) carried out a cross-country study on 461 phytochemical levels in local kale populations; compared to that study, the genotypes analysed in this 462 work ranked lower for chlorophylls and higher for carotenoids. 463

464 *3.7 Tocopherols*

Kale is considered a rich source of essential vitamins. The present work included an analysis of 465 tocopherols (Table 6). In all genotypes, α -tocopherol was the prevalent form. Total average 466 tocopherols $(\alpha+\delta+\gamma)$ ranged from 10.5 (in UNICT 3332) to 22.8 mg/100 g d.w (in UNICT 375). 467 Except for UNICT 3332, which possessed the lowest amount of each tocopherol form, overall higher 468 levels of α - and total tocopherols were found in UNICT genotypes, while γ - and δ - amounts were 469 higher in UL genotypes. UNICT 375 was higher in α -tocopherol (22.0 mg/100 g d.w.), while the 470 higher amounts of γ - and δ - were found in UL 2075, 1.7 and 0.9 mg/100 g d.w. respectively. High 471 variability among plants of the same genotype, especially in genotypes UL 5010 and UL 2066 was 472 observed, as shown by high values of standard deviations (Table 6). The observed distribution and 473 amounts are in accordance with those reported by Korus (2020, 2022). Total tocopherols content 474 significantly correlated with sucrose (r = 0.45, P = 0.039) and total carotenoids (r = 0.47, P = 0.033). 475

476 *3.8 Phenolic compounds*

The UPLC-ESI-MS analysis led to the identification of 43 phenolic compounds (Table 7), which 477 have been grouped into four main classes (Table S1): hydroxycinnamic acid derivatives (HAD) 478 flavonols, distinguished in quercetin (QD) and kaempferol derivatives (KD), and, finally, 479 480 hydroxycinnamoyl gentiobiosides (HCG), as shown in Figure 3, excluding the isorhamnetins derivatives, present in trace amounts (Table S1). The composition of the phenolic profile did not show 481 great differences among the samples, and flavonols were the prevalent compounds in all genotypes. 482 The total phenols content, instead, significantly differed among genotypes, except for the kaempferol 483 derivatives group (Figure 3). However, wide variability among plants was observed, especially in the 484 UL genotypes (Table S1). Overall, UNICT genotypes had higher phenol levels. Total phenols content 485 486 ranged from 763.4 mg/100 g d.w. in UL 2075 to 1920.2 mg/100 g d.w. in UNICT 3332. As regards the single compounds, a higher content of neo-chlorogenic acid was found in UNICT 3332 (130.3 487 mg/100 g d.w.). K-3-O-caffeoyldiglucoside-7-O-glucoside (higher content in UNICT 4946, 132.0-488 166.5 mg/100 g d.w.), K-3-O-sinapoyldiglucoside-7-O-glucoside (higher content in UNICT 375, 489

113.3-153.0 mg/100 g d.w.) and K-3-O-sinapoyldiglucoside-7-O-diglucoside (higher content in 490 UNICT 4959, 63.8- 189.2 mg/100 g) were in most genotypes the prevalent phenols. The quercetin 491 sinapoyl glucosides (compounds nr. 32 and 33, Figure 1, Table 7) were found in high amounts (82.0-492 197.1 mg/100 g d.w.) in UNICT 4946. Conversely, the lowest amounts of 32 and 33 were in UL2075 493 (19.2-126.9 mg/100 g d.w.) UL 2075 (Table S1). Among gentiobiosides, the most abundant 494 compounds were compounds nr. 39 and 42: UNICT 4959 was characterized by a higher amount of 495 39 (55.3-110.4 mg/100 g d.w.), while UNICT 3332 possessed a notable content of 42 (75.2-137.4 496 mg/100 g d.w). 497

Phenolic compound profiles agreed with previously reported data (Olsen et al., 2009; Hagen et al., 498 2009; Ferioli et al., 2013); the sum of all quercetin and kaempferol derivatives indicated a very high 499 flavonol content (above 1000 mg/100 g. d.w., except for UL 2075 and UL 2066), higher than those 500 reported by Hagen and colleagues (2009). Moreover, total phenolic content was in accordance with 501 the ranges found by Ferioli et al (2013). Interestingly, the quali-quantitative comparison with data 502 reported by Soengas et al. (2012) reported a perfect match both with regard to the presence of different 503 504 phenol classes and to their concentration. It could be resumed that kale is an interesting source of soluble phenolic phytochemicals, mainly ascribed to the class of flavonols. The literature data 505 describe higher amounts of total flavonoids in collard kale than broccoli, 20-50 mg/100g fw 506 (Radošević et al., 2017): these values resulted much higher in the present study, ranging from around 507 700 to 1500 mg/100g dw for the sum of kaempferol and quercetin derivatives (Figure 3), meaning 508 104-224 mg/100g fw, if assumed a presence of dry matter of 15 % (Lefsrud et al., 2008). 509

510 3.9 Principal Component Analysis (PCA)

To summarize and highlight the main differences among genotypes, PCA was performed, including all genotypes and the average values of each significantly different biochemical parameter. Five PCs were extracted, together accounting for 98.5% of the variance. Figure 4 shows a biplot of the first two components which explain 74.5% of the variance. As regards the scores, PC1 divides UL genotypes, with positive values, from UNICT genotypes, with negative values. Loadings for volatile apocarotenoids, isothiocyanates, hexanal and delta-tocopherols have negative PC1 and PC2 values and are associated with UL 2075, while other UL genotypes have positive PC2 values and are associated with organic acids, SSC, fructose, high DP sugars, dimethyl trisulfide, hexanal and chlorophylls. UNICT 3332 has a positive PC2 value and is associated with phenol compounds and total thiols, while other UNICT genotypes have negative PC2 values and are associated with SRA and I3C.

522

523 **4. Conclusions**

The genotypes analysed in this work are part of the BRESOV Brassica core collection; however, they 524 originate from two collections selected in different environments (Liverpool, UL, and Catania, 525 UNICT). The selected genotypes were cultivated under organic conditions to highlight differences in 526 both morphotype and genetic profile. As part of this, the aim of the present work was to select suitable 527 accessions that can be used directly for organic growing, and to contribute towards organic breeding 528 programmes for enhanced nutritional content. Interestingly, it has been determined that the two 529 groups of genotypes have different biochemical characteristics which are maintained when grown in 530 the same controlled environment. This was surprising since the major phytochemicals studied are 531 secondary metabolites, which can be largely influenced by environmental conditions. This study 532 shows that the breeding history of germplasm can strongly determine their biochemical profile. 533 Moreover, this work also highlights that kale crops with adequate to elevated levels of phytochemicals 534 potentially relevant for human health can be produced under organic growing conditions. 535

The presence of considerable concentrations of potential biologically active sulphur compounds, mainly in the forms of sulphoxides, reduced thiols and isothiocyanates, as well as the relevant concentration of ascorbic acid and phenols, makes these genotypes of kale very interesting material with appropriate nutritional characteristics related to *Brassica* spp.

540 **Funding information:**

541 The present work was supported by the project BRESOV (Breeding for Resilient. Efficient and

- 542 Sustainable Organic Vegetable production) funded by EU H2020 Program SFS-07-2017. Grant
- 543 Agreement n. 774244.

544

References

Akdaş, Z. Z., Bakkalbaşı, E., 2017. Influence of different cooking methods on color, bioactive compounds, and antioxidant activity of kale. Int. J. Food Prop. 20(4), 877-887. https://doi.org/10.1080/10942912.2016.1188308.

Arena, D., Treccarichi, S., Di Bella, M.C., Achkar, N., Ben Ammar, H., Picchi, V., Lo Scalzo, R., Amari, M. and Branca, F., 2022. Evaluation of *Brassica oleracea* L. crops and wild relatives for bio-morphometric and biochemical characteristics. Acta Hortic. 1355, *71-80*. DOI: 10.17660/ActaHortic.2022.1355.10

Armesto, J., Carballo, J., Martínez, S., 2015. Physicochemical and Phytochemical Properties of Two Phenotypes of Galega Kale (*Brassica oleracea* L. var. *acephala* cv. Galega). J. Food Biochem. 39, 439–448. https://doi.org/10.1111/jfbc.12151.

Ashenafi, E. L., Nyman, M. C., Holley, J. M., Mattson, N. S., Rangarajan, A., 2022. Phenotypic plasticity and nutritional quality of three kale cultivars (*Brassica oleracea* L. var. *acephala*) under field, greenhouse, and growth chamber environments. Environ. Exp. Bot. 199, 104895. https://doi.org/10.1016/j.envexpbot.2022.104895.

Ayaz, F. A., Glew, R. H., Millson, M., Huang, H. S., Chuang, L. T., Sanz, C., Hayırlıoglu-Ayaz, S., 2006. Nutrient contents of kale (*Brassica oleracea* L. var. *acephala* DC.). Food chem. 96(4), 572-579. https://doi.org/10.1016/j.foodchem.2005.03.011.

Ayaz, F. A., Hayırlıoglu-Ayaz, S., Alpay-Karaoglu, S., Grúz, J., Valentová, K., Ulrichová, J., Strnad, M., 2008. Phenolic acid contents of kale (*Brassica oleracea* L. var. *acephala* DC.) extracts and their antioxidant and antibacterial activities. Food Chem. 107(1), 19-25. https://doi.org/10.1016/j.foodchem.2007.07.003.

Cao, J., Deng, L., Zhu, X. M., Fan, Y., Hu, J. N., Li, J., Deng, Z. Y., 2014. Novel approach to evaluate the oxidation state of vegetable oils using characteristic oxidation indicators. J. Agric. Food Chem. 62(52), *12545-12552*. https://doi.org/10.1021/jf5047656.

Di Bella, M. C., Melilli, M. G., Treccarichi, S., Tribulato, A., Arena, D., Ruffino, A., Argento, S., Branca, F., 2021. Influence of irrigation regime on productive and qualitative traits of kale (*Brassica oleracea* var. *acephala* DC) under organic farming system. In *III International Organic Fruit Symposium and I International Organic Vegetable Symposium 1354* (pp. 301-308). 10.17660/ActaHortic.2022.1354.39

Diederichsen, A. "Cruciferae: Brassica". In: Mansfeld's encyclopedia of agricultural and horticultural crops. Ed. P. Hanelt, Institute of Plant Genetics and Crop Plant Research (Berlin:Springer) (2001). 1435-1446.

