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Keywords

electrospinning, titania nanoparticles, polyacrylonitrile, electrospun membranes, fouling

Abstract

Electrospun membranes have very promising properties such as high surface area, high surface area-to-pore volume ratio, high pore interconnectivity, and uniform pore distribution. Nanoparticles are a promising alternative for improving the properties of the electrospun membranes. Titania nanoparticles which are stable, resistant, and non-toxic, have various applications including water treatment, sensors, food additive and cosmetics. Due to the high hydrophilicity of titania nanoparticles, membrane fouling is reduced in titania nanoparticles doped membranes. Titania nanoparticles doped PAN nanocomposite electrospun membranes were prepared by electrospinning method in this work. Compared to bare polyacrylonitrile PAN electrospun membranes 0.05% titania nanoparticles doped electrospun membranes have thinner nanofibers, higher hydrophilicity and almost 2 times lower bovine serum albumin adsorption, which shows lower fouling tendency.

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Titania Nanoparticles Doped Electrospun Membranes

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Abstract

Electrospun membranes have very promising properties such as high surface area, high surface area-to-pore volume ratio, high pore interconnectivity, and uniform pore distribution. Nanoparticles are a promising alternative for improving the properties of the electrospun membranes. Titania nanoparticles which are stable, resistant, and non-toxic, have various applications including water treatment, sensors, food additive and cosmetics. Due to the high hydrophilicity of titania nanoparticles, membrane fouling is reduced in titania nanoparticles doped membranes. Titania nanoparticles doped PAN nanocomposite electrospun membranes were prepared by electrospinning method in this work. Compared to bare polyacrylonitrile PAN electrospun membranes 0.05% titania nanoparticles doped electrospun membranes have thinner nanofibers, higher hydrophilicity and almost 2 times lower bovine serum albumin adsorption, which shows lower fouling tendency.

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1. Introduction

One dimensional nanocomposite fibers are used in sensors, membrane filtration, biomedical applications, energy production, etc. due to their improved electrical, optical and chemical properties [1-7]. There are various methods to produce nanofibers including drawing, templates, phase separation, self-assembly, and electrospinning [8]. Among them, electrospinning is the most practical, economical, and fast [9, 10]. Electrospinning has attracted attention since its discovery in 1934. Nanofibers prepared by electrospinning has many remarkable features such as small diameter (50 nm - 10 μ m), high aspect ratio (length-to-diameter ratio), large specific surface area (ratio of surface area to volume), diversity in composition, unique physicochemical properties, flexibility in chemical / physical surface functionalization [5, 11, 12]. The properties of electro spun nanofibers can be improved by adding nanomaterials. Nanomaterials doped composite electro spun nanofibers in many fields including filtration, sensors, fuel cell and nanoelectronics.

Electrospun membranes have very promising properties such as high surface area, a high surface area-to-pore volume ratio, high pore interconnectivity, and uniform pore distribution. There are many recent researches in functionalizing electrospun nanofibers to improve their applicability in different areas. Nanoparticles are a promising alternative for improving the properties of the electrospun nanofibers. Properties like uniform pore size, narrow pore size distribution, hydrophilicity, mechanical strength, and stability can be enhanced by nanoparticle addition in the nanofiber structure [13].

Titania nanoparticles are stable, resistant, and non-toxic. Hence, they have various applications, including water treatment, sensors, food additive and cosmetics [14-16]. Due to the high hydrophilicity of titania nanoparticles, membrane fouling is reduced in titania nanoparticles doped membranes [17]. The objective of this work is to synthesize titania nanoparticles doped

47 polyacrylonitrile (PAN) nanocomposite electrospun membranes by electrospinning with lower
48 fouling tendency. Fourier transform infrared (FTIR) spectroscopy and scanning electron
49 microscopy (SEM) were employed for characterizing the nanocomposite electrospun
50 membranes. Moreover, bovine serum albumin adsorption of nanocomposite electrospun
51 membranes was also determined.

52 2. Materials and Methods

53 Titania nanoparticles (with a diameter of less than 25 nm) and anhydrous N, N-dimethyl
54 formamide (DMF, 99.8%) were purchased from Sigma (USA), and PAN (with a molecular
55 weight of 150,000) from Polysciences Inc. (USA).

56 The dope solution for the nanofiber synthesis was prepared as follows: Initially titania
57 nanoparticles was ultrasonicated in DMF, then PAN was dissolved. Afterwards, the solution
58 was ultrasonicated to remove air bubbles. All nanofibers were synthesized by using the
59 electrospinning instrument (NE100, Inovenco Co. Ltd., Turkey) represented in Fig. 1.

60 <Fig. 1.>

61 Electric field was generated between the nozzle system and a rotating collection drum. There
62 were 4 nozzles with an inner diameter of 700 μm each. The rotation speed of the collection
63 drum was 350 rpm. A syringe pump was used for pumping the dope solution. Nanofibers were
64 collected on a non-woven fabric. Initially voltage applied, distance between the nozzle system
65 and collection drum, and feed rate of the dope solutions were optimized for all solutions. Then
66 these values (TABLE 1) were used for nanocomposite electrospun membrane synthesis.

67 <TABLE 1.>

69 A viscosimeter (Vibro, And, Japan) was used to measure the viscosities of the polymer
70 solutions. SEM (Quanta FEG 250; FEI, USA) and FTIR (Spectrum Two, Perkin Elmer) were
71 used for determining the surface morphologies and the structure of the nanocomposite
72 electrospun membranes. Average fiber diameters were reported by averaging the diameters of
73 the 20 nanofibers from each of 20 different SEM images for each membrane.

74 The structures of the electrospun membranes were examined by FTIR (Spectrum Two,
75 PerkinElmer, USA), and their surface hydrophilicity was determined using a contact angle
76 goniometer (Theta Lite, Attension, Sweden). It is calculated by averaging at least seven contact
77 angle values measured for each membrane.

78 For the adsorption tests, membranes were cut into small pieces and immersed into 1 g/L BSA
79 solution at neutral pH at room temperature for 4 h. Afterwards, the coupons were ultrasonicated
80 in DI water for 2 min. The amount of the adsorbed BSA was directly measured using a UV-vis
81 spectrometer (UV-VIS Spectrophotometer Shimadzu, Japan) at 280 nm. The average of at least
82 two measurements was reported.

83 3. Results and Discussion

84 Viscosities of the polymer solutions of the electrospun membranes are given in TABLE 2. As
85 shown in the TABLE titania nanoparticles addition increased the viscosity of the polymer
86 solution. The increase in the viscosity in the polymer solution improved the dispersion of the
87 nanoparticles in the polymer solution.

88 <TABLE 2.>

89 SEM images of the electrospun membranes are given in Fig. 2. SEM images show that no
90

91 beads were formed in any of the electrospun membranes. The balance between the electrostatic
92 repulsion, surface tension, and viscoelastic forces determines the quality of the fibers in the
93 electrospinning process. Increase in viscosity of the polymer solution results in suppression of
94 the surface tension due to the electrostatic repulsion and viscoelastic forces. Bead free and
95 smooth fibers were formed by the dominating viscoelastic forces [9].

96 <Fig. 2.>

97 Reported nanofiber diameters of the electrospun membranes are the average of the 20
98 nanofibers from 20 different SEM images for each membrane. The average of the nanofiber
99 diameters determined is shown in Fig.3. Addition of 0.05% titania nanoparticles in the polymer
100 solution reduced the nanofiber diameter. However, further increase in the titania nanoparticles
101 amount resulted in increased nanofiber diameter.

102 <Fig. 3.>

103 FTIR spectra recorded in the spectral range of 4000 - 400 cm^{-1} of the electrospun membranes
104 are given in Fig.4. The peak around 2250 cm^{-1} corresponds to $\text{C}\equiv\text{N}$ stretching vibration [18],
105 around 1260 cm^{-1} corresponds to weak ether peak ($\text{C}-\text{O}-\text{C}$) [19] of PAN. Compared to the
106 bare PAN electrospun membrane (P/T-0) two new peaks were observed in titania nanoparticles
107 doped PAN electrospun membranes (P/T-0.05, P/T-0.1, P/T-0.2) around 874 cm^{-1} and 972 cm^{-1}
108 ¹ corresponding to $\text{Ti}-\text{O}-\text{Ti}$ vibrations [18, 20], which confirmed the presence of titania
109 nanoparticles into the nanofiber matrix.

110 <Fig. 4.>

111 Hydrophilicity of the membranes was evaluated by determining contact angles (Fig.5). Lower
112 contact angle shows higher hydrophilicity and higher contact angle shows higher

114 hydrophobicity. Addition of titania nanoparticles increased the hydrophilicity of the
115 electrospun membranes. Moreover, increasing the titania nanoparticles amount in the fiber
116 composition increased the hydrophilicity of the electrospun membranes.