Dos Reis, L. C. R., de Oliveira, V. R., Hagen, M. E. K., Jablonski, A., Flôres, S. H., de Oliveira Rios, A., 2015. Effect of cooking on the concentration of bioactive compounds in broccoli (*Brassica oleracea* var. Avenger) and cauliflower (*Brassica oleracea* var. Alphina F1) grown in an organic system. Food Chem. 172, 770-777. https://doi.org/10.1016/j.foodchem.2014.09.124.

Edmands, W. M., Gooderham, N. J., Holmes, E., Mitchell, S. C., 2013. S-Methyl-L-cysteine sulphoxide: the Cinderella phytochemical? Toxicol. Res. 2(1), 11-22. https://doi.org/10.1039/c2tx20030a

Fatima, A., Singh, A. A., Mukherjee, A., Agrawal, M., Agrawal, S. B., 2019. Ascorbic acid and thiols as potential biomarkers of ozone tolerance in tropical wheat cultivars. Ecotoxicol. Environ. Saf.171, *701-708*. https://doi.org/10.1016/j.ecoenv.2019.01.030.

Ferioli, F., Giambanelli, E., D'Antuono, L. F., Costa, H. S., Albuquerque, T. G., Silva, A. S., ... Koçaoglu, B., 2013. Comparison of leafy kale populations from Italy, Portugal, and Turkey for their bioactive compound content: phenolics, glucosinolates, carotenoids, and chlorophylls. J. Sci. Food Agric. 93(14), 3478-3489. https://doi.org/10.1002/jsfa.6253.

Fernandes, F., Pereira, D. M., de Pinho, P. G., Valentão, P., Pereira, J. A., Bento, A., Andrade, P. B. 2010. Headspace solid-phase microextraction and gas chromatography/ion trap-mass spectrometry applied to a living system: *Pieris brassicae* fed with kale. Food chem. 119(4), *1681-1693*. http://dx.doi.org/10.1016/j.foodchem.2009.09.046.

Francisco, M., Moreno, D.A., Cartea, M.E., Ferreres, F., García-Viguera, C., Velasco, P., 2009. Simultaneous identification of glucosinolates and phenolic compounds in a representative collection of vegetable *Brassica rapa*. J. Chromatogr. A 1216(*38*), *6611–6619*. https://doi.org/10.1016/j.chroma.2009.07.055.

Hagen, S. F., Borge, G. I. A., Solhaug, K. A., Bengtsson, G. B., 2009. Effect of cold storage and harvest date on bioactive compounds in curly kale (*Brassica oleracea* L. var. *acephala*). Postharvest Biol. Technol. 51(1), 36-42. https://doi.org/10.1016/j.postharvbio.2008.04.001.

Hahn, C., Müller, A., Kuhnert, N., Albach, D., 2016. Diversity of kale (*Brassica oleracea* var. *sabellica*): glucosinolate content and phylogenetic relationships. J. Agric. Food Chem. 64(16), 3215-3225. https://doi.org/10.1021/acs.jafc.6b01000.

Hammer, Ø., Harper, D. A., Ryan, P. D., 2001. PAST: Paleontological statistics software package for education and data analysis. Palaeontologia electronica 4(1), 9.

Heimler, D., Vignolini, P., Dini, M. G., Vincieri, F. F., Romani, A., 2006. Antiradical activity and polyphenol composition of local Brassicaceae edible varieties. Food chem. 99(3), 464-469. https://doi.org/10.1016/j.foodchem.2005.07.057.

Jayaprakasha, G. K., Jena B.S., Sakariah K. K., 2003. Improved Liquid Chromatographic Method for Determination of Organic Acids in Leaves, Pulp, Fruits, and Rinds of Garcinia. J. of AOAC International, 86(5), 1063-1068.

Jin, L., Zhang, Y., Yan, L., Guo, Y., Niu, L., 2012. Phenolic compounds and antioxidant activity of bulb extracts of six Lilium species native to China. Molecules, 17(8), 9361-9378. https://doi.org/10.3390/molecules17089361.

Jovanovic-Malinovska, R., Kuzmanova, S., Winkelhausen, E., 2014. Oligosaccharide profile in fruits and vegetables as sources of prebiotics and functional foods. Int. J. Food Prop.17(5), 949-965. https://doi.org/10.1080/10942912.2012.680221.

Kapusta-Duch, J., Kusznierewicz, B., Leszczyńska, T., Borczak, B., 2016. Effect of Culinary Treatment on Changes in the Contents of Selected Nutrients and Non-Nutrients in Curly Kale (*Brassica oleracea* Var. *acephala*). J. Food Process. Preserv. 40(6), *1280-1288*. https://doi.org/10.1111/jfpp.12713.

Kopsell, D. A., Kopsell, D. E., Curran-Celentano, J., 2007. Carotenoid pigments in kale are influenced by nitrogen concentration and form. J. Sci. Food Agric. 87(5), 900-907. https://doi.org/10.1002/JSFA.2807.

Korus, A., 2011. Level of vitamin C, polyphenols, and antioxidant and enzymatic activity in three varieties of kale (*Brassica oleracea* L. var. *acephala*) at different stages of maturity. Int. J. Food Prop. 14(5), *1069-1080*. https://doi.org/10.1080/10942910903580926.

Korus, A., 2020. Changes in the content of minerals, B-group vitamins and tocopherols in processed kale leaves. J. Food Compos. Anal. 89, 103464. https://doi.org/10.1016/j.jfca.2020.103464.

Korus, A., 2022. Effect of pre-treatment and drying methods on the content of minerals, Bgroup vitamins and tocopherols in kale (*Brassica oleracea* L. var. *acephala*) leaves. J. Food Sci. Technol. 59(1), 279-287. https://doi.org/10.1007/s13197-021-05012-9.

Lamb F C, 1972. Modification of AOAC Method for Determination of Soluble Solids in Tomato Products. Journal of Association of Official Analytical Chemists, 55(4), 809–810, https://doi.org/10.1093/jaoac/55.4.809

Łata, B., 2014. Variability in enzymatic and non-enzymatic antioxidants in red and greenleafy kale in relation to soil type and N-level. Sci. Hortic. 168, *38-45*. https://doi.org/10.1016/j.scienta.2014.01.009.

Lefsrud, M., Kopsell, D., Sams, C, Wills, J., Both, A.J., 2008. Dry matter content and stability of carotenoids in kale and spinach during drying. HortScience 43(6) 1731-1736. https://doi.org/10.21273/HORTSCI.43.6.1731.

Li B. W., Andrews K. W., Pehrsson P. R., 2002. Individual Sugars, Soluble, and Insoluble Dietary Fiber Contents of 70 High Consumption Foods. J. Food Comp. Anal., 15, 715-723. doi:10.1006/jfca.2002.1096

Li, Z., Liu, Y., Fang, Z., Yang, L., Zhuang, M., Zhang, Y., ..., Sun, P., 2016. Analysis of sulforaphane in Chinese kale (*Brassica albograbra* LH Bailey) by HPLC. China Vegetables 4, 53-57.

Liang, H., Yuan, Q. P., Dong, H. R., Liu, Y. M., 2006. Determination of sulforaphane in broccoli and cabbage by high-performance liquid chromatography. J. Food Compos. Anal. 19(5), 473-476. https://doi.org/10.1016/j.jfca.2005.11.005.

Lichtenthaler, H. K., Wellburn, A. R., 1983. Determinations of total carotenoids and chlorophylls a and b of leaf extracts in different solvents. Biochem. Soc. Trans. 603, 591–592.

Lin, L.Z.; Sun, J.; Chen, P.; Harny, J. UHPLC-PDA-ESI/HRMS/MSⁿ analysis of anthocyanins, flavonol, glucosides, and hydroxycinnamic acid derivatives in red mustard greens

(*Brassica juncea* Coss variety). J. Agric. Food. Chem. 2011, 59, 12059–12072. https://doi.org/10.1021/jf202556p.

Ljubej, V., Karalija, E., Salopek-Sondi, B., Šamec, D., 2021. Effects of Short-Term Exposure to Low Temperatures on Proline, Pigments, and Phytochemicals Level in Kale (*Brassica oleracea var. acephala*). Horticulturae 7(*10*), 341. https://doi.org/10.3390/horticulturae7100341.

Lo Scalzo, R., Campanelli, G., Paolo, D., Fibiani, M., Bianchi G., 2020. Influence of organic cultivation and sampling year on quality indexes of sweet peppers during 3 years of production. Eur. Food Res, Technol. 246(6), *1325-1339*. https://doi.org/10.1007/s00217-020-03492-1.

Maggioni, L., von Bothmer, R., Poulsen, G., Branca, F., Bagger Jørgensen, R., 2014. Genetic diversity and population structure of leafy kale and *Brassica rupestris* Raf. in south Italy. *Hereditas*, *151*(6), 145-158. https://doi.org/10.1111/hrd2.00058.

Manchali, S., Murthy, K. N. C., Patil, B. S., 2012. Crucial facts about health benefits of popular cruciferous vegetables. J. Funct. Foods 4(1), 94-106. https://doi.org/10.1016/j.jff.2011.08.004.

Marks, H. S., Hilson, J. A., Leichtweis, H. C., Stoewsand, G. S., 1992. S-Methylcysteine sulfoxide in *Brassica* vegetables and formation of methyl methanethiosulfinate from Brussels sprouts. J. Agric. Food Chem. 40(11), 2098-2101. https://doi.org/10.1021/jf00023a012.

Martínez, S., Olmos, I., Carballo, J., Franco, I., 2010. Quality parameters of *Brassica* spp. grown in northwest Spain. Int. J. Food Sci. Technol. 45(4), 776-783. https://doi.org/10.1111/j.1365-2621.2010.02198.x.

Matusheski N.V., Juvik J A, Jeffery E H, 2004. Heating decreases epithiospecifier protein activity and increases sulforaphane formation in broccoli, Phytochemistry, 65(9), 1273-1281, https://doi.org/10.1016/j.phytochem.2004.04.013.

Meckelmann, S. W., Riegel, D. W., Van Zonneveld, M., Ríos, L., Peña, K., Mueller-Seitz, E., Petz, M., 2015. Capsaicinoids, flavonoids, tocopherols, antioxidant capacity and color a ttributes in 23 native Peruvian chili peppers (*Capsicum* spp.) grown in three different locations Eur. Food Res. Technol. 240(2), 273-283. https://doi.org/10.1007/s00217-014-2325-6.

Megías-Pérez, R., Hahn, C., Ruiz-Matute, A. I., Behrends, B., Albach, D. C., Kuhnert, N., 2020. Changes in low molecular weight carbohydrates in kale during development and acclimation to cold temperatures determined by chromatographic techniques coupled to mass spectrometry. Food Res. Int. 127, *108727*. https://doi.org/10.1016/j.foodres.2019.108727.

Naguib, A. E. M. M., El-Baz, F. K., Salama, Z. A., Hanaa, H. A. E. B., Ali, H. F., Gaafar, A. A., 2012. Enhancement of phenolics, flavonoids and glucosinolates of Broccoli (*Brassica oleracea*, var. *Italica*) as antioxidants in response to organic and bio-organic fertilizers. J. Saudi Soc. Agric. Sci., 11(2), *135-142*. https://doi.org/10.1016/j.jssas.2012.03.001.

Nemzer, B., Al-Taher, F., Abshiru, N., 2021. Extraction and natural bioactive molecules characterization in spinach, kale and purslane: A comparative study. Molecules 26(9), 2515. https://doi.org/10.3390/molecules26092515. Noctor, G., Foyer, C. H., 1998. Ascorbate and glutathione: keeping active oxygen under control. Annu. Rev. Plant Biol. 49(1), 249-279. https://doi.org/10.1146/annurev.arplant.49.1.249.

Oh, J. & Cho, I.H., 2021. The aroma profile and aroma-active compounds of *Brassica* oleracea (Kale) tea. Food Sci. Biotechnol., 30(9), *1205-1211*. https://doi.org/10.1007/s10068-021-00962-4.

Olsen, H., Aaby, K., Borge, G. I. A., 2009. Characterization and quantification of flavonoids and hydroxycinnamic acids in curly kale (*Brassica oleracea* L. convar. *acephala* var. *sabellica*) by HPLC-DAD-ESI-MS. J. Agric. Food Chem. 57(7), 2816-2825. https://doi.org/10.1021/jf803693t.

Perry, A., Rasmussen, H., Johnson, E. J., 2009. Xanthophyll (lutein, zeaxanthin) content in fruits, vegetables and corn and egg products. J Food Compos. Anal. 22(1), 9-15. https://doi.org/10.1016/j.jfca.2008.07.006.

Picchi, V., Lo Scalzo, R., Tava, A., Doria, F., Argento, S., Toscano, S., Treccarichi, S., Branca, F., 2020. Phytochemical characterization and in vitro antioxidant properties of four *Brassica* wild species from Italy. Molecules 25, *3495*. https://doi.org/10.3390/molecules25153495.

Picchi, V., Migliori, C., Lo Scalzo, R., Campanelli, G., Ferrari, V., Di Cesare, L. F., 2012. Phytochemical content in organic and conventionally grown Italian cauliflower. Food Chem. 130(*3*), *501-509*. https://doi.org/10.1016/j.foodchem.2011.07.036.

Picchi, V., Monga, R., Marzuoli, R., Gerosa, G., Faoro, F., 2017. The ozone-like syndrome in durum wheat (*Triticum durum* Desf.): Mechanisms underlying the different symptomatic responses of two sensitive cultivars. Plant Physiol. Biochem. 112, 261-269. https://doi.org/10.1016/j.plaphy.2017.01.011.

Pilipczuk, T., Dawidowska, N., Kusznierewicz, B., Namieśnik, J., Bartoszek, A. 2015. Simultaneous determination of indolic compounds in plant extracts by solid-phase extraction and high-performance liquid chromatography with UV and fluorescence detection. Food anal. methods 8, 2169-2177. https://doi.org/10.1007/s12161-015-0106-x.

Pilipczuk, T., Kusznierewicz, B., Chmiel, T., Przychodzeń, W., Bartoszek, A., 2017. Simultaneous determination of individual isothiocyanates in plant samples by HPLC-DAD-MS following SPE and derivatization with N-acetyl-l-cysteine. Food chem. 214, 587-596. https://doi.org/10.1016/j.foodchem.2016.07.125.

Radošević, K., Srček, V. G., Bubalo, M. C., Rimac B., S., Takács, K., Redovniković, I. R., 2017. Assessment of glucosinolates, antioxidative and antiproliferative activity of broccoli and collard extracts. Journal of Food Composition and Analysis, 61, 59-66. http://dx.doi.org/10.1016/j.jfca.2017.02.001

Riener, C.K., Kada, G., Gruber, H.J., 2002. Quick measurement of protein sulfhydryls with Ellman's reagent and with 4,4'-dithiodipyridine. Anal Bioanal Chem 373, 266–276. https://doi.org/10.1007/s00216-002-1347-2

Rosa, E., Heaney, R., 1996. Seasonal glucosinolate variation in protein, mineral and composition of Portuguese cabbages and kale. Anim. Feed Sci. Technol, 57(11), 111-127. https://doi.org/10.1016/0377-8401(95)00841-1.

Šamec, D., Urlić, B., Salopek-Sondi, B., 2019. Kale (*Brassica oleracea* var. *acephala*) as a superfood: Review of the scientific evidence behind the statement. Crit. Rev. Food Sci. Nutr. 59(15), 2411-2422. https://doi.org/10.1080/10408398.2018.1454400.

Sarıkamış, G., Balkaya, A., Yanmaz R., 2008. Glucosinolates in Kale Genotypes from the Blacksea Region of Turkey. Biotechnol. Biotechnol. Equip. 22(4), 942-946. https://doi.org/10.1080/13102818.2008.10817584.

Sasaki, K., Neyazaki, M., Shindo, K., Ogawa, T., Momose, M., 2012. Quantitative profiling of glucosinolates by LC–MS analysis reveals several cultivars of cabbage and kale as promising sources of sulforaphane. J. Chromatogr. B 903, 171-176. https://doi.org/10.1016/j.jchromb.2012.07.017.

Singh, A. A., Patil, M. P., Kang, M. J., Niyonizigiye, I., Kim, G. D., 2021. Biomedical application of indole-3-carbinol: A mini-review. Phytochem. Lett. 41, 49-54. https://doi.org/10.1016/j.phytol.2020.09.024.

Sivakumar, G., Aliboni, A., Bacchetta, L., 2007. HPLC screening of anti-cancer sulforaphane from important European *Brassica* species. Food Chem. 104(4), *1761-1764*. https://doi.org/10.1016/j.foodchem.2006.11.040.

Soengas, P., Cartea, M. E., Francisco, M., Sotelo, T., Velasco, P., 2012. New insights into antioxidant activity of *Brassica* crops. Food chem. 134(2), 725-733. https://doi.org/10.1016/j.foodchem.2012.02.169.

Son, Y. J., Park, J. E., Kim, J., Yoo, G., Lee, T. S., Nho, C. W., 2021. Production of low potassium kale with increased glucosinolate content from vertical farming as a novel dietary option for renal dysfunction patients. Food Chem. 339, *128092*. https://doi.org/10.1016/j.foodchem.2020.128092.

Steinbrecher, A., Rohrmann, S., Timofeeva, M., Risch, A., Jansen, E., Linseisen, J., 2010. Dietary glucosinolate intake, polymorphisms in selected biotransformation enzymes, and risk of prostate cancer. Cancer Epidemiol. Biomarkers Prev. 19(1), 135-143. https://doi.org/10.1158/1055-9965.EPI-09-0660.

Thavarajah, D., Lawrence, T., Powers, S., Jones, B., Johnson, N., Kay, J., Bandaranayake, A., ... Thavarajah, P., 2021. Genetic variation in the prebiotic carbohydrate and mineral composition of kale (*Brassica oleracea* L. var. *acephala*) adapted to an organic cropping system. J Food Compos. Anal. 96, *103718*. https://doi.org/10.1016/j.jfca.2020.103718.

Traka, M. H., Saha, S., Huseby, S., Kopriva, S., Walley, P. G., Barker, G. C., ..., Mithen, R. F., 2013. Genetic regulation of glucoraphanin accumulation in Beneforté® broccoli. New Phytol. 198(4), *1085-1095*. doi: 10.1111/nph.12232.

Treccarichi, S., Ben Ammar, H., Amari, M., Cali, R., Tribulato, A., Branca, F., 2023. Molecular Markers for Detecting Inflorescence Size of Brassica oleracea L. Crops and B. oleracea Complex Species (n = 9) Useful for Breeding of Broccoli (B. oleracea var. italica) and Cauliflower (B. oleracea var. botrytis). Plants, 12(2):407 https://doi.org/10.3390/plants12020407.

Tribulato, A., Toscano, S., Di Bella, M.C., Romano, D., Branca, F., 2020. *Brassica oleracea* complex species in Sicily: diversity, uses and conservation strategies. Acta Hortic. 1297, 61-68. https://doi.org/10.17660/ActaHortic.2020.1297.9

Vargas, L., Kapoor, R., Nemzer, B., Feng, H., 2022. Application of different drying methods for evaluation of phytochemical content and physical properties of broccoli, kale, and spinach. LWT, 155, *112892*. https://doi.org/10.1016/j.lwt.2021.112892.

Velasco, P., Cartea, M. E., González, C., Vilar, M., Ordás, A., 2007. Factors affecting the glucosinolate content of kale (*Brassica oleracea acephala* group). J. Agric. Food Chem. 55(3), 955-962. https://doi.org/10.1021/jf0624897.

Velasco, P., Francisco, M., Moreno, D.A., Ferreres, F., García-Viguera, C., Cartea, M.E., 2011. Phytochemical fingerprinting of vegetable *Brassica oleracea* and *Brassica napus* by simultaneous identification of glucosinolates and phenolics. Phytochem. Anal. 22, 144–152. https://doi.org/10.1002/pca.1259.

West, L. G., Meyer, K. A., Balch, B. A., Rossi, F. J., Schultz, M. R., Haas, G. W., 2004. Glucoraphanin and 4-hydroxyglucobrassicin contents in seeds of 59 cultivars of broccoli, raab, kohlrabi, radish, cauliflower, brussels sprouts, kale, and cabbage. J. Agric. Food Chem. 52(4), 916-926. https://doi.org/10.1021/jf0307189.

Wibowo, S., Afuape, A. L., De Man, S., Bernaert, N., Van Droogenbroeck, B., Grauwet, T., ... Hendrickx, M., 2019. Thermal processing of kale purée: The impact of process intensity and storage on different quality related aspects. Innov. Food Sci. Emerg. Technol. 58, *102213*. https://doi.org/10.1016/j.ifset.2019.102213.