117 <Fig. 5.>

118 BSA was chosen as a model protein for determining the adsorption resistances of the
119 electrospun membranes. BSA adsorption on electrospun membranes is shown in Fig.6.
120 Addition of titania nanoparticles in the structure of the nanofibers of the electrospun
121 membranes reduced the BSA adsorption almost 2 times compared to the bare membranes at
122 neutral pH. This indicates the potential usage of titania nanoparticles for alleviating the fouling.

123 <Fig. 6.>

124 4. Conclusions

125 The preparation and fouling behavior of titania nanoparticles doped PAN electrospun
126 membranes were investigated, with several conclusions subsequently drawn. These
127 conclusions include the following.

- 128 • Titania nanoparticles doped electrospun membranes were successfully synthesized by the
129 electrospinning method. New peak formations in FTIR analysis confirmed that titania
130 nanoparticles successfully blended in the nanofiber structure.
- 131 • Nanofibers of the titania nanoparticles doped PAN electrospun membranes are thinner
132 than the bare membranes.

134 • Titania nanoparticles doped PAN electrospun membranes are more hydrophilic than the
135 bare electrospun PAN membranes.

136 • BSA adsorption on titania nanoparticles doped PAN electrospun membranes are almost
137 2 times lower than the bare membranes, indicating potential usage of titania nanoparticles
138 for alleviating the fouling.

139 • Titania nanoparticles doped PAN electrospun membranes can be an alternative to
140 disposable membranes like syringe filters with their reduced fouling tendency.

141 **Acknowledgments**

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200 Figures

201 Fig. 1. Schematic representation of electrospinning instrument

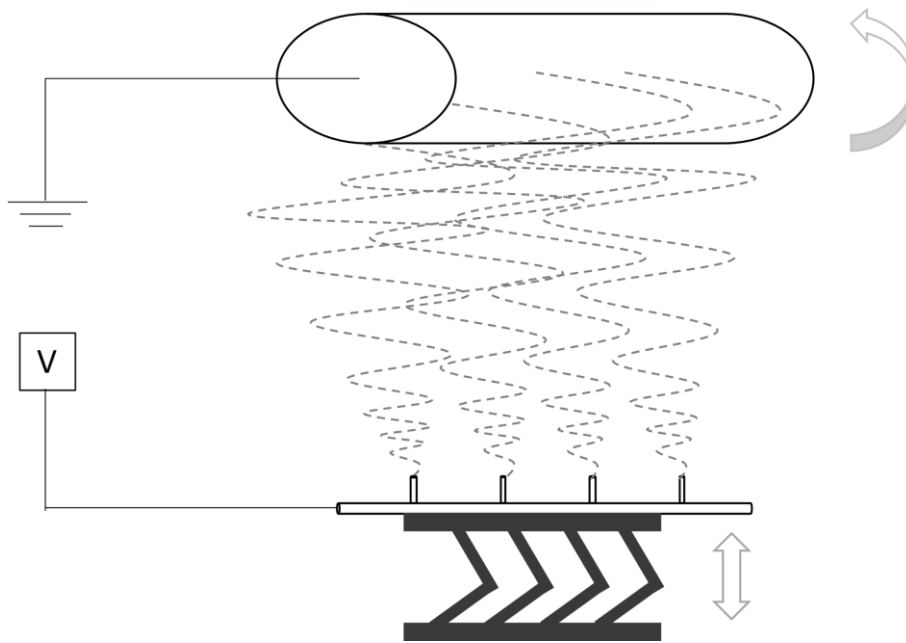
202 Fig. 2. SEM images of electrospun membranes at different magnifications; (a) P/T-0 at 1000x
203 magnification, (b) P/T-0 at 10000x magnification, (c) P/T-0.05 at 1000x magnification, (d) P/T-
204 0.05 at 10000x magnification, (e) P/T-0.1 at 1000x magnification, (f) P/T-0.1 at 10000x
205 magnification, (g) P/T-0.2 at 1000x magnification, (h) P/T-0.2 at 10000x magnification

206 Fig. 3. Fiber thicknesses of the electrospun membranes

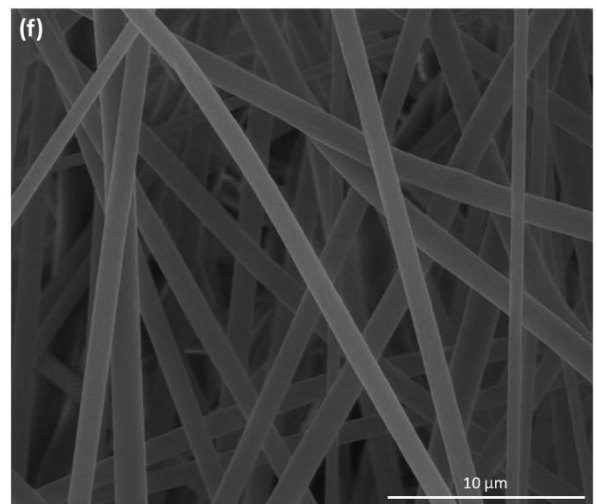
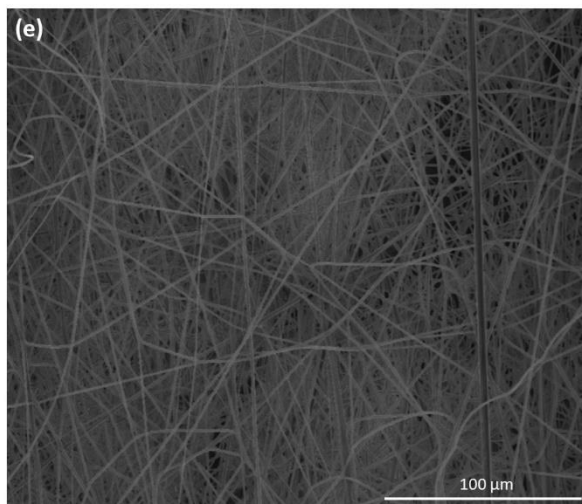
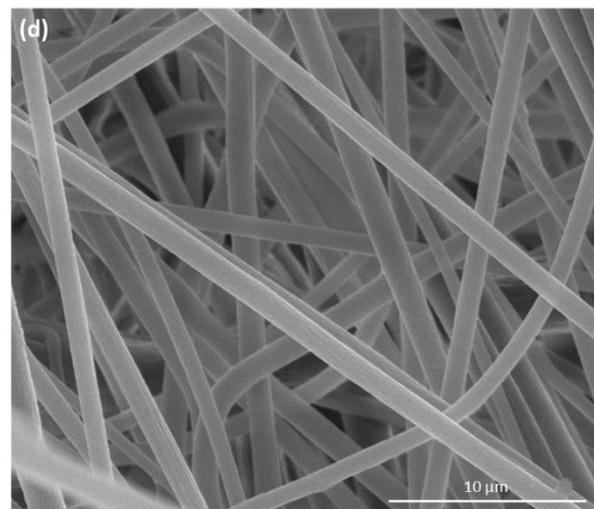
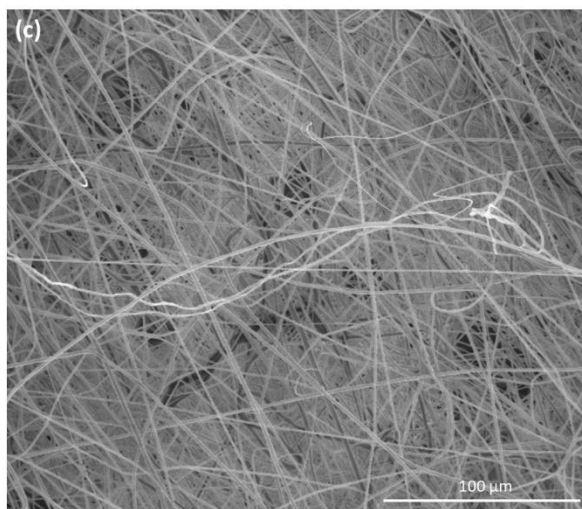
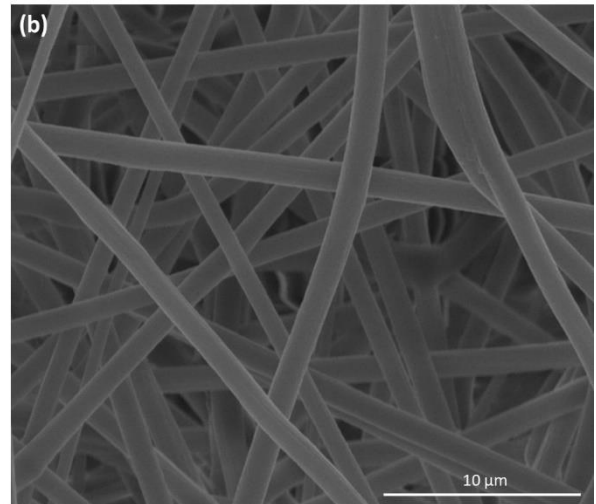
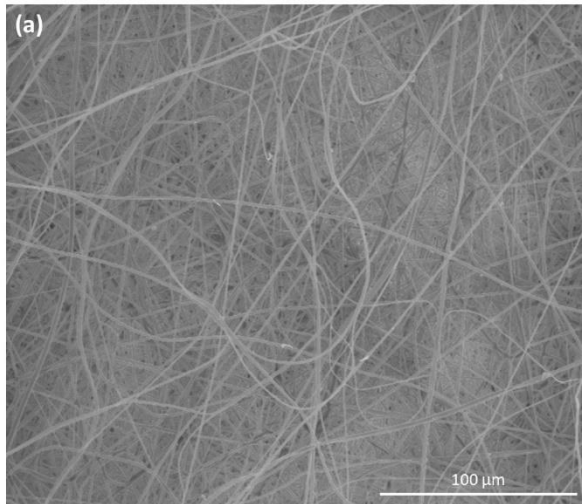
207 Fig. 4. FTIR spectra of electrospun membranes

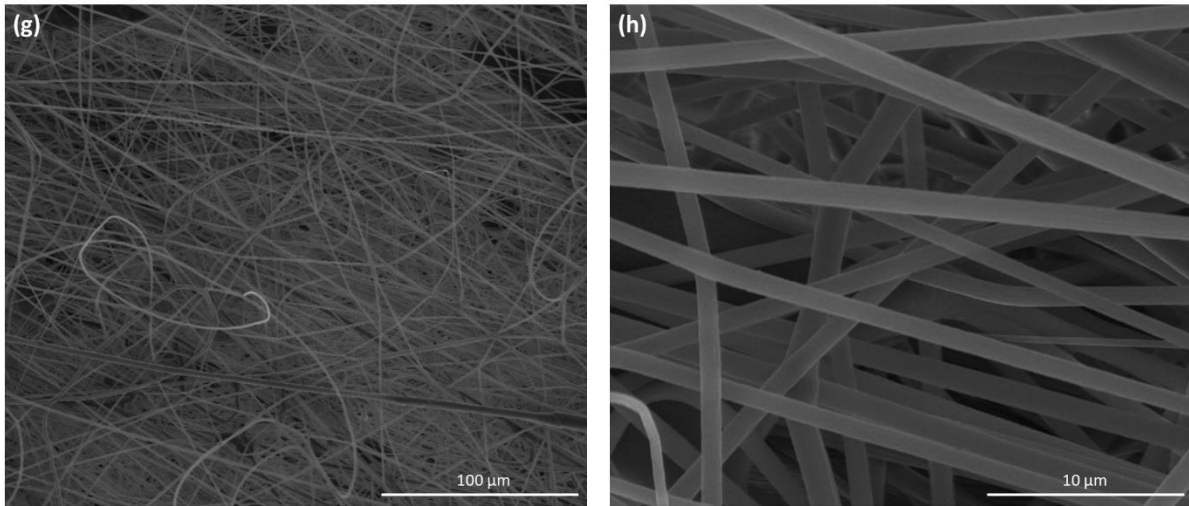
208 Fig. 5. Contact angles of the electrospun membranes

209 Fig. 6. BSA adsorption on electrospun membranes



211 Fig. 1 Schematic representation of electrospinning instrument





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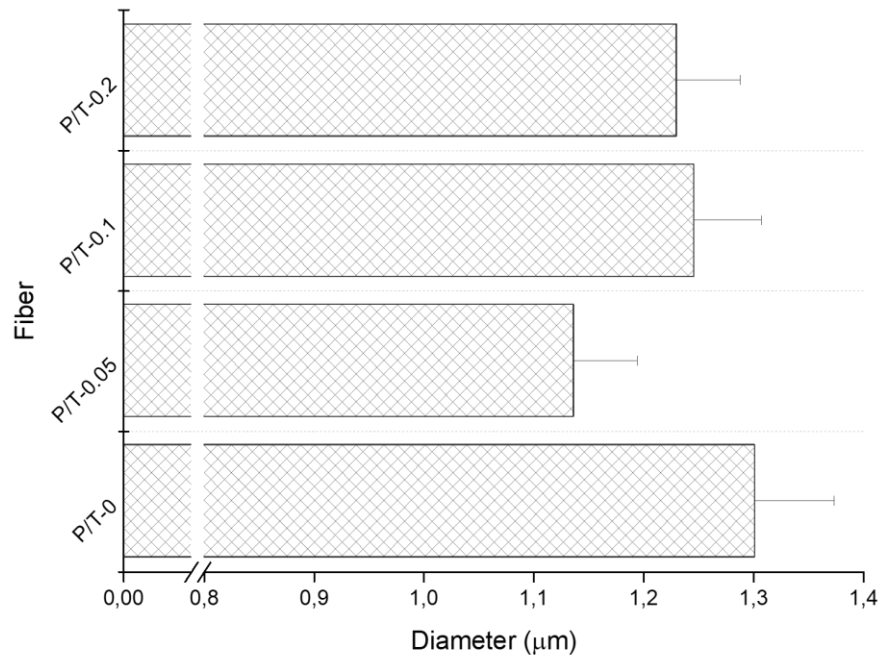


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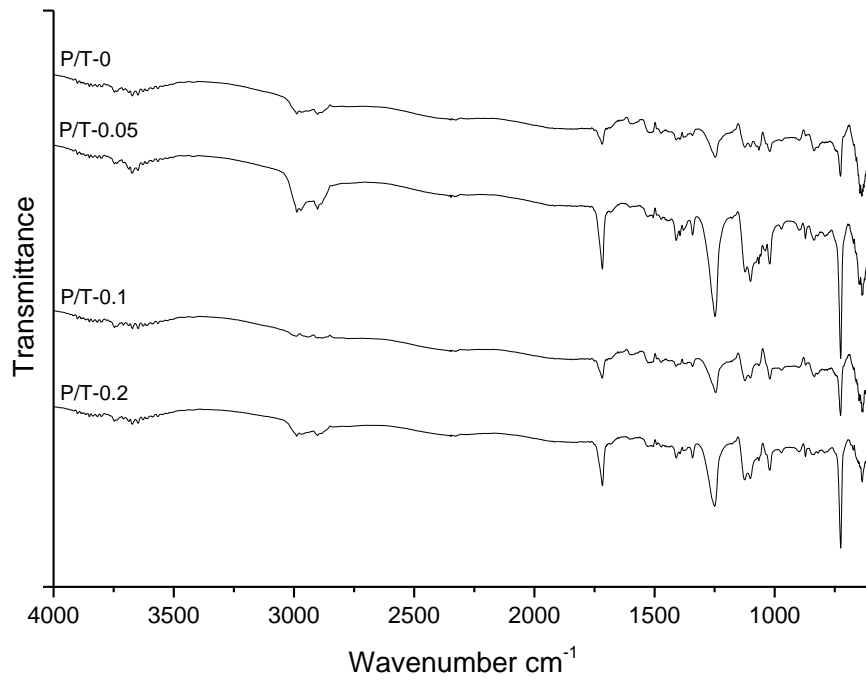


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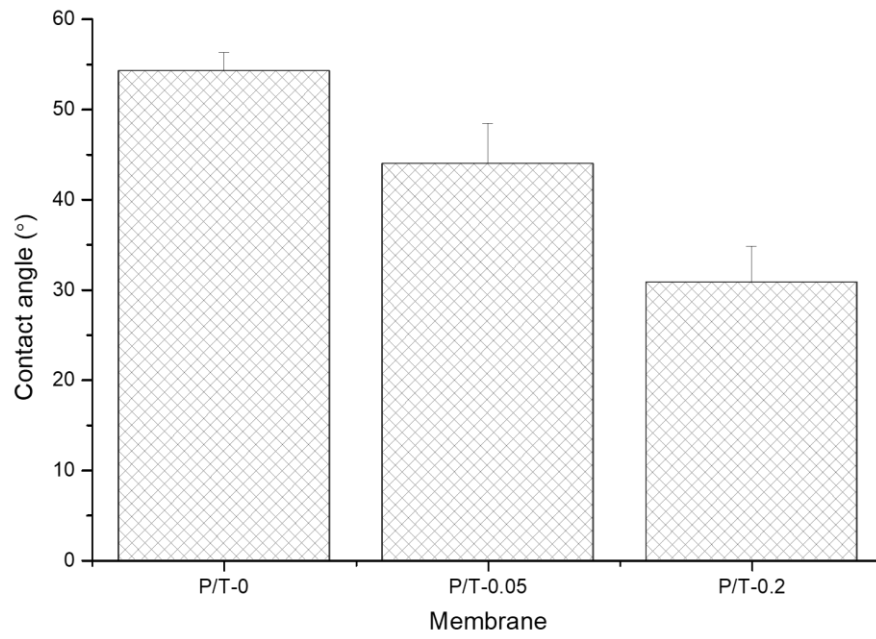