Yang, S.C., Arasu, M.V., Chun, J.H., Jang, Y.S., Lee, Y.H. Kim, I.H., Lee, K.T., Hong, S.T., Kim, S.J., 2015. Identification and determination of phenolic compounds in rapeseed meals (*Brassica napus* L.). J. Agric. Chem. Env. 4, *14–23*. https://doi.org/10.4236/jacen.2015.41002.

Yu, L., Gao, B., Li, Y., Wang, T. T., Luo, Y., Wang, J., Yu, L. L., 2018. Home food preparation techniques impacted the availability of natural antioxidants and bioactivities in kale and broccoli. Food Funct. 9(1), 585-593. https://doi.org/10.1039/C7FO00948H.

Zeng W., Tao H., Li Y., Wang J., Xia C., Li S., Wang M., Wang Q., Miao H., 2021. The flavor of Chinese kale sprouts is affected by genotipic variation of glucosinolates and their breakdown products. Food Chem., 359:129824. https://doi.org/10.1016/j.foodchem.2021.129824.









Figure 1. HPLC chromatogram (UV 330 nm) of raw extract of the *Brassica oleracea* var. *acephala* sample. For compound identification see Table S1.

Figure 2. Content (μ mol/g d.w.) of sulforaphane (SRA) and indole-3-carbinol (I3C) in the seven genotypes of kale. Data are represented as means and error bars refer to the standard deviation. Different letters above each data series indicate significant differences among genotypes according to the Tukey's HSD test.

Figure 3. Content (mg/100 g d.w.) of total hydroxycinnaic acid derivatives (HAD), flavonol derivatives (quercetin, QD, and kaempferol, KD, derivatives) and hydroxycinnamoyl gentiobiosides (HCG) in the seven genotypes of kale. Data are represented as mean and bars refer to the standard deviation. Different letters above each data series indicate significant differences among genotypes according to the Tukey's HSD test.

Figure 4. PCA of average values of main parameters: biplot of the first two components.

(Footnote): T-TH, total thiols; HDPS, high degree polymerization sugars; d-toc, delta-tocopherol; FRU, fructose; itc, isothiocyanate; DM3S: dimethyl trisulphide; 6M5H, 6-methyl-5-hepten-2-one

Species	Trivial	Genebank code	BRESOV id code*	Plant codes	Seeds production
	name				
B. oleracea cv acephala	kale	UL 5010	B1800343	93-1, 93-3, 93-5	University of Liverpool, UK
B. oleracea cv acephala	kale	UL 2075	B1800315	55-1, 55-3, 55-5	University of Liverpool, UK
B. oleracea cv acephala	kale	UL 2066	B1800309	100-1, 100-2, 100-4	University of Liverpool, UK
B. oleracea cv acephala	kale	UNICT 3332	B1800091	145-1, 145-4, 145-7	University of Catania, Di3A
B. oleracea cv acephala	kale	UNICT 375	B1800087	215-4, 215-6, 215-7	University of Catania, Di3A
B. oleracea cv acephala	kale	UNICT 4946	NA	218-1, 218-2, 218-6	University of Catania, Di3A
B. oleracea cv acephala	kale	UNICT 4959	B1800903	219-1, 219-7, 219-8	University of Catania, Di3A

Table 1. List of the seven selected genotypes of *B. oleracea* cv *acephala* analysed in the present work.

*BRESOV id code lists all the accessions studied in the frame of the H2020 BRESOV, NA means not available.

Genotype		SS	С	High	DP	sugars	S	ucro	se	G	luco	se	Fructose			
UL 5010	41.1	±	3.2 a	10.8	±	3.4a	0.1	±	0.0	3.8	±	2.7	2.8	±	0.8ab	
UL 2075	40.5	±	2.3 ab	9.3	±	3.1ab	0.1	±	0.1	5.3	±	0.6	2.8	±	0.2ab	
UL 2066	37.6	±	2.6 ab	6.0	±	0.5ab	0.1	±	0.0	9.0	±	1.9	4.5	±	0.6a	
UNICT 3332	26.8	±	3.3 ab	5.4	±	1.4ab	0.0	±	0.0	3.9	±	2.1	1.8	±	0.4b	
UNICT 375	27.4	±	3.5 ab	4.8	±	0.9ab	0.6	±	0.5	5.4	±	3.9	2.6	±	0.7b	
UNICT 4946	29.3	±	6.9 ab	4.8	±	2.3ab	0.5	±	0.1	4.6	±	1.3	2.5	±	0.5b	
UNICT 4959	26.2	±	10.3 b	4.2	±	2.6b	0.5	±	0.4	4.7	±	1.5	2.7	±	0.9b	
Р		**			*			ns			ns			**	:	

Table 2. Soluble solids content (SSC, °Bx on d.w.), high DP sugars and single sugars content (g/100g d.w.) of B. oleracea cv acephala

Results are reported as means \pm standard deviation. Different letters on the same column indicate significant differences among genotypes according to the Tukey's HSD test. *P* significance: ns >0.05, * < 0.05, ** <0.01, *** <0.001

Genotype	Citric acid	Malic acid	Pyruvic acid	Fumaric acid	AsA	Total thiols	AsA/total thiols		
UL 5010	2347 ± 286	3175 ± 210a	148 ± 54a	199 ± 13a	222.5 ± 25.5	106.6 ± 9.8 ab	2.1 ± 0.2 b		
UL 2075	3743 ± 984	3072 ± 750a	154 ± 49a	63 ± 25b	$243.8 \hspace{0.2cm} \pm \hspace{0.2cm} 28.1$	$77.7 \text{ b} \pm 13.4 \text{ b}$	3.2 ± 0.7 a		
UL 2066	1962 ± 351	1896 ± 376b	149 ± 43a	$28 \pm 7b$	235.3 ± 34.3	110.7 ± 12.0 ab	$2.1 \hspace{.1in} \pm \hspace{.1in} 0.3 \hspace{.1in} b$		
UNICT 3332	2693 ± 726	$664 \pm 232c$	$151 \pm 43a$	$51 \pm 46b$	212.7 ± 28.0	119.9 a ± 19.6 a	$1.8 \hspace{0.2cm} \pm \hspace{0.2cm} 0.5 \hspace{0.2cm} b$		
UNICT 375	$3186 \hspace{0.1in} \pm \hspace{0.1in} 1137$	$736 \pm 427c$	$36 \pm 7b$	$24 \pm 2b$	$219.3 \hspace{0.1in} \pm \hspace{0.1in} 8.6$	$112.9 \pm 9.2 \text{ ab}$	$2.0 \hspace{.1in} \pm \hspace{.1in} 0.2 \hspace{.1in} b$		
UNICT 4946	$3109 \hspace{0.1in} \pm \hspace{0.1in} 1555$	$454 \pm 175c$	$60 \pm 23ab$	$36 \pm 22b$	$215.6 \hspace{0.2cm} \pm \hspace{0.2cm} 1.9$	$105.8 \pm 9.1 \text{ ab}$	$2.0 \hspace{0.2cm} \pm \hspace{0.2cm} 0.2 \hspace{0.2cm} b$		
UNICT 4959	$2800 \hspace{0.1in} \pm \hspace{0.1in} 1026$	$601 \pm 216c$	81 ± 14ab	$40 \pm 13b$	$209.4 \hspace{0.2cm} \pm \hspace{0.2cm} 10.0$	100.7 ± 16.4 ab	2.1 ± 0.2 b		
Р	ns	***	**	***	ns	*	**		

Table 3. Organic acids, ascorbic acid (AsA) and total thiols content (mg/100 g d.w.) of *B. oleracea* cv acephala

Results are reported as means \pm standard deviation. Different letters on the same column indicate significant differences among genotypes according to the Tukey's HSD test. *P* significance: ns >0.05, * < 0.05, ** <0.01, *** <0.001

Compound	U	JL 20)75	U	L 50	10		UL	2066		UNI	CT 3332	2	UN	ICT 375		UN	ICT 4946	ι	JNIC	T 4959	Р
3-Methyl butanal	19.8	±	9.0a	1.5	±	0.6b	10.0	±	4.2ab	0.7	±	0.2b	0.1	±	0.0b	0.8	±	0.1b	0.3	±	0.1b	*
2-Methyl butanal	5.8	±	0.3a	5.3	±	2.3ab	2.1	±	0.3b	0.8	±	0.3 b	0.5	±	0.2 b	6.0	±	0.9 a	5.1	±	1.3ab	**
Hexanal	8.3	±	0.8b	6.5	±	0.7bc	6.2	±		1.0	±	0.1c	2.0	±	0.7c	17.	±	1.9a	8.4	±	2.4b	***
									0.3bc							1						
(E)-2-Hexenal	63.1	±	3.7ab	100.0	±	27.9a	56.2	±		10.	±	3.6b	10.5	±	2.4b	68.	±	10.1a	65.6	±	4.3a	***
									5.0ab	4						1						
Allyl isothiocyanate	141.3	±	123.3	62.8	±	51.5	4.5	±	1.8	1.5	±	0.8	0.7	±	0.2	1.0	±	0.2	2.2	±	0.6	ns
Dimethyl trisulfide	15.1	±	9.5ab	23.9	±	7.8a	10.6	±	1.3ab	1.5	±	0.5ab	0.5	±	0.0b	7.7	±	0.9ab	7.4	±	1.4ab	*
3-Butenyl isothiocyanate	539.0	±	381.0	104.4	±	58.2	15.4	±		1.7	±	1.2	0.4	±	0.3	14.	±	11.0	7.1	±	5.7	ns
									8.1							8						
6-Methyl-5-hepten-2-one	6.4	±	1.1ab	6.8	±	0.3a	3.8	±	0.7b	0.6	±	0.2c	0.7	±	0.1c	5.3	±	0.4ab	4.3	±	0.5ab	***
Phenylacetaldehyde	3.1	±	0.1ab	4.1	±	0.4a	5.9	±	1.2a	0.5	±	0.1b	0.6	±	0.0b	4.5	±	0.2a	5.3	±	0.9a	***
Phenylethanone	3.5	±	0.9ab	3.9	±	0.5a	1.9	±	0.3abc	0.2	±	0.0c	0.3	±	0.1c	2.4	±	0.4abc	1.6	±	0.1bc	***
3,5-Octadien-2-one	3.6	±	2.1	1.2	±	0.6	0.8	±	0.3	0.1	±	0.0	0.3	±	0.1	3.7	±	0.6	2.2	±	0.9	ns
4,5-epithiovaleronitrile	23.8	±	18.0	3.2	±	1.4	7.0	±	2.0	0.5	±	0.0	0.3	±	0.0	6.2	±	1.6	4.2	±	2.8	ns
Nonanal	6.2	±	2.6a	1.3	±	0.5ab	1.1	±	0.3ab	0.2	±	0.1b	0.2	±	0.0b	1.7	±	0.4ab	1.0	±	0.2b	*
2,6-Dimethylcyclohexanol	3.3	±	0.4abc	4.7	±	1.0a	4.3	±	1.9ab	0.6	±	0.2bc	0.3	±	0.0c	2.0	±	0.1abc	2.2	±	0.2abc	**
Geranyl acetone	1.9	±	0.3a	1.5	±	0.2a	1.2	±	0.2ab	0.1	±	0.0c	0.3	±	0.1bc	1.3	±	0.3a	0.9	±	0.1abc	***
β-Ionone	4.1	±	1.2a	2.0	\pm	0.4ab	1.8	±	0.4ab	0.3	\pm	0.1b	0.3	±	0.0b	2.6	±	0.2ab	2.7	±	0.4ab	**

Table 4. Volatile compounds amount (ng/g d.w.) in B. oleracea cv acephala

Results are reported as means \pm standard deviation. Different letters on the same row indicate significant differences among genotypes according to the Tukey's HSD test. *P* significance: ns >0.05, * < 0.05, ** <0.01, *** <0.001. Compound identification is described in Table S3.