Fig. 5. Contact angles of the electrospun membranes

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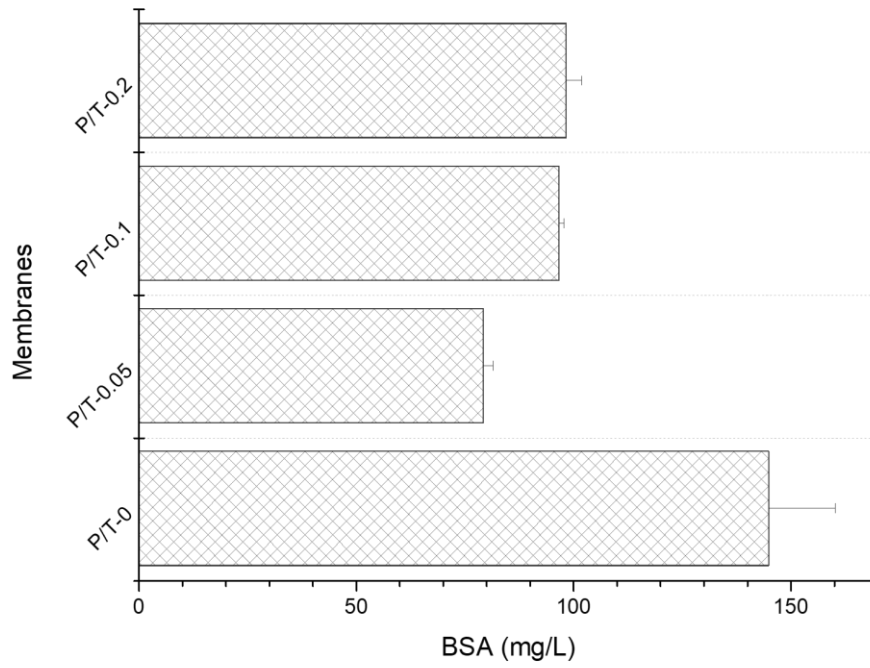


Fig. 6. BSA adsorption on electrospun membranes

226

227

228 Tables

229 TABLE 1. Electrospinning characteristics

230 TABLE 2. Viscosities of the polymer solutions

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233	Membrane	nTiO₂ Ratio	Voltage (kV)	Feed Rate	Distance (cm)
234	Name	(%)		(mL/h)	
235	P/T-0	0	35	6	16
236	P/T-0.05	0.05	34	12	16
237	P/T-0.10	0.10	34	12	16
238	P/T-0.20	0.20	34	10	16

240 TABLE 2. Viscosities of the polymer solutions

Membrane Name	Viscosity (Pa.s)	Temperature (°C)
P/T-0	3.25	21.0
P/T-0.05	4.09	20.3
P/T-0.10	3.70	20.1
P/T-0.20	3.65	20.2

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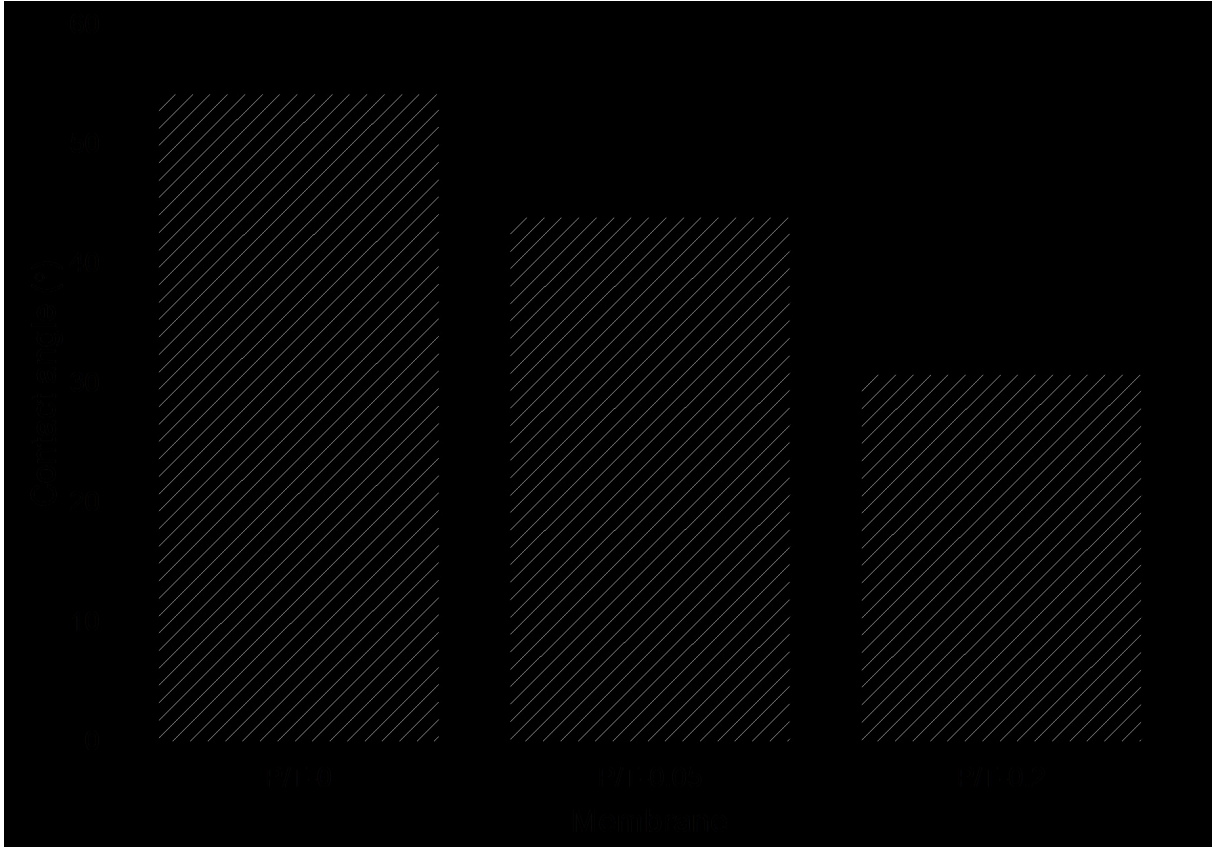