Genotype		Chl	a		Ch	1 b	Total	enoids			
UL 5010	199.8	±	78.3a	71.1	±	22.1 a	27.9	±	14.0		
UL 2075	114.9	±	4.8 ab	39.9	±	0.9 abc	16.7	±	1.9		
UL 2066	167.4	±	25.5ab	59.2	±	6.3 ab	25.3	±	5.4		
UNICT 3332	85.4	±	22.2b	29.8	±	9.1 bc	11.9	±	1.0		
UNICT 375	106.8	±	16.6ab	31.6	±	17.9 bc	17,9	±	3.2		
UNICT 4946	86.5	±	28.0b	25.0	±	8.7 c	14.2	±	3.8		
UNICT 4959	$125.1 \pm 40.4ab$		39,6	±	18.6 abc	21.3	±	6.5			
P		*			*	*	ns				

Table 5. Chlorophyll a, chlorophyll b and total carotenoid contents (mg/100 g d.w.) in B. oleracea cv acephala

Results are reported as means \pm standard deviation. Different letters on the same column indicate significant differences among genotypes according to the Tukey's HSD test. P significance: ns >0.05, * < 0.05, ** <0.01, *** < 0.001

F			$I \longrightarrow 0$	0											
Genotype		α-tocoph	erol	δ-	tocophe	rol	γ	-tocoph	erol	Total tocopherols					
UL 5010	15.0	±	10.1	0.5	\pm	0.4	1.5	<u>±</u>	1.2	16.9	\pm	11.7			
UL 2075	11.1	±	3.2	0.9	±	0.4	1.7	<u>+</u>	0.98	13.7	\pm	4.2			
UL 2066	14.8	±	9.3	0.5	±	0.2	1.2	±	0.4	16.5	<u>+</u>	11.7			
UNICT 3332	10.0	±	4.9	0.2	±	0.2	0.3	±	0.2	10.5	<u>+</u>	5.0			
UNICT 375	22.0	±	5.3	0.3	±	0.2	0.6	±	0.2	22.8	<u>+</u>	5.5			
UNICT 4946	20.4	±	5.7	0.3	±	0.2	0.7	±	0.4	21.3	\pm	6.0			
UNICT 4959	21.2	±	8.4	0.4	±	0.3	0.6	±	0.1	22.2	±	8.5			
Р		ns		ns				ns		ns					

Table 6. Tocopherols content in *B. oleracea* cv *acephala* (mg/100 g d.w.)

Results are reported as means \pm standard deviation. Different letters on the same column indicate significant differences among genotypes according to the Tukey's HSD test. P significance: ns >0.05, * < 0.05, ** <0.01, *** <0.001