Fig. 5. Contact angles of the electrospun membranes



Fig. 4. FTIR spectra of electrospun membranes



Fig. 3. Fiber thicknesses of the electrospun membranes

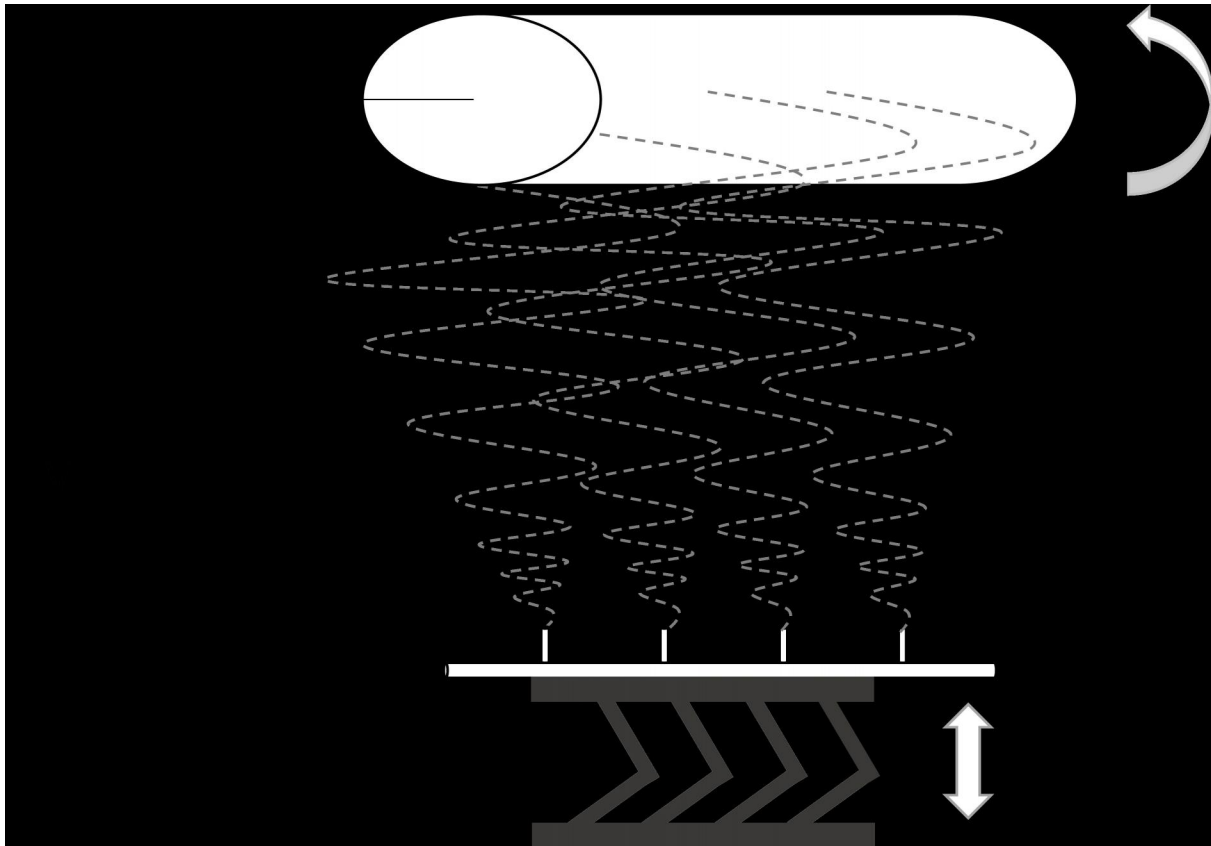


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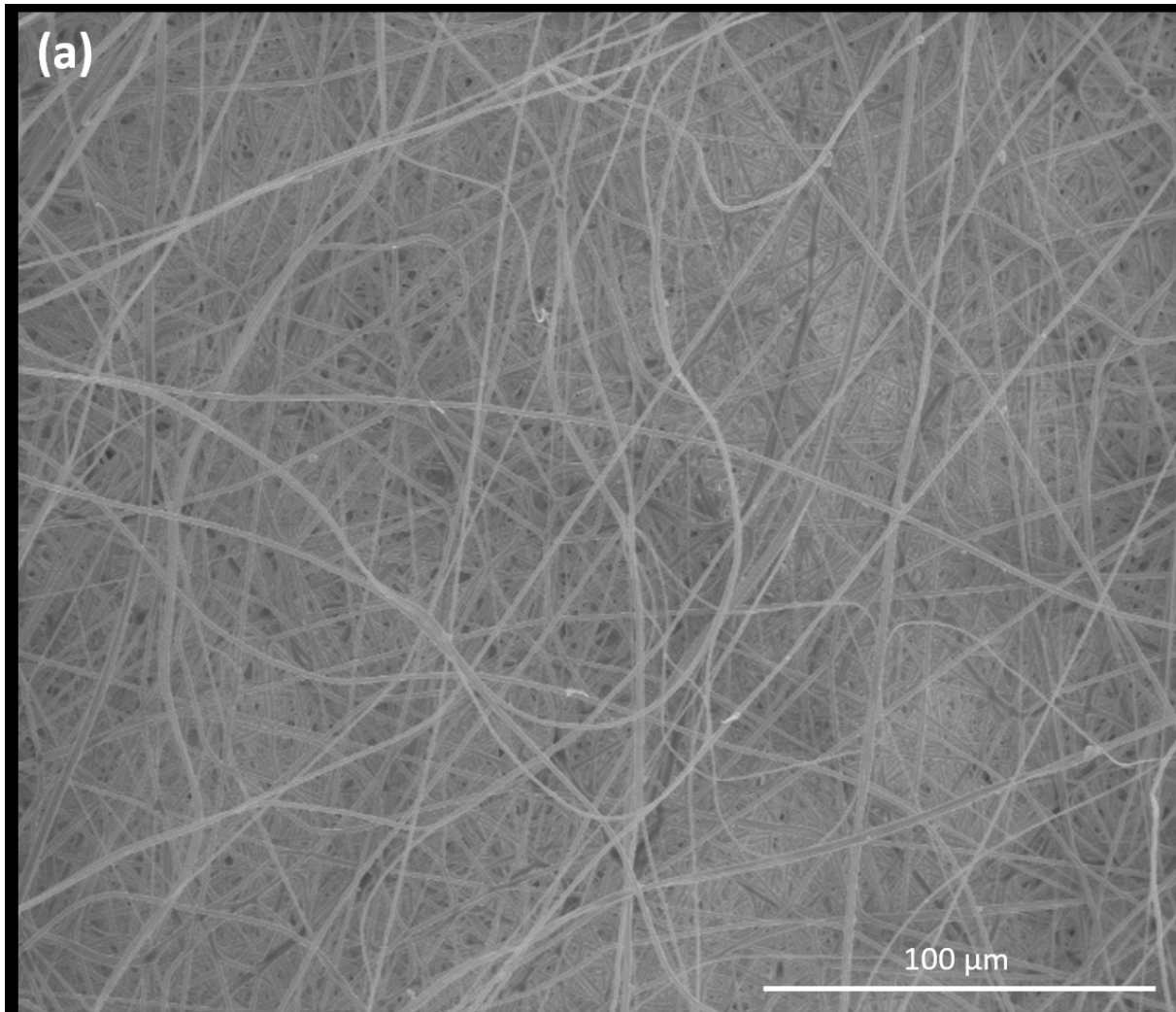