UV $\lambda_{max}(nm)$	[M+H] ⁺ /[M-H] ⁻	Diagnostic MS ² negative ions, m/z (%)	Formula	MW	Compound ^a
240, 298, 327	-/353	191(100): 179(43): 135(10)	C16H18O0	354	Caffeovlauinic acid
226, 310	-/337	191(9): 163(100): 119(5)	C ₁₆ H ₁₈ O ₈	338	<i>p</i> -Coumarovlouinic acid
254, 266sh, 352	789/787	625(100); 463(21); 301(79)	$C_{30}H_{40}O_{22}$	788	O-3- <i>O</i> -diglucoside-7- <i>O</i> -glucoside
256, 266, 334	981/979	817(100); 787(85); 625(51); 301(3)	C ₄₃ H ₄₈ O ₂₆	980	Q-3- <i>O</i> -hydroxyferuloyldiglucoside-7- <i>O</i> -glucoside
265, 316sh, 346	773/771	609(100); 447(35); 285(38)	$C_{33}H_{40}O_{21}$	772	K-3- <i>O</i> -diglucoside-7- <i>O</i> -glucoside
252, 266sh, 352	951/949	787(100); 625(95); 301(7)	C ₃₉ H ₅₀ O ₂₇	950	Q-3-O-diglucoside-7-O-diglucoside
256, 266, 332	951/949	787(100); 625(61); 463(3); 301(3)	$C_{42}H_{46}O_{25}$	950	Q-3-O-caffeoyldiglucoside-7-O-glucoside
256, 266, 332	1143/1141	949(100); 817(18); 625(33); 301(1)	$C_{49}H_{58}O_{31}$	1142	Q-3-O-hydroxyferuloyldiglucoside-7-O-diglucoside
255, 266, 334	1113/1111	949(100); 787(52); 301(2)	$C_{48}H_{56}O_{30}$	1112	Q-3-O-caffeoyltriglucoside-7-O-glucoside
266, 332	965/963	801(100); 609(4); 285(2)	$C_{43}H_{48}O_{25}$	964	K-3-O-hydroxyferuloyldiglucoside-7-O-glucoside
266, 332	935/933	771(100); 609(5); 285(2)	$C_{42}H_{46}O_{24}$	934	K-3-O-caffeoyldiglucoside-7-O-glucoside
256, 266, 334	995/993	831(100); 787(88); 625(54); 301(3)	$C_{44}H_{50}O_{26}$	994	Q-3-O-sinapoyldiglucoside-7-O-glucoside
256, 266, 336	965/963	801(100); 787(45);625(28); 301(2)	$C_{43}H_{48}O_{25}$	964	Q-3-O-feruloyldiglucoside-7-O-glucoside
255, 266, 334	1157/1155	993(10); 949(100); 831(20); 625(38)	$C_{50}H_{60}O_{31}$	1156	Q-3-O-sinapoyltriglucoside-7-O-glucoside
266, 332	1097/1095	933(5); 771(100); 609(21); 285(1)	$C_{48}H_{56}O_{29}$	1096	K-3-O-caffeoyldiglucoside-7-O-diglucoside
256, 265, 334	1127/1125	963(5); 949(100); 801(48); 625(35)	$C_{49}H_{58}O_{30}$	1126	Q-3-O-feruloyltriglucoside-7-O-glucoside
266, 332	979/977	815(100); 609(5); 285(2)	$C_{44}H_{50}O_{25}$	978	K-3-O-sinapoyldiglucoside-7-O-glucoside
266, 332	1141/1139	977(10); 815(100);609(25); 285(1)	$C_{50}H_{60}O_{30}$	1140	K-3-O-sinapoyldiglucoside-7-O-diglucoside
266, 332	949/947	785(100); 609(7); 285(2)	$C_{43}H_{48}O_{24}$	948	K-3-O-feruloyldiglucoside-7-O-glucoside (25 isomer)
266, 332	1111/1109	947(2); 785(100); 609(26); 285(1)	$C_{49}H_{58}O_{29}$	1110	K-3-O-feruloyldiglucoside-7-O-diglucoside
266, 319	919/917	755(100); 609(3); 285(2)	$C_{42}H_{46}O_{23}$	918	K-3-O-coumaroyldiglucoside-7-O-glucoside
266, 319	1081/1079	755(100); 609(16); 285(1)	$C_{42}H_{46}O_{23}$	1080	K-3-O-coumaroyldiglucoside-7-O-diglucoside
254, 266, 335	819/817	625(100); 463(3); 301(2)	$C_{37}H_{38}O_{21}$	818	Q-3-O-hydroxyferuloyldiglucoside
256, 265sh, 353	627/625	463(72); 301(100)	$C_{27}H_{30}O_{17}$	626	Q-3- <i>O</i> -diglucoside
256, 266, 335	789/787	625(100); 463(1); 301(2)	$C_{36}H_{36}O_{20}$	788	Q-3-O-caffeoylglucoside-7-O-glucoside
266, 332	949/947	785(100); 609(3); 285(3)	$C_{43}H_{48}O_{24}$	948	K-3-O-feruloyldiglucoside-7-O-glucoside (18 isomer)
265, 295sh, 348	611/609	429(91);285(100)	$C_{27}H_{30}O_{16}$	610	K-3-O-diglucoside
	UV λ _{max} (nm) 240, 298, 327 226, 310 254, 266sh, 352 256, 266, 334 265, 316sh, 346 252, 266sh, 352 256, 266, 332 256, 266, 332 255, 266, 334 266, 332 256, 266, 334 266, 332 256, 266, 334 266, 332 256, 266, 334 266, 332 256, 266, 334 266, 332 256, 266, 334 266, 332 256, 266, 334 266, 332 266, 332 266, 332 266, 332 266, 332 266, 332 266, 332 266, 332 266, 332 266, 332 266, 319 256, 265sh, 353 256, 265sh, 353 256, 266, 335 266, 332 256, 266, 335 256, 266, 335 266, 332 256, 265sh, 353 265, 295sh, 348	UV λ _{max} (nm) [M+H] ⁺ /[M-H] ⁺ 240, 298, 327 -/353 226, 310 -/337 254, 266sh, 352 789/787 256, 266, 334 981/979 265, 316sh, 346 773/711 252, 266sh, 352 951/949 256, 266, 332 951/949 256, 266, 332 951/949 256, 266, 332 951/949 256, 266, 332 951/949 256, 266, 332 951/949 256, 266, 332 951/949 256, 266, 332 95/963 266, 332 95/963 256, 266, 334 995/993 256, 266, 334 995/993 256, 266, 334 995/993 256, 266, 334 1157/1155 266, 332 1097/1095 256, 265, 334 1127/1125 266, 332 979/977 266, 332 919/917 266, 332 949/947 266, 335 819/817 266, 335 819/817 266, 265 h, 353 627/625 256, 265 h, 353 <th>UV X_{max} (nm) [M+H]//[M-H] Diagnostic MS* negative ions, m/z (%) 240, 298, 327 -/353 191(100); 179(43); 135(10) 226, 310 -/337 191(9); 163(100); 119(5) 254, 266sh, 352 789/787 625(100); 463(21); 301(79) 256, 266, 334 981/979 817(100); 787(85); 625(51); 301(3) 265, 316sh, 346 773/771 609(100); 447(35); 285(38) 252, 266sh, 352 951/949 787(100); 625(95); 301(7) 256, 266, 332 951/949 787(100); 625(33); 301(3) 256, 266, 332 951/949 787(100); 625(33); 301(1) 255, 266, 334 1113/111 949(100); 787(52); 301(2) 266, 332 965/963 801(100); 609(4); 285(2) 266, 332 935/933 771(100); 609(5); 285(2) 266, 334 995/993 831(100); 787(45); 625(28); 301(3) 255, 266, 334 95/963 801(100); 787(45); 625(28); 301(2) 256, 266, 332 1097/1095 933(5); 771(100); 609(21); 285(1) 256, 266, 334 1157/1155 993(10); 949(100); 811(20); 625(35) 266, 332 1097/1095 933(5); 771(100); 609(21); 285(1)</th> <th>UV λ_{max} (nm) [M+H]/[M-H] Diagnostic MS' negative ions, m/z (%) Formula 240, 298, 327 -/353 191(100); 179(43); 135(10) $C_{16}H_{18}O_{9}$ 254, 266sh, 352 789/787 625(100); 463(21); 301(79) $C_{38}H_{49}O_{22}$ 256, 266, 334 981/979 817(100); 787(85); 625(51); 301(3) $C_{41}H_{49}O_{55}$ 265, 316sh, 346 773/771 609(100); 447(35); 285(38) $C_{33}H_{49}O_{21}$ 252, 266sh, 352 951/949 787(100); 625(95); 301(7) $C_{29}H_{59}O_{27}$ 256, 266, 332 951/949 787(100); 625(95); 301(7) $C_{29}H_{59}O_{27}$ 256, 266, 332 951/949 787(100); 625(53); 301(3) $C_{42}H_{40}O_{25}$ 256, 266, 332 951/949 787(100); 625(33); 301(1) $C_{39}H_{59}O_{21}$ 256, 266, 332 955/963 801(100); 609(4); 285(2) $C_{41}H_{40}O_{24}$ 256, 266, 334 995/993 831(100); 787(83); 625(54); 301(3) $C_{41}H_{50}O_{52}$ 256, 266, 334 995/993 831(100); 787(45); 625(28); 301(2) $C_{41}H_{50}O_{52}$ 256, 266, 334 1157/1155 993(10); 949(100); 831(20); 625(38)</th> <th>UV λ_{max} (nm)[M+H]/[M-H]'Diagnostic MS' negative ions, m/z (%)FormulaMW240, 298, 327-/353191(100); 179(43); 135(10)$C_{18}H_{18}O_{9}$354226, 310-/337191(9); 163(100); 119(5)$C_{18}H_{18}O_{5}$338254, 266sh, 352789/787625(100); 463(2); 301(79)$C_{38}H_{40}O_{22}$788265, 364981.979817(100); 787(85); 625(51); 301(3)$C_{38}H_{40}O_{55}$980265, 316sh, 346773/71609(100); 447(35); 285(38)$C_{33}H_{40}O_{27}$950256, 266, 332951.949787(100); 625(61); 463(3); 301(3)$C_{42}H_{48}O_{25}$950256, 266, 332951.949787(100); 625(61); 463(3); 301(3)$C_{42}H_{48}O_{25}$950256, 266, 3321143/114949(100); 817(18); 625(33); 301(1)$C_{68}H_{50}O_{50}$1112266, 332965/963801(100); 609(4); 285(2)$C_{41}H_{40}O_{25}$964266, 332935/933771(100); 609(5); 285(2)$C_{41}H_{40}O_{25}$964256, 266, 334995/993831(100); 787(45); 625(28); 301(2)$C_{41}H_{40}O_{25}$964256, 266, 334995/993831(100); 787(45); 625(28); 301(2)$C_{41}H_{40}O_{25}$964256, 266, 3321097/1095933(5); 771(100); 609(21); 285(1)$C_{61}H_{40}O_{25}$964256, 266, 3341157/1155993(10); 787(45); 625(28); 301(2)$C_{41}H_{40}O_{25}$964256, 266, 3321097/1095933(5); 771(100); 609(21); 285(1)$C_{81}H_{80}O_{26}$918266, 332<!--</th--></th>	UV X _{max} (nm) [M+H]//[M-H] Diagnostic MS* negative ions, m/z (%) 240, 298, 327 -/353 191(100); 179(43); 135(10) 226, 310 -/337 191(9); 163(100); 119(5) 254, 266sh, 352 789/787 625(100); 463(21); 301(79) 256, 266, 334 981/979 817(100); 787(85); 625(51); 301(3) 265, 316sh, 346 773/771 609(100); 447(35); 285(38) 252, 266sh, 352 951/949 787(100); 625(95); 301(7) 256, 266, 332 951/949 787(100); 625(33); 301(3) 256, 266, 332 951/949 787(100); 625(33); 301(1) 255, 266, 334 1113/111 949(100); 787(52); 301(2) 266, 332 965/963 801(100); 609(4); 285(2) 266, 332 935/933 771(100); 609(5); 285(2) 266, 334 995/993 831(100); 787(45); 625(28); 301(3) 255, 266, 334 95/963 801(100); 787(45); 625(28); 301(2) 256, 266, 332 1097/1095 933(5); 771(100); 609(21); 285(1) 256, 266, 334 1157/1155 993(10); 949(100); 811(20); 625(35) 266, 332 1097/1095 933(5); 771(100); 609(21); 285(1)	UV λ_{max} (nm) [M+H]/[M-H] Diagnostic MS' negative ions, m/z (%) Formula 240, 298, 327 -/353 191(100); 179(43); 135(10) $C_{16}H_{18}O_{9}$ 254, 266sh, 352 789/787 625(100); 463(21); 301(79) $C_{38}H_{49}O_{22}$ 256, 266, 334 981/979 817(100); 787(85); 625(51); 301(3) $C_{41}H_{49}O_{55}$ 265, 316sh, 346 773/771 609(100); 447(35); 285(38) $C_{33}H_{49}O_{21}$ 252, 266sh, 352 951/949 787(100); 625(95); 301(7) $C_{29}H_{59}O_{27}$ 256, 266, 332 951/949 787(100); 625(95); 301(7) $C_{29}H_{59}O_{27}$ 256, 266, 332 951/949 787(100); 625(53); 301(3) $C_{42}H_{40}O_{25}$ 256, 266, 332 951/949 787(100); 625(33); 301(1) $C_{39}H_{59}O_{21}$ 256, 266, 332 955/963 801(100); 609(4); 285(2) $C_{41}H_{40}O_{24}$ 256, 266, 334 995/993 831(100); 787(83); 625(54); 301(3) $C_{41}H_{50}O_{52}$ 256, 266, 334 995/993 831(100); 787(45); 625(28); 301(2) $C_{41}H_{50}O_{52}$ 256, 266, 334 1157/1155 993(10); 949(100); 831(20); 625(38)	UV λ_{max} (nm)[M+H]/[M-H]'Diagnostic MS' negative ions, m/z (%)FormulaMW240, 298, 327-/353191(100); 179(43); 135(10) $C_{18}H_{18}O_{9}$ 354226, 310-/337191(9); 163(100); 119(5) $C_{18}H_{18}O_{5}$ 338254, 266sh, 352789/787625(100); 463(2); 301(79) $C_{38}H_{40}O_{22}$ 788265, 364981.979817(100); 787(85); 625(51); 301(3) $C_{38}H_{40}O_{55}$ 980265, 316sh, 346773/71609(100); 447(35); 285(38) $C_{33}H_{40}O_{27}$ 950256, 266, 332951.949787(100); 625(61); 463(3); 301(3) $C_{42}H_{48}O_{25}$ 950256, 266, 332951.949787(100); 625(61); 463(3); 301(3) $C_{42}H_{48}O_{25}$ 950256, 266, 3321143/114949(100); 817(18); 625(33); 301(1) $C_{68}H_{50}O_{50}$ 1112266, 332965/963801(100); 609(4); 285(2) $C_{41}H_{40}O_{25}$ 964266, 332935/933771(100); 609(5); 285(2) $C_{41}H_{40}O_{25}$ 964256, 266, 334995/993831(100); 787(45); 625(28); 301(2) $C_{41}H_{40}O_{25}$ 964256, 266, 334995/993831(100); 787(45); 625(28); 301(2) $C_{41}H_{40}O_{25}$ 964256, 266, 3321097/1095933(5); 771(100); 609(21); 285(1) $C_{61}H_{40}O_{25}$ 964256, 266, 3341157/1155993(10); 787(45); 625(28); 301(2) $C_{41}H_{40}O_{25}$ 964256, 266, 3321097/1095933(5); 771(100); 609(21); 285(1) $C_{81}H_{80}O_{26}$ 918266, 332 </th

Table 7. Chemical characteristics of the tentatively identified phenolic components detected in the *Brassica oleracea* cv *acephala* samples.

28	266, 331	773/771	609(100); 285(5)	$C_{36}H_{36}O_{19}$	772	K-3-O-caffeoyldiglucoside
29	266, 331	817/815	609(100); 285(7)	$C_{38}H_{40}O_{20} \\$	816	K-3-O-sinapoyldiglucoside
30	266, 317	757/755	609(100); 285(11)	$C_{36}H_{36}O_{18}$	756	K-3-O-coumaroyldiglucoside
31	265, 331	787/785	609(100); 285(4)	$C_{37}H_{38}O_{19}$	786	K-3-O-feruloyldiglucoside
32	256, 266, 334	1363/1361	1199(100); 1155(45); 993(62); 787(6)	$C_{61}H_{70}O_{35}$	1362	Q-3-O-disinapoyltriglucoside-7-O-glucoside
33	256, 266, 332	1333/1331	1169(100); 1155(28); 1125(11); 930(35); 963(11)	$C_{60}H_{68}O_{34}$	1332	Q-3-O-sinapoylferuloyltriglucoside-7-O-glucoside
34	266, 348	449/447	285(100)	$C_{21}H_{20}O_{11}$	448	K-3-O-glucoside
35	256, 354	479/477	285(100)	$C_{22}H_{22}O_{12}$	478	I-3-O-glucoside
36	265, 365	449/447	285(100)	$C_{21}H_{20}O_{11}$	448	K-7-O-glucoside
37	256, 266sh, 368	479/477	285(100)	$C_{22}H_{22}O_{12}$	478	I-7- <i>O</i> -glucoside
38	240, 332	777 ^b /753	529(100)	$C_{34}H_{42}O_{19}$	754	1,2-disinapoyl gentiobioside
39	240, 332	747 ^b /723	529(18); 499(100)	$C_{33}H_{40}O_{18} \\$	724	1-sinapoyl-2-feruloyl gentiobioside
40	240, 331	717 ^b /693	499(100)	$C_{32}H_{48}O_{17}$	694	1,2-diferuloyl gentiobioside
41	239, 332	983 ^b /959	735(100); 511(18)	$C_{45}H_{52}O_{23}$	960	1,2,2'-trisinapoyl gentiobioside
42	239, 331	953 ^b /929	735(4); 705(100); 511(7)	$C_{44}H_{50}O_{22}$	930	1,2'-disinapoyl-2-feruloyl gentiobioside
43	240, 332	923 ^b /899	705(100); 675(38); 511(15)	$C_{43}H_{48}O_{21}$	900	1-sinapoyl-2,2'-diferuloyl gentiobioside

^aK: kaempferol; Q: quercetin; I: isorhamnetin; ^b[M+Na]⁺ for gentiobiosides

Table S2. Identification of volatile compounds listed in Table 2.

Compound	KI calc	KI tab (NIST)	MW	MS spectrum*
3-Methyl butanal	625	634	86	44(100), 58(50), 57(40), 41(40), 86(15)
2-Methyl butanal	645	639	86	44(100), 41(95), 43(90), 58(70), 86(10)
Dimethyl disulphide	717	718	94	94(100), 79(50), 45(40), 46(30), 61(30)
Hexanal	774	777	100	44(100), 56(100), 41(75), 57(70), 43(70)
(E)-2-Hexenal	826	830	98	41(100), 55(90), 69(90), 83(80), 39(60)
Allyl isothiocyanate	849	846	99	99(100), 41(50), 39(50), 72(30)
Dimethyl trisulphide	944	944	126	126(100), 45(40), 79(30), 111(20)
3-Butenyl isothiocyanate	947	951	113	72(100), 113(50), 55(30), 41(30), 39(20)
6-Methyl-5-hepten-2-one	964	966	126	43(100), 108(60), 41(50), 69(50), 126(10)
Phenylacetaldehyde	1006	1007	120	91(100), 92(40), 120(30), 85(20), 51(10)
Phenylethanone	1035	1044	120	105(100), 77(60), 43(50), 120(40), 51(30)
3,5-Octadien-2-one	1044	1046	124	95(100), 43(75), 81(50), 53(30), 124(20)
4,5-Epithiovaleronitrile	1062	1075	113	113(100), 45(50), 73(40), 60(30), 67(20)
Nonanal	1063	1078	142	57(100), 41(90), 43(90), 56(80), 124(10)
2,6-Dimethylcyclohexanol	1087	1098	128	71(100), 68(30), 95(25), 110(20), 128(10)
Geranyl acetone	1430	1428	194	43(100), 69(60), 41(40), 125(20), 136(20)
β -lonone	1464	1466	192	177(100), 149(50), 43(50)

* main product ion fragments and relative percentage amounts; m/z range 39-400

Table S1. Quantitative evaluation (mg/100g DW) of the tentatively identified polyphenols detected in the Brassica oleracea cv acephala samples. See Table 7 for compounds identification. HAD: hydroxycinnamic derivatives, QD: quercetin derivatives, KD: kaempferol derivatives, HCG: hydroxycinnamic gentiobiosides.