Fig. 2a. P/T-0 at 1000x magnification

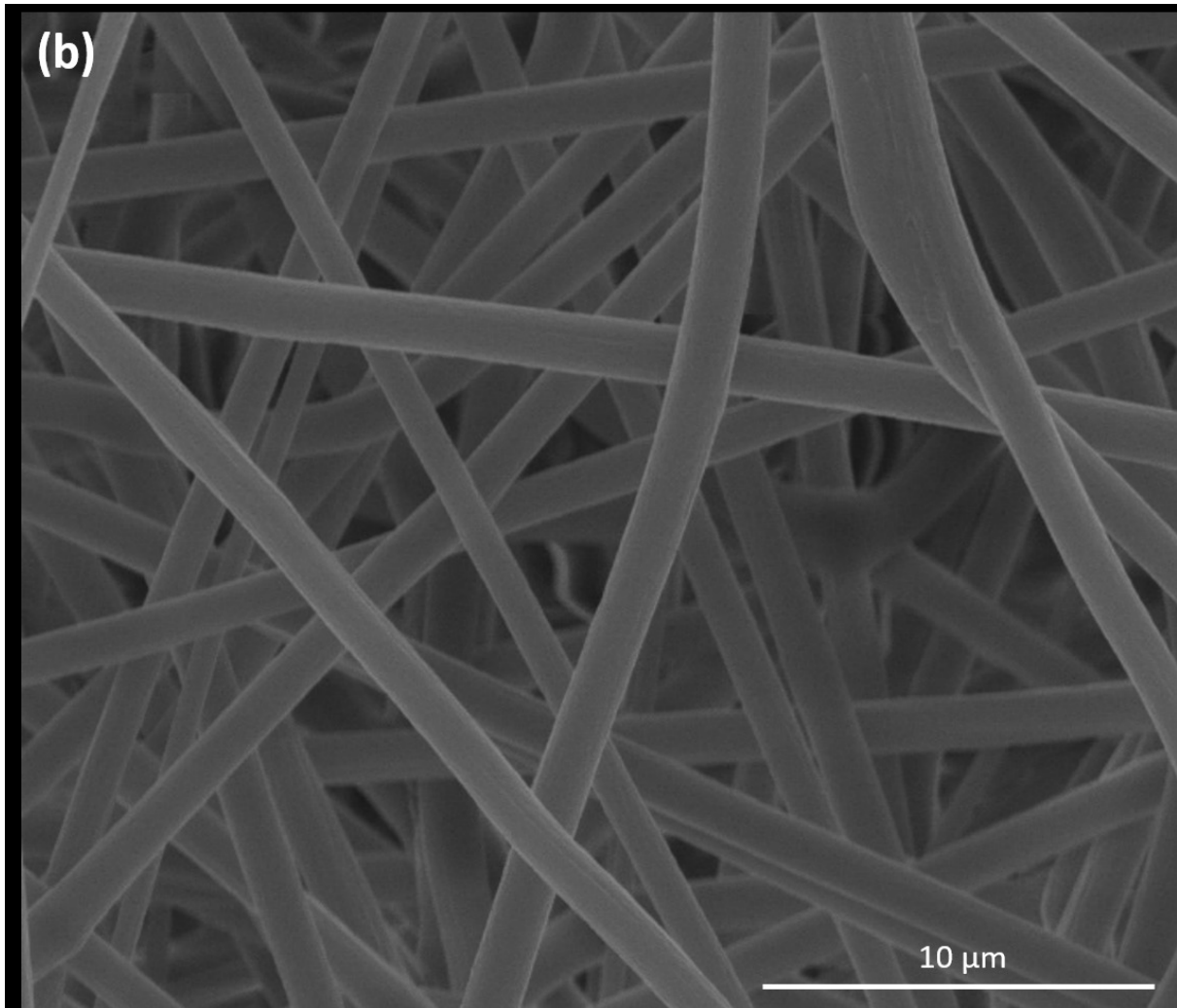


Fig. 2b. P/T-0 at 10000x magnification

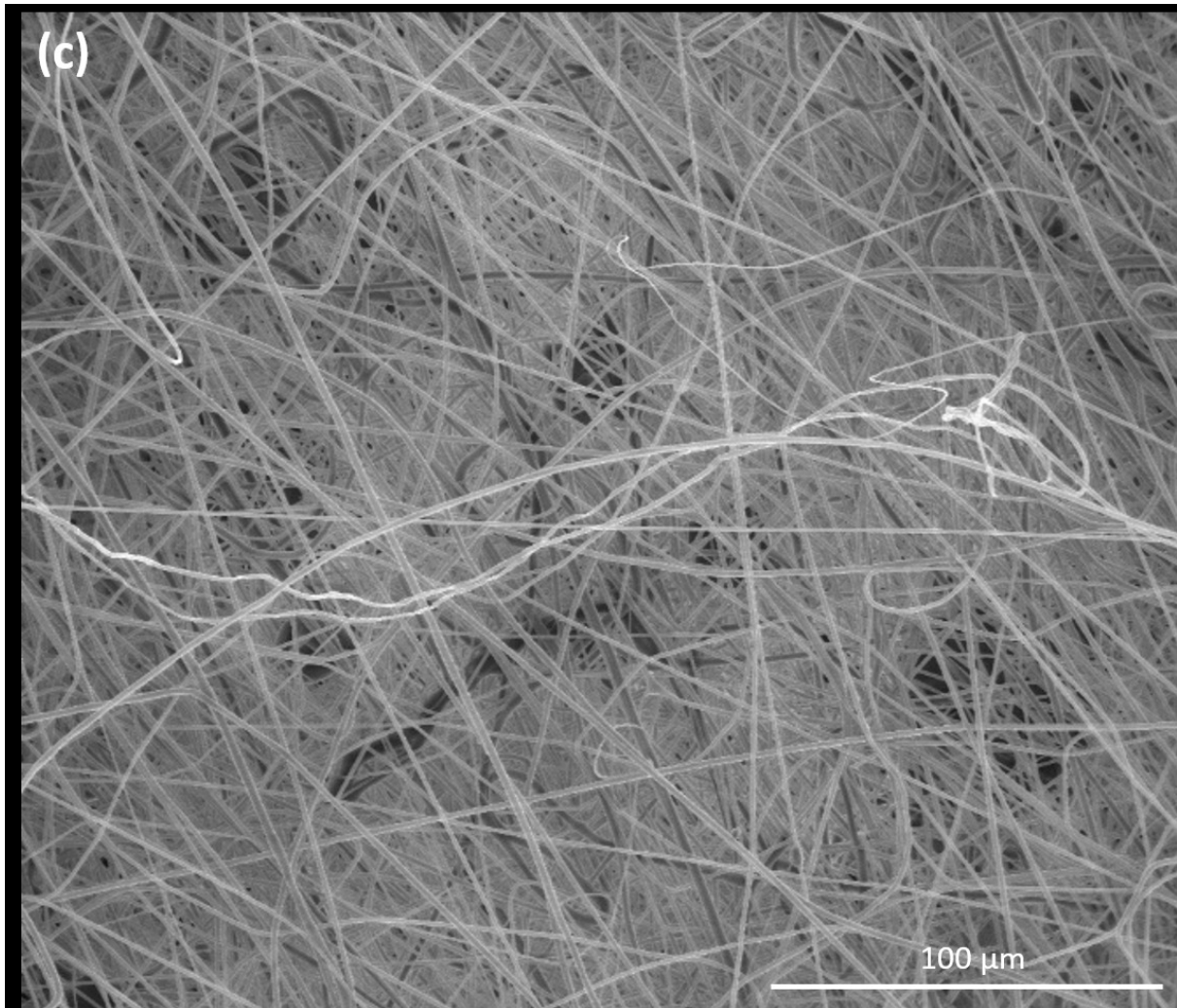


Fig. 2c. P/T-0.05 at 1000x magnification

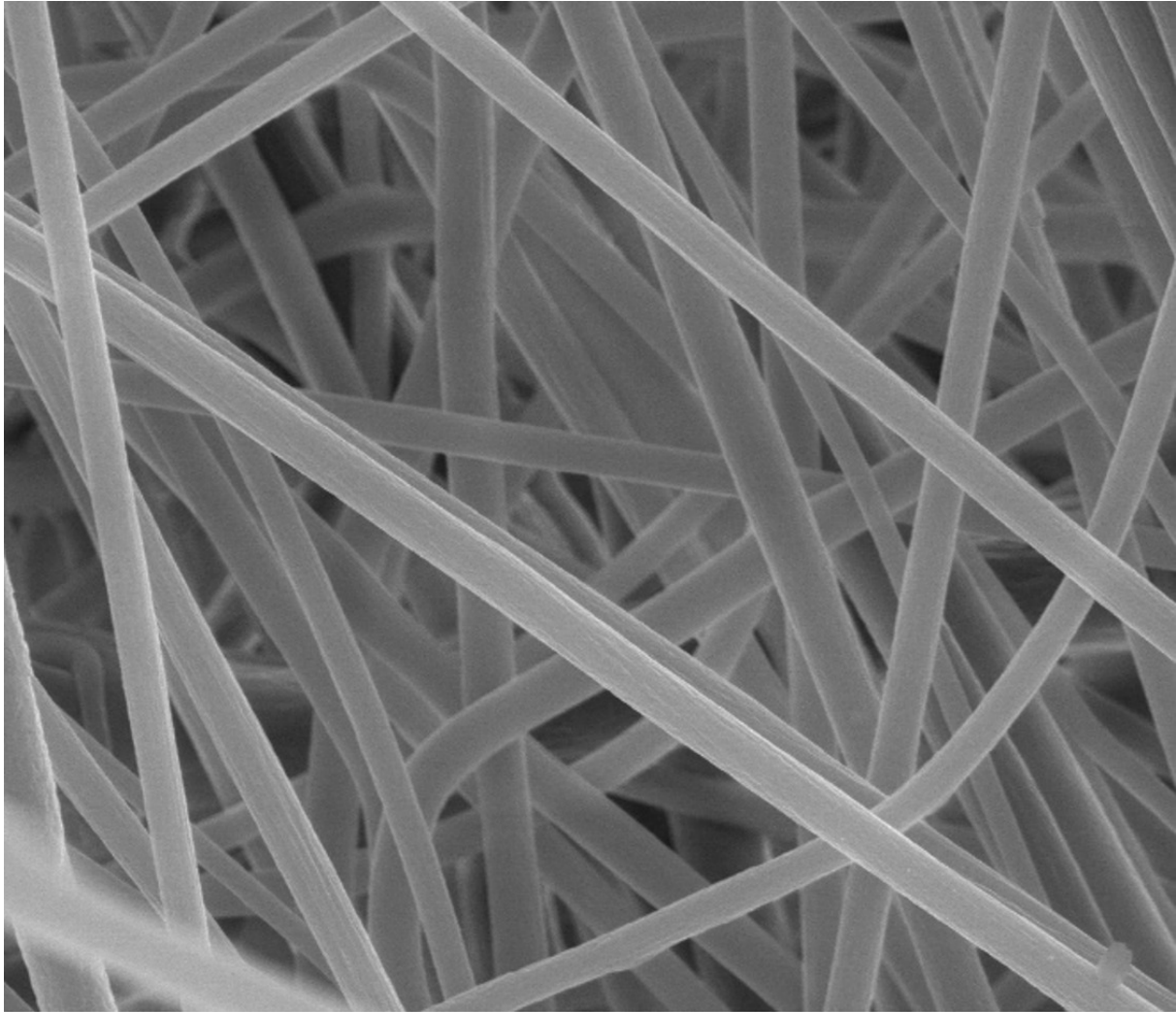


Fig. 2d. P/T-0.05 at 10000x magnification

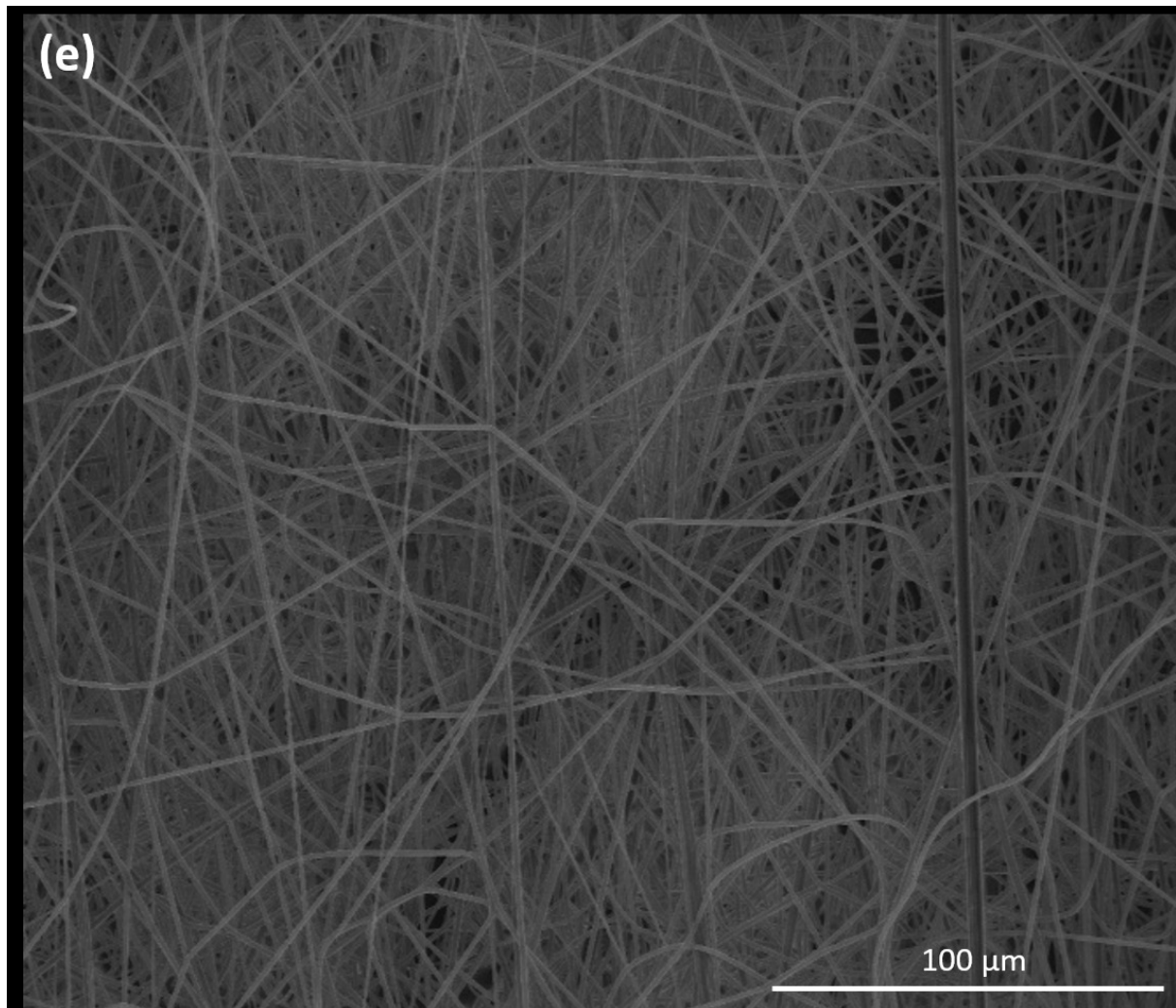


Fig. 2e. P/T-0.1 at 1000x magnification

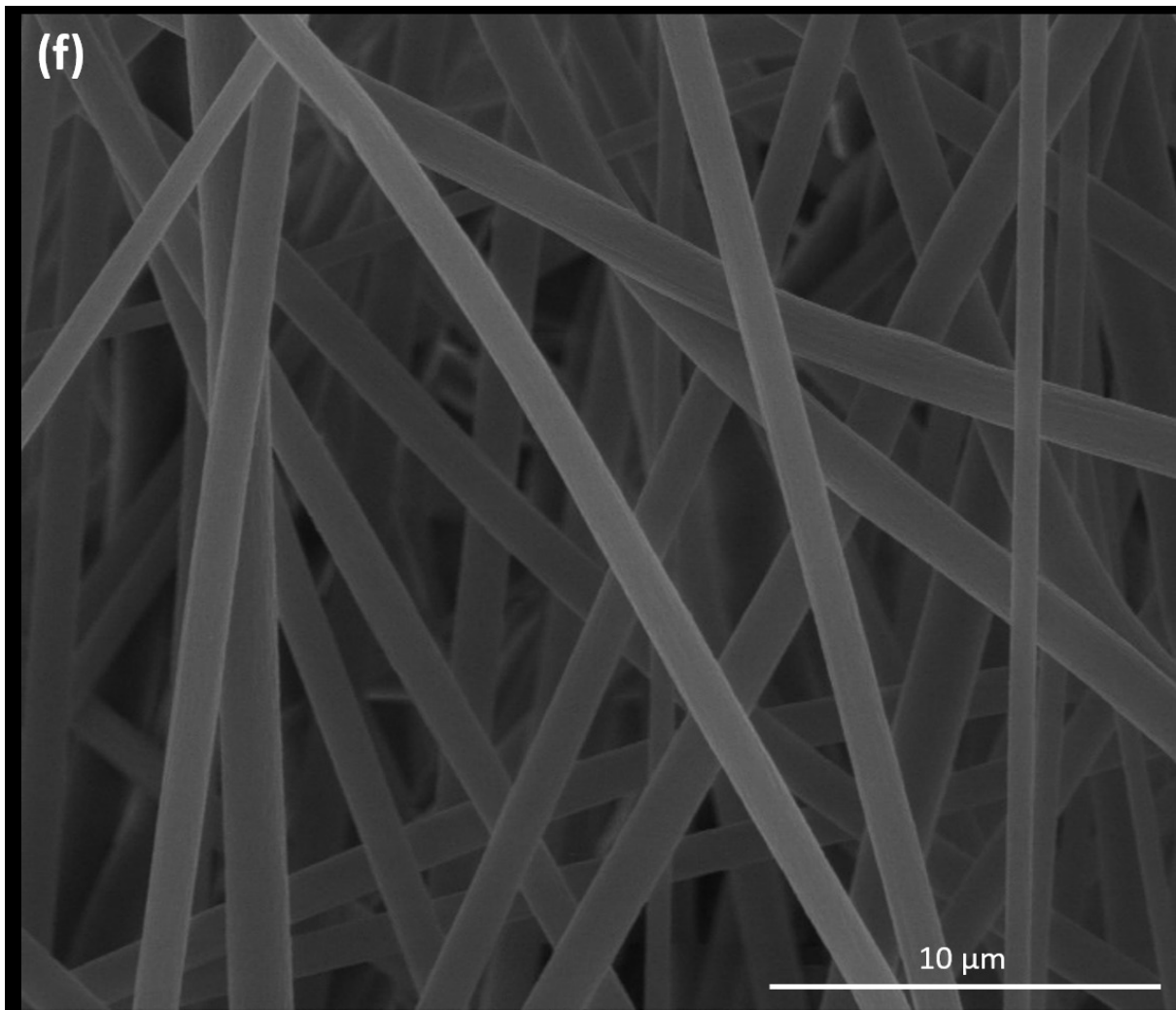


Fig. 2f. P/T-0.1 at 10000x magnification

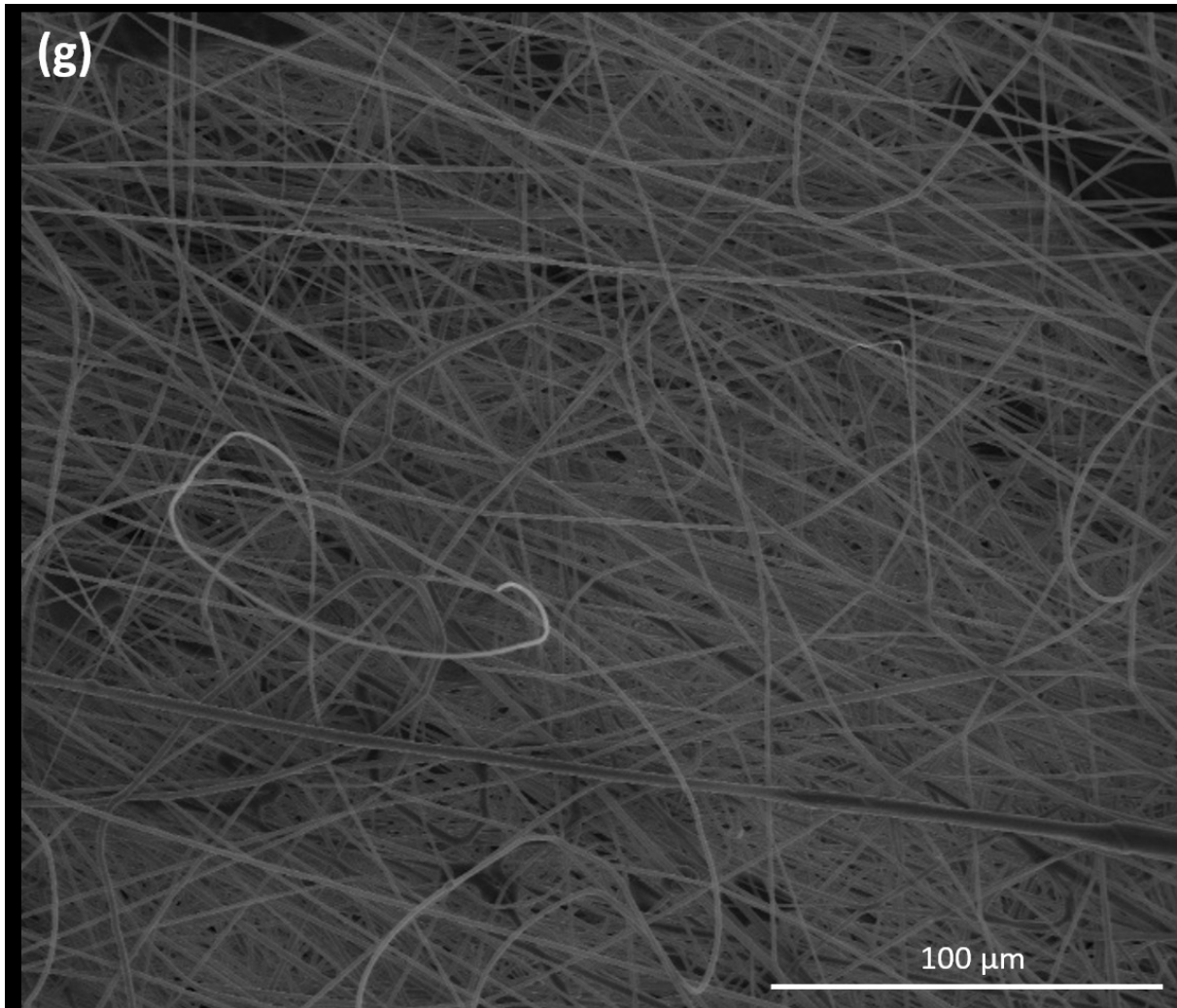


Fig. 2g. P/T-0.2 at 1000x magnification

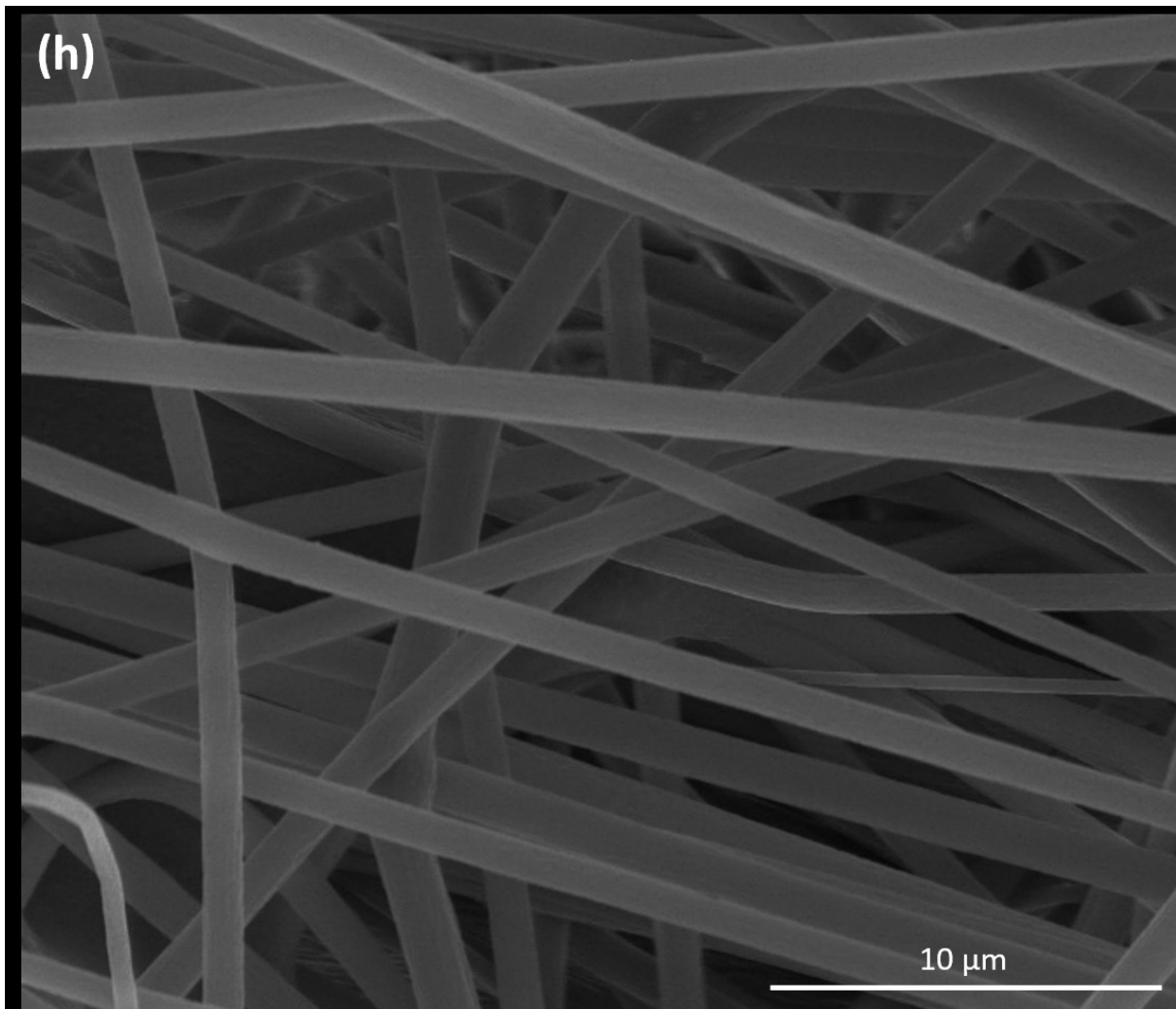


Fig. 2h. P/T-0.2 at 10000x magnification

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Tables