	1	1			B.			<i>B. oleracea</i> acephala <i>B. oleracea</i> acephala				phala	B. ole	racea ace	phala	<i>B. oleracea</i> acephala						
			UL 5010, kale		U	L 2075, I	kale	i	JL 2066, ka	le	ls	nello BH50	0	BH2	5 (Portoga	allo)		BH96		BH100 (Galati Man	nertino)
		93_5	93_3	93_1	55_5	55_3	55_1	100_1	100_2	100_4	145_1	145_4	145_7	215_7	215_6	215_4	218_1	218_6	218_2	219_1	219_7	219_8
#	HAD	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21
1	Caffeoylquinic acid	90.26	9.61	11.48	49.73	4.50	1.16	9.03	44.51	85.36	187.86	134.98	67.94	60.30	41.61	93.82	62.13	5.50	44.24	61.72	59.35	83.35
2	<i>p</i> -Coumaroylquinic acid	0.31	1.57	0.92	0.53	0.39	0.08	0.24	0.77	1.52	0.23	0.08	2.50	0.12	0.08	0.02	0.31	0.27	0.69	0.28	0.24	0.04
	QD																					
3	Q-3-O-diglucoside-7-O-glucoside	23.33	12.09	46.17	19.58	37.12	42.56	13.38	24.49	41.53	10.08	7.66	12.45	25.94	99.19	96.29	29.75	12.66	1.95	5.52	7.61	9.97
4	Q-3- <i>O</i> -hydroxyferuloyldiglucoside-7- <i>O</i> -glucoside	22.96	56.06	129.15	25.92	91.64	145.66	56.67	24.95	61.09	13.72	38.77	39.07	15.82	91.90	54.42	14.22	49.27	52.56	56.00	47.65	42.11
6	Q-3- <i>O</i> -diglucoside-7- <i>O</i> -diglucoside	27.66	5.81		16.80	10.19	14.09	4.51	23.83	33.53	46.39	45.56	14.62	20.58	91.97	38.11	83.37	67.22	107.89	58.20	54.25	36.52
7	Q-3-O-caffeoyldiglucoside-7-O-glucoside	30.87	39.63	80.53	17.12	6.03		15.03	9.44	29.01	5.26	33.05	13.51	11.17	23.85	32.20	22.15	67.66	142.38	24.80	82.00	86.35
8	Q-3-O-hydroxyferuloyldiglucoside-7-O-diglucoside		19.42		11.14		6.43	4.46			54.32	118.75	94.44	6.58	32.58		137.83	29.72		57.59	0.00	65.16
9	Q-3-O-caffeoyltriglucoside-7-O-glucoside	7.21	8.24	11.82	6.09	7.17	3.71	8.22	1.99	22.02	9.53	22.91	30.50	2.86	5.24	13.05	15.43	33.73	57.00	13.91	28.27	23.55
12	Q-3-O-sinapoyldiglucoside-7-O-glucoside	48.37	86.70	94.81	44.53	6.20	4.74	41.70	26.05	45.60	58.82	80.27	107.28	48.22	98.06	103.37	46.48	5.46		74.13	17.69	87.16
13	Q-3-O-feruloyldiglucoside-7-O-glucoside	33.37	80.22	89.21	44.78	26.97	12.03	48.20	42.51	45.74	93.09	145.82	146.68	41.69	98.01	99.83	106.37	58.24	33.94	54.96	45.34	123.19
14	Q-3-O-sinapoyltriglucoside-7-O-glucoside	3.51	3.50	2.85	1.41	3.52	2.00	0.60	0.86	1.42	1.16	0.29	0.87	1.47	5.69	1.44	4.36	4.26	11.45	4.40	7.35	1.03
16	Q-3-O-feruloyltriglucoside-7-O-glucoside	20.38	43.08	31.09	18.40	9.39	27.35	19.16	7.89	32.76	46.44	112.81	58.82	21.52	113.59	50.21	87.44	36.75	30.12	85.09	71.82	102.23
23	Q-3-O-hydroxyferuloyldiglucoside	2.82	4.59	7.40	1.00	1.45		5.50	3.73	3.85	35.77	8.12	21.18	4.43	5.00	7.41	18.97	4.55	8.18	2.57	5.84	7.87
24	Q-3-O-diglucoside	13.37	8.35	2.61	12.73	9.34	11.90	7.45	9.35	9.04	30.40	12.90	36.69	58.09	32.05	19.71	12.84	10.10	10.84	14.66	16.85	15.67
25	Q-3-O-caffeoylglucoside-7-O-glucoside	2.58	5.78	8.38	2.31	2.17	8.43	3.57	2.94	1.66	71.98	8.96	13.57	12.96	3.59	6.92	12.93	5.97	6.82	4.45	7.33	19.26
32	Q-3-O-disinapoyltriglucoside-7-O-glucoside	61.46	56.11	81.85	28.07	21.20	20.61	42.30	43.95	85.03	137.95	54.67	27.11	139.10	96.45	110.26	82.01	115.99	197.10	71.12	99.07	55.96
33	Q-3-O-sinapoylferuloyltriglucoside-7-O-glucoside	115.43	24.36	42.16	48.81	14.25	22.80	19.19	63.66	125.85	24.25	21.88	7.18	11.90	50.10	100.69	32.23	72.70	90.30	24.81	63.98	28.58
	KD																					
5	K-3-O-diglucoside-7-O-glucoside	167.15	25.18	36.15	89.41	51.12	70.61	16.90	90.05	130.30	44.61	18.73	25.38	86.19	157.93	271.00	77.73	35.19	12.22	8.66	24.29	7.80
10	K-3-O-hydroxyferuloyldiglucoside-7-O-glucoside	35.76	15.63	20.58	23.62	9.24	4.77	12.22	17.27	27.56	95.14	74.13	26.48	6.03	49.65	61.22	44.85	48.64	11.93	36.09	54.03	55.77
11	K-3-O-caffeoyldiglucoside-7-O-glucoside	145.30	82.44	129.42	83.12	17.84	18.47	51.71	62.83	145.87	43.45	130.76	120.19	61.46	155.20	189.20	166.49	202.81	213.02	92.45	167.88	188.32
15	K-3-O-caffeoyldiglucoside-7-O-diglucoside	12.96	9.32	54.19	10.10	19.17	9.40	32.90	11.94	36.66	68.99		2.45	8.71	8.09	39.65	5.51	17.20	19.10	15.76	2.79	18.12
17	K-3-O-sinapoyldiglucoside-7-O-glucoside	160.67	64.22	78.41	102.89	38.51	49.46	34.80	92.83	154.48	85.41	54.48	103.49	113.25	77.53	152.95	65.00	47.93	36.00	22.33	55.80	59.48
18	K-3-O-sinapoyldiglucoside-7-O-diglucoside	32.65	35.62	35.93	34.40	29.55	20.63	31.38	35.39	40.14	55.41	55.99	51.60	33.15	119.56	60.27	123.85	83.38	49.19	171.63	63.76	189.15
19	K-3-O-feruloyldiglucoside-7-O-glucoside (25 isomer)	92.29	68.29	73.49	85.03	70.28	61.69	48.75	88.55	86.31	64.43	78.67	147.74	73.60	82.78	146.29	108.23	89.91	49.54	11.91	62.90	85.62
20	K-3-O-feruloyldiglucoside-7-O-diglucoside	19.59	0.28	7.07	9.75	2.82	2.75	3.47	4.26	18.85	21.66	33.90	22.60	2.46	14.95	2.83	19.52	9.33	12.93	15.41	7.72	18.58
21	K-3-O-coumaroyldiglucoside-7-O-glucoside	37.48	3.15	21.57	19.32	24.17	3.83	3.32	14.27	34.31	27.29	6.43	17.50		8.25	16.67	6.57	10.40	3.97	3.04	4.83	5.00
22	K-3-O-coumaroyldiglucoside-7-O-diglucoside	0.27	6.81	9.77	2.63	0.00	36.93	6.71	0.27	0.26	0.27	0.03	13.61				0.27			0.27	0.27	6.03
26	K-3-O-feruloyldiglucoside-7-O-glucoside (18 isomer)	12.47	4.76	1.36	4.84	5.44	2.00		5.01	7.76			12.04	24.12			0.00	3.64	3.24	0.00	4.52	12.25
27	K-3-O-diglucoside	43.66	11.24	6.46	27.88	50.15	81.11	7.98	25.71	32.18	88.13	14.23	24.74	140.40	50.83	71.45	30.19	41.75	20.66	2.08	28.38	16.57
28	K-3-O-caffeoyldiglucoside	2.65	3.29	36.34	8.01	4.25	1.94	1.84	4.11	5.67	131.15	35.70	10.51	79.51	4.65	24.79	20.13	85.68	109.78	12.68	79.81	48.34
29	K-3-O-sinapoyldiglucoside	7.24	1.80		2.92	4.16	2.13		5.58	4.41	195.36		17.35	71.86	2.29	13.66	5.79	29.39	33.25	81.61	26.47	17.72
30	K-3-O-coumaroyldiglucoside	10.48	6.73	11.01	3.05		1.06		6.73	18.23	97.92	91.86	24.67	40.58	23.71	13.16	104.02	37.80	79.43	64.81	32.49	7.57
31	K-3-O-feruloyldiglucoside	13.95	5.39		8.88	7.55	12.22	5.50	9.80	13.73	47.94	68.89	23.84	5.25	2.67	11.00	66.66	7.47	11.84	7.34	13.83	15.13
34	K-3-O-glucoside	13.23	10.10	10.13	12.00	16.29	18.11	6.78	13.61	12.08		5.26	1.31	10.71	9.38	18.67	11.93	25.55	25.11	11.31	20.47	21.64
36	K-7-O-glucoside	10.91	0.94				4.03						0.47			4.26						
05	ISORHAMNETINS (FLAVONOLS)	0.00	4.07	0.57	0.04		4.07	4.00	0.40	0.04		4 70	0.05		4.00	F 40	4 70	40.74	40.00		0.40	0.00
35	1-3-O-glucoside	8.29	4.27	3.57	3.81	0.70	1.07	1.80	2.48	8.04	7 4 0	4.70	3.85		4.23	5.13	4.72	12.74	12.38		9.13	2.62
37	I-/-O-glucoside	2.03		1.51	1.60	3.76		1.19		0.73	7.12	4.95	2.72		1.74	2.08	1.44	11.01	16.40		2.98	1.45
	НСС																					
20	1.2-disinanovl gentichioside	6 06	12 57	20 00	2 17	3 63	1 10	12.25	1 10	17 96	21 20	22.00	1/ /0	10.02	6 97	18 00	16 20	1/ 00	18 66	20 71	<u> 0</u> 0 0	12 21
30	1,2-disinapoyl 2 femioploside	32.47	10.07	20.00	17 10	11 00	4.10	21 16	20.35	18.32	97.50 87.57	23.90	14.43	3.00	27 42	51 56	82.68	84 38	80.77	103.34	0.90 55 31	110.35
3 3 40	1.2 diferularyl gentiobioside	11 17	7 45	12 71	5.46	1 01	14.01	21.10	20.33	40.32	37 38	27 70	45.77	5.20	5 / 8	5 35	02.00 17.55	50.26	53.56	55 23	20.10	88 13
40 <u>4</u> 1	1.2.2'-trisinanovl gentiobioside	808	6 28	Q 17	5.40	1.91	4.01	2.00	12.44	10.43	10 12	109 13	34 13	0.00 8 /0	J.40 1 10	9.55 8 07	31 R	5 01	20 50	10 12	5 02	11 50
42	1.2'-disinapoyl-2-femilovi centiobioside	11 87	13 76	<u>41</u> 07	2 1 1	0.84	4 10	4 88	9 63	42 34	98.15	137 95	75 20	0.70 Q 26	16 58	32.26	41 NR	<u>45</u> 70	20.00 95 18	20.74	32 74	38.81
43	1-sinapoyl-2.2'-diferuloyl gentiobioside	9.62	4 20	3.96	3 29	0.07	7.10	5.35	5 33	13.31	40.55	62.38	45.58	1 71	7 21	4 52	19.53	21 44	39 77	12 65	14.38	25.68
-10		0.02	7.20	0.00	0.20			0.00	0.00	10.01	10.00	02.00	10.00			1.02	10.00	<u>~</u> 1.77	00.11	12.00		20.00
Total		1411.94	910.35	1333.07	913.42	624.12	752.80	612.52	873.87	1534.84	2248.74	1935.42	1576.50	1288.45	1730.15	2052.77	1847.04	1600.75	1823.87	1394.39	1423.49	1852.93

The corresponding authors, on behalf of all other co-Authors, regarding the paper entitled "Insights into the phytochemical composition of selected genotypes of organic kale (*Brassica oleracea* L. var. *acephala*), submitted to "Journal of Food Composition and Analysis" declares that the data presented in the manuscript are all authentic and that they are all available, upon a reasonable request.

Sincerely,

Roberto Lo Scalzo (corresponding Author)

Rolly to tel

Author declaration

1. Conflict of Interest

Potential conflict of interest exists:

We wish to draw the attention of the Editor to the following facts, which may be considered as potential conflicts of interest, and to significant financial contributions to this work:

The nature of potential conflict of interest is described below:

X No conflict of interest exists.

We wish to confirm that there are no known conflicts of interest associated with this publication and there has been no significant financial support for this work that could have influenced its outcome.

2. Funding

X Funding was received for this work.

All of the sources of funding for the work described in this publication are acknowledged below:

The present work has been fully funded by the BRESOV Project (Shaping the Future in Organic Breeding and Farming), from the European Union's Horizon 2020 research and innovation programme under grant agreement No 774244.

3. Intellectual Property

X We confirm that we have given due consideration to the protection of intellectual property associated with this work and that there are no impediments to publication, including the timing of publication, with respect to intellectual property. In so doing we confirm that we have followed the regulations of our institutions concerning intellectual property.

4. Research Ethics

X We further confirm that any aspect of the work covered in this manuscript that has involved human patients has been conducted with the ethical approval of all relevant bodies and that such approvals are acknowledged within the manuscript.

X IRB approval was obtained (required for studies and series of 3 or more cases)

X Written consent to publish potentially identifying information, such as details or the case and photographs, was obtained from the patient(s) or their legal guardian(s).

5. Authorship

The International Committee of Medical Journal Editors (ICMJE) recommends that authorship be based on the following four criteria:

- 1. Substantial contributions to the conception or design of the work; or the acquisition, analysis, or interpretation of data for the work; AND
- 2. Drafting the work or revising it critically for important intellectual content; AND
- 3. Final approval of the version to be published; AND
- 4. Agreement to be accountable for all aspects of the work in ensuring that questions related to the accuracy or integrity of any part of the work are appropriately investigated and resolved.

All those designated as authors should meet all four criteria for authorship, and all who meet the four criteria should be identified as authors. For more information on authorship, please see <u>http://www.icmje.org/recommendations/browse/roles-and-responsibilities/defining-the-role-of-authors-and-contributors.html#two</u>.

X All listed authors meet the ICMJE criteria. We attest that all authors contributed significantly to the creation of this manuscript, each having fulfilled criteria as established by the ICMJE.

One or more listed authors do(es) not meet the ICMJE criteria.

We believe these individuals should be listed as authors because:

X We confirm that the manuscript has been read and approved by all named authors.

X We confirm that the order of authors listed in the manuscript has been approved by all named authors.

6. Contact with the Editorial Office

The Corresponding Author declared on the title page of the manuscript is:

[Roberto Lo Scalzo]

X This author submitted this manuscript using his account in EVISE.

X We understand that this Corresponding Author is the sole contact for the Editorial process (including EVISE and direct communications with the office). He is responsible for communicating with the other authors about progress, submissions of revisions and final approval of proofs.

X We confirm that the email address shown below is accessible by the Corresponding Author, is the address to which Corresponding Author's EVISE account is linked, and has been configured to accept email from the editorial office of American Journal of Ophthalmology Case Reports:

[roberto.loscalzo@crea.gov.it]

Someone other than the Corresponding Author declared above submitted this manuscript from his/her account in EVISE:

We the undersigned agree with all of the above.

Author's name (Fist, Last) Signature Date

 1. _Giulia Bianchi_____ Cuilia Brauclus _27/06/2023__

 2. _Valentina Picchi______ (Joluntino Riedus _27/06/2023______)

 3. _Aldo Tava_____ Qeolo Bule____27/06/2023___ 4. Filippo Doria 27/06/23 5. Peter Glen Walley Konesa Dn.E. _____27/06/2023__ 6. Louisa Dever