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TABLE 1. Electrospinning characteristics

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TABLE 2. Viscosities of the polymer solutions

Figures

Figure 1 - [Download source file \(105.38 kB\)](#)

Fig. 6. BSA adsorption on electrospun membranes

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Fig. 5. Contact angles of the electrospun membranes

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Fig. 4. FTIR spectra of electrospun membranes

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Fig. 3. Fiber thicknesses of the electrospun membranes

Figure 5 - [Download source file \(263.77 kB\)](#)

Fig. 1 Schematic representation of electrospinning instrument

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Fig. 2a. P/T-0 at 1000x magnification

Figure 7 - [Download source file \(924 kB\)](#)

Fig. 2b. P/T-0 at 10000x magnification

Figure 8 - [Download source file \(1.49 MB\)](#)

Fig. 2c. P/T-0.05 at 1000x magnification

Figure 9 - [Download source file \(1.16 MB\)](#)

Fig. 2d. P/T-0.05 at 10000x magnification

Figure 10 - [Download source file \(1.29 MB\)](#)

Fig. 2e. P/T-0.1 at 1000x magnification

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Fig. 2f. P/T-0.1 at 10000x magnification

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Fig. 2g. P/T-0.2 at 1000x magnification

Figure 13 - [Download source file \(797.51 kB\)](#)

Fig. 2h. P/T-0.2 at 10000x magnification